Signature of Invigilator

# COMBINED ENTRANCE EXAMINATION, 2017 <br> <br> M.Sc. AGRICULTURAL BIOTECHNOLOGY <br> <br> M.Sc. AGRICULTURAL BIOTECHNOLOGY <br> <br> \section*{[ Field of Study Code : BAG ]} 

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Time Allowed: 3 hours
Centre of Eram. :

Name of Candidate : $\qquad$

## INSTRUCTIONS FOR CANDIDATES

Candidates must read carefully the following instructions before attempting the Question Paper :
(i) Write your Name and Registration Number in the space provided for the purpose on the top of this Question Paper and in the Answer Sheet.
(ii) Please darken the appropriate Circle of Question Paper Series Code on the Answer Sheet.
(iii) The Question Paper is divided into two Parts : Part-A and Part-B. Both Parts have multiple-choice questions. All answers are to be entered in the Answer Sheet provided with the Question Paper for the purpose.
(iv) Part-A consists of 60 questions and all are compulsory. Answer all the questions in the Answer Sheet provided for the purpose by darkening the correct choice, i.e., (a) or (b) or (c) or (d) with BALLPOINT PEN only against each question in the corresponding circle. Each correct answer carries 1 mark. There will be negative marking and $1 / a$ mark will be deducted for each wrong answer.
(v) Part-B consists of 100 questions. Answer any 60 questions in the Answer Sheet by darkening the correct choice, i.e., (a) or (b) or (c) or (d) with BALLPOINT PEN only against the corresponding circle. Each correct answer carries 3 marks. There will be negative marling and 1 mark will be deducted for each wrong answer.
In case any candidate answers more than the required 60 questions, the first 60 questions attempted will be evaluated.
(vi) Answer written by the candidates inside the Question Paper will not be evaluated.
(vii) Calculators and Log Tables may be used.
(viii) Pages at the end have been provided for Rough Work.
(ix) Return the Question Paper and Answer Sheet to the Invigilator at the end of the Entrance Examination. DO NOT FOLD THE ANSWER SHEET.

## INSTRUCTIONS FOR MARKING ANSWERS

1. Use only Blue/Black Ballpoint Pen (do not use Pencil) to darken the appropriate Circle.
2. Please darken the whole Circle.
3. Darken ONLY ONE CIRCLE for each question as shown in example below :

| Wrong | Wrong | Wrong | Wrong | Correct |
| :---: | :---: | :---: | :---: | :---: |
| 0 (b) © | © (b) © (d) | © (b) (c) | O (b) © 0 | (a) (b) © 0 |

4. Once marked, no change in the answer is allowed.
5. Please do not make any stray marks on the Answer Shect.
6. Please do not do any rough work on the Answer Sheet.
7. Mark your answer only in the appropriate space against the number corresponding to the question.
8. Ensure that you have darkened the appropriate Circle of Question Paper Series Code on the Answer Sheet.

