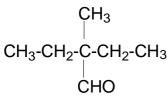
## **CHEMISTRY**

46.	The number of electrons present on the oil drop which has the static electric charge of $-3.02044 \times 10^{-19}$ C is						
	1) 2	2) 5	3) 7	4) 8			
<b>47.</b>	7. Electrons are emitted with zero velocity from a metal surface when it is exposed to radiate						
	of wavelength 4000 $A^{\scriptscriptstyle 0}$ . The threshold frequency is						
	1) $5 \times 10^{34} \text{ sec}^{-1}$	2) $7.5 \times 10^{14} \text{ sec}^{-1}$	3) $9.2 \times 10^{14} \text{ sec}^{-1}$	4) $7.5 \times 10^{34} \text{ sec}^{-1}$			
48.				d you place this element			
40	1) IV A	2) VII A	3) II A	4) V A			
49.	9						
	•		2) Radius : $I^+ < I > I^-$				
	3) $E.N$ : $F < Cl < Br$		4) $EA$ : $S > Se > O$				
50.		•	tom (A) contains two bo	nd pairs (B) and three			
	lone pairs(L). Then th	•		4) T *			
<i>E</i> 1	1) T-shape	2) See saw	3) V-shape	4) Linear			
51.	oxygen is	e, the ratio of Kineuc (	energy of 3 gram of hyd	rogen and 4 gram of			
	1) 1:12	2) 12:1	3) 5:6	4) 3:4			
<b>52.</b>	The prefix "pico" repr	resents the multiple of					
	1) 10 <sup>-9</sup>	2) 10 <sup>-12</sup>	3) $10^{-18}$	4) $10^{-21}$			
53.	The specific gravity of 84% $\left(\frac{w}{w}\right)$ $H_2SO_4$ is 1.752. The normality of solution is						
	1) 30.03 N	2) 2.05N	3) 39.5N	4) 4.5N			
54.	The Oxidation State of	f $Fe$ in $[Fe(H_2O)_5 NO]$	eg (Brown ring) is				
	1) +1	2) +2	3) +3	4) +6			
55.	In which of the followi	$ng \Delta H > \Delta E$					
	1) $H_2 + I_2 \longrightarrow 2HI$ (g) (g)		2) $N_2 + 3H_2 \longrightarrow 2NH_3$ (g) (g)				
<b>E</b> (	3) $PCl_5 \Longrightarrow PCl_3 + Cl_2$ (g) (g) (g)		4) $2SO_2 + O_{(g)} \longrightarrow 2SO_{(g)}$	<b>O</b> <sub>3</sub>			
56.	Identify the salt whose	-		A) UCM			
	1) <i>NH</i> <sub>4</sub> <i>Cl</i>	2) $CuSO_4$	3) $Al_2(SO_4)_3$	4) <i>KCN</i>			
57.			ion exchange resin in sy				
<b>50</b>	1) <7	2) >7	3) 7	4) 10			
58.	The cation which show	•	-	4) C <sup>+</sup>			
	$1) \stackrel{Li}{_{(aq)}}^+$	2) $K^+_{(aq)}$	3) $Rb^+_{(aq)}$	4) $Cs^+$			
<b>59.</b>	For a good quality cer	nent, the ratio of silica	to Alumina should be l	oetween			
	1) 1 to 2	2) 2.5 to 4	3) 5 to 7	4) 4.5 to 7.5			
<b>60.</b>	0. $B_2H_6 + 2NH_3 \xrightarrow{120^0C} A$ where 'A' is formulated as						
	1) $\left[BH_4\right]^+ \left[BH_2\left(NH_3\right)_2\right]$		$2) \left[ BH_2 \left( NH_3 \right)_2 \right]^+ \left[ BH_4 \right]^-$				
	3) $\left[BH_3(NH_3)\right]^+ \left[BH_4(NH_3)\right]^-$ 4) $\left[BH_4(NH_3)_2\right]^+ \left[BH_2\right]^-$			$\begin{bmatrix} 1 \\ 2 \end{bmatrix}^-$			
<b>61.</b>	The basic unit of pyrosilicate is						
	1) $SiO_4^{-4}$	2) $(SiO_3)_n^{-2n}$	3) $Si_2O_7^{-6}$	4) $(Si_2O_5)_n^{-2n}$			

## SRIGAYATRI EDUCATIONAL INSTITUTIONS - AP&TS

- 62. The pollutant that causes methemoglobinemia (blue baby syndrome) is
  - 1)  $Pb^{+2}$
- 2)  $SO_4^{-2}$
- 3)  $NO_{3}^{-}$
- 4)  $AsO_3^{-3}$

**63.** The IUPAC name of



- 1) 3 Formyl 3 methyl Pentane
- 2) 2 ethyl 2 methyl butanal

3) 2 - ethyl - 2 formylbutane

- 4) 2,2 Diethyl propanal
- $CH_2 = CH CH_2OH$  and  $CH_2 = CH O CH_3$  are **64.** 

