XE (G): Q. 1 – Q. 9 carry one mark each & Q. 10 – Q. 22 carry two marks each.

Q.1 Which of the following is oil soluble pigment present in fruits and vegetables?

(A) Flavonoids	(B) Carotenoids	(C) Anthocyanins	(D) Tannins
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Q.2 Which of the following represent the group of saturated fatty acids?

(A) Lauric, Myristic, Arachidic	(B) Palmitic, Linoleic, Linolenic
(C) Capric, Stearic & Oleic	(D) Behenic, Caprylic, Arachidonic

Q.3 The anti-nutritional factor present in fava bean is

(A) Gossypol	(B) Curcine
(C) Vicine	(D) Cyanogen

Q.4 Irradiation carried out to reduce viable non-spore forming pathogenic bacteria using a dose between 3 to 10 kGy is called

(A) Radurization	(B) Thermoradiation
(C) Radappertization	(D) Radicidation

Q.5 Identify the correct statement related to the viscosity of Newtonian fluids from the following

- (A) It is not influenced by temperature
- (B) It increases with shearing rate
- (C) It decreases with shearing rate
- (D) It is not influenced by shearing rate
- Q.6 Adult male Wistar rats were fed with a protein based diet. Total 150 g of protein was ingested per animal. If the average weight increased from 110 g to 350 g after the end of the experiment, the Protein efficiency ratio of the given protein would be _____. (up to two decimal points)
- Q.7 The initial moisture content of a food on wet basis is 50.76%. Its moisture content (%) on dry basis is _____.(up to two decimal points)

- Q.8 The oxygen transmission rate through a 2.54 x 10^{-3} cm thick low density polyethylene film with air on one side and inert gas on the other side is 3.5 x 10^{-6} mL cm⁻² s⁻¹. Oxygen partial pressure difference across the film is 0.21 atm. The permeability coefficient of the film to oxygen is _____ x 10^{-11} mL (STP) cm cm⁻² s⁻¹ (cm Hg)⁻¹.
- Q.9 Ambient air at 30°C dry bulb temperature and 80% relative humidity was heated to a dry bulb temperature of 80°C in a heat exchanger by indirect heating. The amount of moisture gain (g kg⁻¹ dry air) during the process would be _____.
- Q.10 Match the commodity in Group I with the bioactive constituent in Group II

Group I	Group II
P. Ginger	1. Lutein
Q. Green tea	2. Gingerol
R. Spinach	3. Curcumin
S. Turmeric	4. Epigallocatechin gallate

- (A) P-1, Q-2, R-3, S-4 (B) P-2, Q-4, R-1, S-3 (C) P-4, Q-1, R-3, S-2 (D) P-2, Q-3, R-1, S-4
- Q.11 Match the process operation in Group I with the separated constituent in Group II

Group I		Group II
P. Extraction		1. Phospholipids
Q. Degumming		2. Free fatty acids
R. Neutralization		3. Pigments
S. Bleaching		4. Crude oil
(A) P-3, Q-2, R-4, S-1	(B) P-4, Q-3, R-1, S-2	
(C) P-4, Q-1, R-2, S-3	(D) P-4, Q-1, R-3, S-2	

Q.12 Match the spoilage symptom in Group I with the causative microorganism in Group II

Group I	Group II
P. Green rot of eggs	1. <i>Micrococcus</i> spp.
Q. Putrid swell in canned fish	2. Serratia marcescens
R. Red bread	3. Pseudomonas fluorescens
S. Yellow discoloration of meat	4. Clostridium sporogenes
(A) P-4, Q-3, R-2, S-1	(B) P-2, Q-1, R-4, S-3

(D) P-1, Q-4, R-3, S-2

(C) P-3, Q-4, R-2, S-1

Q.13 Match the fermented product in Group I with the base material in Group II

Group I	Group II
P. Sake	1. Milk
Q. Chhurpi	2. Cabbage
R. Natto	3. Rice
S. Sauerkraut	4. Soybean
(A) P-3, Q-1, R-4, S-2 (C) P-4, Q-1, R-3, S-2	(B) P-1, Q-3, R-4, S-2 (D) P-3, Q-2, R-1, S-4

Q.14 Match the operation in Group I with the process in Group II

Group I	Group II
P. Cleaning	1. Quality separation
Q. Grading	2. Clarification
R. Size reduction	3. Screening
S. Filtration	4. Comminution
(A) P-1, Q-3, R-4, S-2	(B) P-4, Q-1, R-3, S-2
(C) P-2, Q-4, R-1, S-3	(D) P-3, Q-1, R-4, S-2

- Q.15 Out of 7 principles of HACCP system, 4 are listed below. Arrange these principles in the order in which they are applied.
 - (P) Conduct a hazard analysis
 - (Q) Establish monitoring process
 - (R) Establish critical limit
 - (S) Establish record keeping and documentation process

 $(A) P, R, Q, S \qquad (B) Q, R, P, S \qquad (C) P, Q, R, S \qquad (D) R, S, P, Q$

- Q.16 Apple juice of 10% total solids (TS) is being concentrated in a single effect evaporator working with a surface condenser to 40% TS under a vacuum of 20 kPa. After some time the vacuum pump stops but the evaporation process continued. Choose the combination of possible implications from the following.
 - (P) Product quality is affected
 - (Q) Substantial increase in thermal energy requirement
 - (R) Decrease in the rate of evaporation

 $(A) P \& Q \qquad (B) Q \& R \qquad (C) R \& P \qquad (D) P, Q \& R$

- Q.17 Identify an example of a classical diffusional mass transfer process without involving heat, among the following.
 - (A) Drying of food grains
 - (B) Carbonation of beverages
 - (C) Distillation of alcohol
 - (D) Concentration of fruit juice
- Q.18 For an enzyme catalyzed reaction $S \rightarrow P$, the kinetic parameters are: $[S] = 40 \ \mu M$, $V_0 = 9.6 \ \mu M \ s^{-1}$ and $V_{max} = 12.0 \ \mu M \ s^{-1}$. The K_m of the enzyme in μM will be _____.(up to one decimal points)
- Q.19 A microbial sample taken at 10 AM contained 1×10^5 CFU/mL. The count reached to 1×10^{10} CFU/mL at 8 PM of the same day. The growth rate (h⁻¹) of the microorganism would be _____.(up to two decimal points)
- Q.20 Black pepper is ground from an equivalent particle size of 6 mm to 0.12 mm using a 10 hp motor. Assuming Rittinger's equation and that 1 hp = 745.7 W, the power (hp) of motor required to fine grind black pepper to 0.08 mm would be _____.(up to two decimal points)
- Q.21 Green pea (average diameter 0.8 cm) is frozen in a blast freezer operating at -40°C and with a surface heat transfer coefficient of 30 W m⁻² K⁻¹. The thermal conductivity of pea is 2.5 W m⁻¹K⁻¹, and latent heat of crystallization is 2.74 x 10² kJ kg⁻¹. If the freezing point of pea is -1°C and the density is 1160 kg m⁻³, the freezing time in minutes will be _____.(up to two decimal points)
- Q.22 The rate of heat transfer from a metal plate is 1000 W m⁻². The surface temperature of the plate is 120°C and ambient temperature is 20°C. The convective heat transfer coefficient (W m⁻² °C⁻¹) using the Newton's law of cooling will be _____.

END OF THE QUESTION PAPER