PART-A

## Answer all questions

1. If the temperature of the sun were to increase from $T$ to $2 T$ and its radius from $R$ to $2 R$, then the ratio of the radiant energy received on the earth to what it received previously will be
(a) 4
(b) 64
(c) 16
(d) 32
2. Which of the following is a redox reaction?
(a) $\mathrm{NaCl}+\mathrm{KNO}_{3} \rightarrow \mathrm{NaNO}_{3}+\mathrm{KCl}$
(b) $\mathrm{CaC}_{2} \mathrm{O}_{4}+2 \mathrm{HCl} \rightarrow \mathrm{CaCl}_{2}+\mathrm{H}_{2} \mathrm{C}_{2} \mathrm{O}_{4}$
(c) $2 \mathrm{~K}\left[\mathrm{Ag}(\mathrm{CN})_{2}\right]+\mathrm{Zn} \rightarrow 2 \mathrm{Ag}+\mathrm{K}_{2}\left[\mathrm{Zn}\left(\mathrm{CN}_{4}\right]\right.$
(d) $\mathrm{Ca}(\mathrm{OH})_{2}+2 \mathrm{NH}_{4} \mathrm{Cl} \rightarrow \mathrm{CaCl}_{2}+2 \mathrm{NH}_{3}+2 \mathrm{H}_{2} \mathrm{O}$
3. Which of the following concentration factors is affected by change in temperature?
(a) Molarity
(b) Molality
(c) Mole fraction
(d) Weight fraction
4. The total number of neutrons in $\mathrm{Zn}^{2+}$ with mass number 70 is
(a) 34
(b) 40
(c) 36
(d) 38
5. The electron identified by quantum numbers $n$ and $l$, (i) $n=4, l=1$; (ii) $n=4, l:=0$; (iii) $n=3, l=2$ and (iv) $n=3, l=1$ can be placed in the order of increasing energy, from the lowest to the highest as
(a) (iv) $<$ (ii) $<$ (iii) $<$ (i)
(b) (ii) $<$ (iv) $<$ (i) $<$ (iii)
(c) (i) $<$ (iii) $<$ (ii) $<$ (iv)
(d) (iii) $<$ (i) $<$ (iv) $<$ (ii)
6. Which one of the following constitutes a group of the isoelectronic series?
(a) $\mathrm{C}_{2}^{-2}, \mathrm{O}_{2}^{-}, \mathrm{CO}, \mathrm{NO}$
(b) $\mathrm{NO}^{+}, \mathrm{C}_{2}^{-2}, \mathrm{CN}^{-}, \mathrm{N}_{2}$
(c) $\mathrm{CN}^{-}, \mathrm{N}_{2}, \mathrm{O}_{2}^{-2}, \mathrm{C}_{2}^{-2}$
(d) $\mathrm{NO}^{+}, \mathrm{N}_{2}, \mathrm{O}_{2}^{-}, \mathrm{CO}$
7. Which of the following represents the correct order of increasing first ionization enthalpy for $\mathrm{Ca}, \mathrm{Ba}, \mathrm{S}, \mathrm{Se}$ and Ar ?
(a) $\mathrm{Ca}<\mathrm{S}<\mathrm{Ba}<\mathrm{Se}<\mathrm{Ar}$
(b) $\mathrm{Ca}<\mathrm{Ba}<\mathrm{S}<\mathrm{Se}<\mathrm{Ar}$
(c) $\mathrm{S}<\mathrm{Se}<\mathrm{Ca}<\mathrm{Ba}<\mathrm{Ar}$
(d) $\mathrm{Ba}<\mathrm{Ca}<\mathrm{Se}<\mathrm{S}<\mathrm{Ar}$
8. The process requiring the absorption energy is
(a) $\mathrm{F} \rightarrow \mathrm{F}^{-}$
(b) $\mathrm{Cl} \rightarrow \mathrm{Cl}^{-}$
(c) $\mathrm{O} \rightarrow \mathrm{O}^{-2}$
(d) $\mathrm{H} \rightarrow \mathrm{H}^{-}$
9. The calculated bond order of $\mathrm{O}^{-2}$ is
(a) 2.5
(b) 1.5
(c) 2.0
(d) 1.0
10. For which of the following molecules significant $\mu \neq 0$ ?

(ii)

(iii)

(iv)

(a) Only (i)
(b) (i) and (ii)
(c) (iii) and (iv)
(d) Only (iii)
11. In which of the following changes does entropy decrease?
(a) Crystallization of sucrose from solution
(b) Dissolution of sucrose in water
(c) Melting of ice
(d) Vaporization of camphor
12. If the enthalpy change for the transition of liquid $\mathrm{H}_{2} \mathrm{O}$ to steam is $30 \mathrm{~kJ} \mathrm{~mol}^{-1}$ at $27^{\circ} \mathrm{C}$, the entropy change for the process would be
(a) $1.0 \mathrm{~J} \mathrm{~mol}^{1} \mathrm{~K}^{1}$
(b) $10 \mathrm{~J} \mathrm{~mol}^{1} \mathrm{~K}^{1}$
(c) $0.1 \mathrm{~J} \mathrm{~mol}^{1} \mathrm{~K}^{1}$
(d) $100 \mathrm{~J} \mathrm{~mol}^{1} \mathrm{~K}^{1}$
13. The concentration of $\mathrm{NH}_{4} \mathrm{Cl}$ and $\mathrm{NH}_{4} \mathrm{OH}$ in buffer solution is in the ratio $1: 1, K_{b}$ of $\mathrm{NH}_{4} \mathrm{OH}$ is $10^{-10}$. The pH of the buffer is
(a) 4
(b) 5
(c) 9
(d) 10
14. Identify the product for the following transformation :

$$
\mathrm{PhMgBr}+\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH} \rightarrow \text { Product }
$$

(a) $\mathrm{Ph} \cdot \mathrm{H}$
(b) $\mathrm{Ph}-\mathrm{OH}$
(c) $\mathrm{Ph}-\mathrm{OCH}_{2} \mathrm{CH}_{3}$
(d) $\mathrm{Ph}-\mathrm{CH}_{2} \mathrm{CH}_{3}$
15. Natural glucose is termed as D-glucose because
(a) it is dextrorotatory
(b) it is based on D-glyceraldehyde Fischer projection
(c) it is based on Newman projection
(d) None of the above
16. An enantiomerically pure acid is treated with racemic mixture of an alcohol having one chiral carbon. The ester formed will be
(a) enantiomeric compound
(b) optically active mixture
(c) optically inactive mixture
(d) Difficult to predict due to insufficient information
17. One agro industry produces a certain number of manure bags in a day. It was observed on a particular day that the cost of production of each manure bag (in F) was 3 more than twice the number of manure bags produced on that day. If the total cost of production on that day was $F 90$, then find the number of articles produced and the cost of each article.
(a) 3 and F30
(b) 4 and F 20
(c) 6 and $F 15$
(d) 5 and $F 18$
18. The probability that a non-leap year has 53 Sundays, is
(a) $2 / 7$
(b) $5 / 7$
(c) $6 / 7$
(d) $1 / 7$
19. In an experiment, tubes numbered 1 to 20 are mixed up and then the tube is drawn at random. What is the probability that the tube drawn bears a number which is a multiple of 3 ?
(a) $1 / 5$
(b) $2 / 5$
(c) $3 / 10$
(d) $3 / 5$
20. Mode is
(a) middle most value
(b) most frequent value
(c) least frequent value
(d) average value
21. Which of the following is not a measure of central tendency?
(a) Standard deviation
(b) Mode
(c) Mean
(d) Median
22. If the difference of mode and median of a data is 24 , then the difference of median and mean is
(a) 12
(b) 24
(c) 8
(d) 36
23. Jadeja scores runs in 10 consecutive innings as 38, 70, 48, 34, 42, 55, 63, 46, 54 and 44. The mean deviation about mean is
(a) 8.6
(b) 6.4
(c) $10 \cdot 6$
(d) 7.6
24. A pair of dice is rolled. If the outcome is a doublet, a coin is tossed. Determine the total number of elementary events associated to this experiment.
(a) 21
(b) 42
(c) 40
(d) 32
25. Find the probability that in a random arrangement of the letters of the word UNIVERSITY', the two I's do not come together.
(a) $3 / 4$
(b) $1 / 5$
(c) $4 / 5$
(d) $2 / 5$
26. 100 students appeared for two examinations. 60 passed the first, 50 passed the second and 30 passed both. Find the probability that a student selected at random has passed at least one examination.
(a) $4 / 5$
(b) $1 / 4$
(c) $2 / 3$
(d) $3 / 4$
27. If the fifth term of a GP (Geometric Progression) is 2 , then write the product of its 9 terms.
(a) 613
(b) 512
(c) 812
(d) 532
28. A line passes through the point $(2,2)$ and is perpendicular to the line $3 x+y=3$. Its $y$-intercept is
(a) $1 / 3$
(b) $4 / 3$
(c) $2 / 3$
(d) 1
29. The figure formed by the lines $a x \pm b y \pm c=0$ is
(a) a rectangle
(b) a square
(c) a rhombus
(d) a triangle
30. The equation $16 x^{2}+y^{2}+8 x y+74 x+78 y+212=0$ represents
(a) an ellipse
(b) a circle
(c) a hyperbola
(d) a parabola
31. Find the sum of odd integers from 1 to 2001.
(a) 1002001
(b) 1000078
(c) 1280011
(d) 1000203
32. The male gametes of a corn plant have 10 chromosomes in their nucleus. The chromosome number in the female gamete, zygote and cells of the progeny of the plant will be respectively
(a) $10,20,10$
(b) $20,10,10$
(c) $10,20,20$
(d) $20,10,20$
33. Which of the following is used as an atmospheric pollution indicator?
(a) Lepidoptera
(b) Lycopersicon
(c) Lichen
(d) Ocimum
34. All genes located on the same chromosome will
(a) form different groups depending upon their relative distance
(b) form one linkage group
(c) not form any group
(d) form reactive groups that affect the phenotype
35. Mendel's law of independent assortment holds good for genes situated on
(a) non-homologous chromosome
(b) homologous chromosome
(c) chloroplast genome
(d) same chromosome
36. In $F_{2}$ generation of a Mendelian dihybrid cross, the number of phenotypes and genotypes are
(a) phenotypes-4; genotypes-16
(b) phenotypes-9; genotypes-4
(c) phenotypes-4; genotypes-8
(d) phenotypes-4; genotypes-9
37. Appearance of antibiotic-resistant bacterium is an example of
(a) adaptive radiation
(b) transduction
(c) pre-existing variation in the population
(d) divergent evolution
38. Charles Darwin proposed the theory of natural selection based on the knowledge of
(a) spontaneous mutation
(b) phenotypic variation
(c) induced mutation
(d) chromosomal aberration
39. The cotyledon of a monocot is known as
(a) scutellum
(b) endosperm
(c) aleurone layer
(d) epithelial cell
40. The microgametophyte is also known as
(a) pollen
(b) anther
(c) ovule
(d) ovary
41. Haploid plants can be obtained from a
(a) leaf
(b) root
(c) pollen
(d) stem
42. $\qquad$ is a specialized human cell devoid of nucleus.
(a) Skin cell
(b) Red blood cell
(c) Liver cell
(d) Muscle cell
43. The term 'vaccine' was coined by
(a) Edward Jenner
(b) Joseph Lister
(c) Louis Pasteur
(d) Robert Koch
44. Fatty acid breakdown in eukaryote occurs in
(a) cytosol
(b) Golgi body
(c) mitochondria
(d) peroxisome
45. Conversion of a normal cell into a cancer cell is called as
(a) karyokinesis
(b) cytokinesis
(c) carcinogenesis
(d) euthanasia
46. Which of the following refers to the addition of microorganisms to the diet in order to provide health benefits beyond basic nutritive value?
(a) Probiotic
(b) Prebiotic
(c) Antibiotic
(d) Adjuvant
47. Which of the following is the correct order of basic forces in decreasing strength?
(a) Strong, Electromagnetic, Weak, Gravitational
(b) Strong, Electromagnetic, Gravitational, Weak
(c) Electromagnetic, Strong, Weak, Gravitational
(d) Electromagnetic, Strong, Gravitational, Weak
48. Which one of the following is not correct about Newton's second law of motion?
(a) It holds good only in inertial frame of reference
(b) It gives a measure of force as the rate of change of momentum
(c) It cannot describe the motion of particles with relativistic velocities
(d) It can describe the motion of a falling rain drop, which may gather mass as it falls
49. A ball attains a height of $h$, if thrown upward with some initial speed. If the ball is thrown upward by the double of that initial speed, what new maximum height will the ball reach?
(a) $2 h$
(b) $4 h$
(c) $8 h$
(d) $16 h$
50. What will be the change in the entropy of a gas if it expands adiabatically and reversibly?
(a) Increase infinitely
(b) Increase, but remain finite
(c) Decrease
(d) No change
51. Internal energy of an ideal gas depends upon
(a) temperature only
(b) both temperature and pressure
(c) volume only
(d) both volume and temperature
52. Which one of the following is not correct about the work done by a conservative force?
(a) Work done is independent of the path
(b) For a cyclic path, work done is zero
(c) Force is velocity-independent
(d) Work done is irreversible
53. In an explosion, an object of mass $M$ that was initially at rest splits into two pieces of unequal mass ( $m_{1}>m_{2}$ ). Which of the following is correct?
(a) $m_{1}$ will have greater momentum than $m_{2}$
(b) $m_{1}$ will have greater kinetic energy than $m_{2}$
(c) $m_{1}$ will have lower kinetic energy than $m_{2}$
(d) both pieces will have the same kinetic energy
54. The specific heat of a gas
(a) has only two values
(b) has a unique value at a given temperature
(c) can have any value between 0 and $\infty$
(d) depends upon the mass of the gas
55. The volume of a polyatomic gas is compressed adiabatically to $\frac{1}{8}$ of its original volume. If the original pressure of the gas was $P_{0}$, the new pressure will be
(a) $2 P_{0}$
(b) $4 P_{0}$
(c) $8 P_{0}$
(d) $16 P_{0}$
56. The root-mean-square speed of the molecules of a gas enclosed in a vessel is $v$. If the pressure is doubled while the temperature remains the same, the root-mean-square speed will be
(a) $\frac{\nu}{2}$
(b) $v$
(c) $2 v$
(d) $4 v$
57. A vessel contains a mixture of one mole of oxygen and two moles of nitrogen at 300 K . The ratio of the average rotational kinetic energy per $\mathrm{O}_{2}$ molecule to that per $\mathrm{N}_{2}$ molecule will be
(a) $1: 1$
(b) $8: 7$
(c) $1: 2$
(d) Depends upon the moments of inertia of two molecules
58. $C_{v}$ is the specific heat of a system at constant volume $V$. Which of the following (thermodynamic) equations is wrong to obtain $C_{v}$ ?
(a) $C_{\nu}=\sigma_{E}^{2} /\left(k T^{2}\right)$
(b) $C_{v}=\partial H / \partial T$
(c) $C_{\nu}=-T\left(\partial^{2} F / \partial T^{2}\right)$
(d) $C_{\nu}=T(\partial S / \partial T)$

Note : $F=$ Helmholtz free energy, $H=$ enthalpy, $S=$ entropy, $T=$ absolute temperature, $\sigma_{E}^{2}=$ variance of the internal energy
59. In quantum mechanics, which of the following statements is not always a characteristic of a 1 D particle wave-function $\psi(x)$ ?
(a) $\left|\psi(x)^{2}\right|$ is a probability density
(b) $\int_{-\infty}^{+\infty}\left|\psi(x)^{2}\right| d x=1$
(c) $\psi(x)$ is a solution of the 1D stationary Schrödinger equation $H \psi(x)=E \Psi(x)$, where $H$ is the Hamiltonian operator
(d) The energy $E$, obtained from the Schrödinger equation, belongs to a set of discrete values
60. Which of the following states that each point on a wavefront may be considered as a new wave source?
(a) Snell's law
(b) Young's law
(c) Hertz's law
(d) Huygens' principle

## PART-B

Answer any sixty questions
.
61. UPOV was established in the year 1961 with its headquarters at
(a) Geneva
(b) Rome
(c) Zurich
(d) New Delhi
62. A micronutrient which is predominant in alkaline pH is
(a) zinc
(b) iron
(c) manganese
(d) molybdenum
63. A chemical ameliorant to reclaim alkali-sodic soils is
(a) gypsum
(b) dolomite
(c) lime
(d) zeolite
64. Cat clay is present in $\qquad$ type of soils.
(a) acid
(b) acid sulphate
(c) sodic
(d) saline sodic
65. Soil that is developed as a result of consolidation of molten magma is
(a) igneous rock
(b) metamorphic rock
(c) sedimentary rock
(d) primary rock
66. When two organisms live in close association and both benefit, the relationship between the two is called
(a) commensalism
(b) symbiosis
(c) parasitism
(d) pathogenesis
67. Which of the following is a nitrifying bacterium?
(a) Rhizobium
(b) Nitrosomonas
(c) Bacillus
(d) Azotobacter
68. Nodulation in non-legume plants is caused by
(a) Frankia
(b) Sinorhizobium
(c) BGA
(d) Bradyrhizobium
69. Which one of the following forms symbiosis with Azolla?
(a) Nostoc
(b) Anabaena
(c) Tolypothrix
(d) Oscillatoria
70. The 70S prokaryotic ribosome consists of
(a) two $40 S$ subunits
(b) 405 and 305 subunits
(c) 50 S and 20 S subunits
(d) 50 S and 30 S subunits
71. Organisms belonging to the same species living together in the same place at the same time are a/an
(a) niche
(b) community
(c) population
(d) ecosystem
72. _describes the treatment of poliuted soils through use of plants which mitigate the environmental problem without the need to remove and dispose them in an another place.
(a) Phytostimulation
(b) Phytoremediation
(c) Detoxification
(d) Oxidation
73. What does the student's $t$-test measure?
(a) Difference between two means
(b) Difference between three or more means
(c) Difference between two frequency distribution
(d) Whether two distributions are independent or dependent
74. ANOVA is a test of equality of
(a) variances
(b) means
(c) proportions
(d) only two parameters
75. When all members of every block are assigned to all treatments, the design is called
(a) Latin square design
(b) one-way ANOVA
(c) repeated measures design
(d) randomized complete block design
76. Median is defined as
(a) the value that half of the entries are below and half of the entries are above
(b) the value that has lowest frequency
(c) the largest value of the entries
(d) the average calculated by adding up all the values and deviating by the number of entries
77. What are the assumptions of Hardy-Weinberg equilibrium?
(a) Small population size, random mating, no selection, migration, no mutation
(b) Large population size, non-random mating, no selection, no migration, no mutation
(c) Large population size, random mating, no selection, no migration, no mutation
(d) Large population size, random mating, heterozygotes survive the best, no migration, no mutation
78. The progeny of single-self fertilized homozygous individuals is known as
(a) synthesis
(b) clone
(c) pure line
(d) hybrid
79. Dieocy is a mechanism which promotes
(a) self-pollination
(b) cross-pollination
(c) both self- and cross-pollination
(d) self-incompatibility
80. $\qquad$ trait is genetically improved in Swarna Sub 1 through marker-assisted backcross breeding.
(a) Flood tolerance
(b) Salinity tolerance
(c) Drought tolerance
(d) Insect resistance
81. The specific biomolecules which exhibit detectable differences among the different genotypes of plants are termed as
(a) DNA fingerprinting
(b) molecular markers
(c) molecular scissors
(d) transcripts
82. Tihe quickest way to produce homozygous breeding lines from heterozygous parents is through
(a) aneuploidy
(b) polyploidy
(c) double haploidy
(d) introgression
83. Green Revolution in India occurred during
(a) 1950 s
(b) 1960 s
(c) 1970s
(d) 1980 s
84. Multiple effects of a single-gene is known as
(a) polyploidy
(b) heteroploidy
(c) pleiotropy
(d) aneuploidy
85. If the incompatibility reaction of pollen is governed by the genotypes of the plant on which the pollen is produced, then the self-incompatibility is
(a) gametophytic self-incompatibility
(b) sporophytic self-incompatibility
(c) pseudo-self-incompatibility
(d) ritactional self-incompatibility
86. Repeat core sequences consisting of 2,3 or 4 base pairs are known as
(a) SNPs
(b) microsatellites
(c) AFLPs
(d) satellites
87. Blood red is a variety of
(a) mango
(b) mandarin
(c) pineapple
(d) sweet orange
88. Dichogamy is observed in
(a) Aonla
(b) Ber
(c) avocado
(d) grapes
89. The Central Tuber Crops Research Institute is located at
(a) Delhi
(b) Shimla
(c) Thiruvananthapuram
(d) Kasaragod
90. Ben oil is obtained from
(a) moringa
(b) potato
(c) cassava
(d) pumpkin
91. ___ is the rich source of folic acid.
(a) Tomato
(b) Brinjal
(c) Chilly
(d) Amaranth
92. Genetically engineered mustard developed by the Delhi University is modified for which one of the following traits?
(a) Insect resistance
(b) Oil quality
(c) Male sterility
(d) Drought tolerance
93. Which of the following is true of DNA double helix?
(a) Total number of purine bases is equal to the total number of pyrimidine bases
(b) The $5^{\prime}$ end of one DNA strand is aligned with $5^{\prime}$ end of the other strand
(c) The sequence of bases on one strand is identical to the sequence of bases on the other strand
(d) None of the above
94. Of the four different DNA molecules below, which one would you expect to denature at a lower temperature?
(a) GCATTGCCAATGC
(b) ATTAGCCTATCGG
(c) GCCACCGAATCCG
(d) ATATTTTTACTGCC
95. The $3^{\prime}$ to $5^{\prime}$ exonuclease activity of E. coli DNA polymerases I and III allows
(a) polymerase to synthesize a new strand in $3^{\prime}$ to $5^{\prime}$ direction
(b) polymerase to remove misincorporated nucleotides thereby reducing the number of errors made during DNA replication
(c) polymerase to join Okazaki fragments
(d) None of the above
96. If the following parental DNA strand

## $3^{\prime}$ GGCATATTCGCTGCAGT 5'

is used as a template DNA strand, the newly synthesized, antiparallel strand would be as follows :
(a) $3^{\prime}$ CCGTATAAGCGACGTCA 5'
(b) 5' CCGTATAAGCGACGTCA $3^{\prime}$
(c) $5^{\prime}$ TGACGTCGCTTATACGG $3^{\prime}$
(d) $3^{\prime}$ GGCATATTCGCTGCAGT 5'
97. Transcription is the process of
(a) synthesizing a DNA molecule from RNA template
(b) using DNA strand as a template to synthesize a complementary RNA molecule
(c) using DNA strand as a template to synthesize identical RNA molecule
(d) synthesizing a protein using information carried in the nucleotide sequence of a messenger RNA
98. DNA ligase
(a) unwinds the helical DNA by breaking the hydrogen bonds between complementary bases
(b) adds DNA nucleotides to the RNA primer
(c) links the DNA fragments of the lagging strands together
(d) synthesizes a short-RNA primer at the beginning of each origin of replication
99. The direction of synthesis of a new mRNA molecule is
(a) $5^{\prime}$ to $3^{\prime}$ from a $5^{\prime}$ to $3^{\prime}$ DNA template strand
(b) 5' to $3^{\prime}$ from a $3^{\prime}$ to $5^{\prime}$ RNA template strand
(c) 5' to $3^{\prime}$ from a $5^{\prime}$ to $3^{\prime}$ RNA template strand
(d) 5' to $3^{\prime}$ from a $3^{\prime}$ to $5^{\prime}$ DNA template strand
100. In polyadenylation, a long tail of adenine residue is added
(a) to the $3^{\prime}$ end of template DNA
(b) to the $5^{\prime}$ end of mRNA
(c) to the $3^{\prime}$ end of mRNA
(d) None of the above
101. Which of the following is/are not needed by the RNA polymerase?
(a) A primer
(b) GTP, CTP, UTP, ATP
(c) A promoter sequence
(d) A DNA template
102. Which region of an introns' nucleotide sequence is important?
(a) Whole intron sequence
(b) Short sequence near each end of the intron
(c) A 25-base repeat in the middle of intron sequence
(d) None of the above
103. Cry 1 group of protein of Bacillus thuringiensis is toxic to
(a) Lepidoptera
(b) Homoptera
(c) Coleoptera
(d) Hymenoptera
104. The $\qquad$ in a tRNA molecule base pair with a group of three nucleotides is called the $\qquad$ in the mRNA template molecule.
(a) codon, anticodon
(b) anticodon, codon
(c) codon, triplet codon
(d) triplet codon, anticodon
105. An mRNA that gives rise to multiple different proteins is said to be a
(a) monocistronic
(b) polycistronic
(c) Shine-Dalgarno sequence
(d) None of the above
106. The order in which the triplet of bases in an mRNA is read is most precisely called
(a) reading frame
(b) anticodon
(c) codon
(d) wobble
107. Phage M13 DNA is
(a) single-stranded circular
(b) single-stranded linear
(c) double-stranded circular
(d) double-stranded linear
108. Lambda phage DNA is introduced into host cells efficiently by a laboratory process is called
(a) transformation
(b) transfection
(c) in vitro packaging
(d) electroporation
109. The vector that can clone about 300 kbp DNA fragment in $E$. coli is
(a) BAC
(b) YAC
(c) PACMID
(d) Cosmid
110. T-DNA is
(a) DNA of plasmid origin which is transferred to the Agrobacterium chromosome
(b) DNA from the chromosome of Agrobacterium species which is transferred to the plant genome
(c) DNA of plasmid origin which is transferred to the plant genome
(d) DNA of plant genome that is transferred to Agrobacterium
111. Starch is a polymer made up of the monomer
(a) $\alpha$-D-glucose
(b) $\alpha$-D-fructose
(c) $\beta$-D-glucose
(d) $\beta$-D-fructose
112. Hardy-Weinberg equilibrium generally assumes all of the following, except
(a) a large population
(b) genetic drift
(c) absence of selection
(d) random mating

113 The epimer of glucose is
(a) galactose
(b) fructose
(c) arabinose
(d) ribose
114. Which of the following may be absent in a tree log having 33 annual rings?
(a) Primary phloem
(b) Primary xylem
(c) Sclerenchyma cell wall
(d) Interfascicular cambium
115. In glycolytic pathway, under aerobic conditions glucose is converted to
(a) pyruvate
(b) lactate
(c) acetate
(d) formate
116. Pick out the purine nucleotide from the following :
(a) TTP
(b) GTP
(c) CTP
(d) UTP
117. Seedless bananas are
(a) diploid
(b) triploid
(c) tetraploid
(d) hexaploid
118. Which of the following is true sequence of events in a plant transformation experiment?
(a) Preculture, cocultivation, callus formation, shoot development, plantlets
(b) Cocultivation, callus formation, shoot development, preculture, plantlets
(c) Preculture, shoot development, cocultivation, callus formation, plantlets
(d) Explants, callusing, shoot formation, Agrobacterium infection, callus formation, plantlets
119. Limiting amino acid in cereals is
(a) leucine
(b) lysine
(c) methionine
(d) glycine
120. Klenow fragment is
(a) a type of ligase
(b) a large fragment of DNA polymerase I
(c) a DNA polymerase III
(d) an RNA polymerase
121. The linkage between two nucleotides in DNA is
(a) $3^{\prime}, 5^{\prime}$ phosphodiester bond
(b) 2', 5' phosphodiester bond
(c) peptide bond
(d) $3^{\prime}, 6^{\prime}$ phosphodiester bond
122. For generation of shoot, callus can be transferred onto the MS medium that has
(a) high concentration of auxin
(b) high ratio of cytokinin to auxin
(c) high ratio of auxin to cytokinin
(d) required amount of antibiotic
123. Identify the accessory photosynthetic plant pigment :
(a) Chlorophyll
(b) Carotenoid
(c) Lycopene
(d) Betalain
124. Identify the start codon :
(a) AUG
(b) UAA
(c) UAG
(d) UGA
125. If we cut the flowers under the water, they will remain fresh for a longer time because
(a) water is always available for the stems
(b) water columns are not blocked by air bubbies
(c) water is always available to heavily transpiring flowers
(d) by cutting, vascular column is lost from flower stems
126. Which of the following hormones is associated with abiotic stress conditions and is known as stress hormone?
(a) Abscisic acid
(b) Ethylene
(c) $\mathrm{GA}_{3}$
(d) Indole-3-acetic acid
127. The endosymbiotic theory concerns the origin of
(a) endoplasmic reticulum
(b) nucleus
(c) Golgi bodies and vacuoles
(d) mitochondria and chloroplast
128. $\qquad$ is referred as 'traffic police' of the cell.
(a) Golgi body
(b) Vacuole
(c) Lysosome
(d) Vesicle
129. Which of the following can be used for the production of haploid plants?
(a) Meristematic tissue
(b) Anthers
(c) Young embryo
(d) Flower buds
130. Which of the following is oil-storing plastids in plant cell?
(a) Amyloplast
(b) Elaioplast
(c) Chromoplast
(d) Chloroplast
131. Which one of the following conditions inhibits the TCA cycle?
(a) Anaerobic
(b) Salinity
(c) Moisture stress
(d) Darkness
132. Peroxisomes involve in
(a) dark respiration
(b) photorespiration
(c) photosynthesis
(d) lipid synthesis
133. Phloem transport mainly takes in
(a) sieve cells
(b) companion cells
(c) phloem parenchyma
(d) Ray cells
134. Foot and mouth disease is a $\qquad$ disease of cattle.
(a) viral
(b) fungal
(c) bacterial
(d) parasitic
135. __ is/are used for bulk flow of materials between plant cells.
(a) Osmosis
(b) Diffusion
(c) Plasmodesmata
(d) Nucleopores
136. Identify the event that exclusively occurs in meiotic cell division :
(a) Chromatid formation
(b) Spindle formation
(c) Synapsis
(d) Chromosome movement to pole
137. Khaira disease of rice is caused due to the deficiency of $\qquad$ element.
(a) iron
(b) manganese
(c) zinc
(d) copper
138. An example for CAM plant is
(a) apple
(b) pineapple
(c) custard apple
(d) cashew
139. The growth regulator involved in seed germination is
(a) auxin
(b) GA
(c) ethylene
(d) cytokinin
140. The specific role of boron in plants is
(a) synthesis of auxins
(b) transport of proteins
(c) activation of enzymes
(d) transport of sugars
141. Compound present in vascular and nonvascular plants responsible for the elongation of stem cells is
(a) polyamine
(b) brassinolide
(c) salicylic acid
(d) jasmonic acid
142. Which one of the following is a non-climacteric fruit?
(a) Banana
(b) Jackfruit
(c) Mango
(d) Citrus
143. Which of the following cell organelles has a single-layer membrane?
(a) Vacuole
(b) Mitochondria
(c) Chloroplast
(d) Nucleus
144. Which of the following soil types is suitable for cotton crop?
(a) Red soil
(b) Black soil
(c) Sandy soil
(d) Saline soil
145. Raising the same crop in the same field in all the seasons is called
(a) mixed cropping
(b) monocropping
(c) relay cropping
(d) ratoon cropping
146. The biofertilizer recommended for pulse crops is
(a) Azospirillum
(b) Phosphobacteria
(c) Azolla
(d) Rhizobium
147. Direct or indirect harmful effect that one plant has on another through production of chemical substance is
(a) excretion
(b) toxicity
(c) allelopathy
(d) interference
148. The herbicide used as growth regulator is
(a) glyphosate
(b) 2,4-D
(c) pendimethalin
(d) bialophos
149. Chemical used for seed hardening in sorghum is
(a) $\mathrm{ZnSO}_{4}$
(b) $\mathrm{CaCl}_{2}$
(c) $\mathrm{KH}_{2} \mathrm{PO}_{4}$
(d) $\mathrm{FeSO}_{4}$
150. Whitefly belongs to the order
(a) Heteroptera
(b) Homoptera
(c) Diptera
(d) Lepidoptera
151. Banana bunchy top disease is caused by a
(a) virus
(b) fungus
(c) bacteria
(d) mycoplasma
152. Imidacloprid is a/an
(a) avermectin
(b) neonicotinyl compound
(c) organophosphate compound
(d) organochlorine compound
153. Trichogramma is a parasitoid on
(a) egg
(b) larva
(c) pupa
(d) Both egg and larva
154. Disease which interferes with the conduction of water in plants is
(a) wilt
(b) rot
(c) leaf spot
(d) damping off
155. Blast disease of rice is caused by
(a) Magnaporthe grisea
(b) Drechslera oryzae
(c) Ephelis oryzae
(d) Rhizoctonia solani
156. The famous Irish faminine of 1842 was caused due to the outbreak of
(a) helminthosporium of rice
(b) stem rust of wheat
(c) late blight of potato
(d) downy mildew of grapes
157. The common viral disease affecting the sheep is
(a) bluetongue
(b) enterotoxaemia
(c) pasteurella
(d) enteritis
158. In Pisum, type of germination is
(a) epigeal
(b) hypogeal
(c) both epigeal and hypogeal
(d) hypoepigeal
159. $\mathrm{GA}_{3}$ is sprayed in hybrid rice to enhance
(a) panicle exertion
(b) seed set
(c) fertility
(d) flowering
160. The national repository for plant genes is
(a) NBPGR
(b) PPV \& FRA
(c) OECD
(d) ISST

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