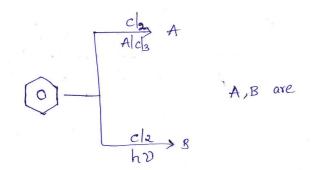
  - 1) Chain isomers 2) Position isomers
- 3) functional group isomers 4) metamers
- $A \leftarrow \frac{H_2}{Pd/BaSO_4} R C \equiv C R \frac{H_2}{Na/NH_3} \rightarrow B$ . Where A and B are respectively **65.** 
  - 1) Trans alkene, cis alkene

2) cis alkene, Trans alkene

3) cis alkene, cis alkene

4) Trans alkene, Trans alkene

**66.** 



- 1) A=chlorobenzene
- B=Chlorobenzene
- 2) A=Lindane
- B=Chlorobenzene

- 3) A=BHC
- B=BHC
- 4) A=Chlorobenzene B=Lindane

- Glycerol is purified by **67.** 
  - 1) Crystallisation

2) Fractional distillation

3) Vacuum distillation

- 4) sublimation
- **68.** The charge heated in the blast furnace contains ore, coke and limestone in the ratio by weight
  - 1) 1:2:3
- 2) 1:4:8
- 3) 8:4:1
- 4) 2:2:2
- 69. The outer electronic configuration of the element with Z=42 is
  - 1)  $5s^2 4d^4$
- 2)  $5s^1 4d^5$
- 3)  $5s^2 5p^4$
- The catalyst used in the preparation of High density polythene (HDP) is **70.** 
  - 1)  $R_3Al + TiCl_4$
- 2)  $SnCl_{4}$

- The complex Fe(CO) follows the EAN rule. Then the value of 'x' is 71.
  - 1)3

3) 5

- 72. The sum of coordination number and oxidation number of the metal 'M' in the complex
  - $\lceil M(en), (C_2O_4) \mid Cl \text{ is } \rceil$
  - 1)6

2) 7

3) 8

- 4) 9
- The IUPAC name of the wilkinsons catalyst  $\lceil RhCl(PPh_3)_3 \rceil$  is **73.** 
  - 1) Chlorotris (triphenyl phosphine) rhodium (I)
  - 2) Chlorotris (triphenyl phosphine) rhodium (IV)
  - 3) Chlorotris (triphenyl phosphine) rhodium (O)
  - 4) Chlorotris (triphenyl phosphine) rhodium (VI)
- 74. Which of the following is 100 times sweeter than sugar
  - 1) sucralose
- 2) saccharin
- 3) Aspartame
- 4) Alitame

## SRIGAYATRI EDUCATIONAL INSTITUTIONS - AP&TS

<i>75</i> .	Which of the following does not exhibit the	phenomenon of mutarotation					
	1) (+) sucrose 2) (+) Lactose	3) (+) Maltose 4) (-) Fructose					
<b>76.</b>	Which of the following varies from species to species						
	1) A=T 2) C=G	3) A+G=C+T 4) $\frac{AT}{GC}$ ratio					
77.	Which of the following is fully fluorinated partial PVC 2) Thiokol	oolymer 3) Teflon 4) Neoprene					
<b>78.</b>	Which of the following can undergo both al	, , , , , , , , , , , , , , , , , , ,					
70.	1) $CH_3 - CH_2 - CHO$ 2) $CH_3$ -CO-CH <sub>3</sub>	3) $Cl_3C - CHO$ 4) $C_6H_5CHO$					
<b>79.</b>	Primary, secondary and tertiary alcohols as	, 0 3					
17.	methods?	te distinguished by which of the following					
	1) Oxidation method	2) Lucas test					
	3) Victor meyer's method	4) All of the above					
80.	$(CH_3)_3 C - O - CH_3$ reacts with dil.HI gives	S					
	1) $\left(CH_3\right)_3 I + CH_3 OH$	$2) \left(CH_3\right)_3 COH + CH_3 I$					
	3) $(CH_3)_2 CHOH + CH_3I$	4) $(CH_3)_3 I + CH_3 I$					
81.	3/2	373					
	CH3						
	KMMOY/KOH A Heat, H30t						
	Heat, HzOt						
	CH_ CH_ CH3						
	CH_ CH_ CH3  KMNOu/KOH  Heat, H30+	B					
	Heat, H, 0+	To these weekless A and D and					
	1) C H COOH and C H COOH	In these reactions A and B are					
		2) $C_6H_5COOH$ and $C_6H_5OH$					
0.0	3) $C_6H_5COOH$ and $C_6H_5CH_2CHO$	7 0 3					
<b>82.</b>	Which of the following would not react with						
	1) Aniline	2) Methylamine					
	3) N,N-Dimethylaniline	4) N-Methyl ethanamine					
83.	Atoms of element 'B' form hcp lattice and t	those of the element 'A' occupy $\frac{2}{3}$ rd of tetrahed	.ral				
	voids. What is the formula of the compound	•					
	1) AB 2) $A_2B$	3) $A_2B_3$ 4) $A_4B_3$					
84.	The molar conductivity of 0.025mol.lit <sup>-1</sup> m	ethanoic acid is 46.1 s.cm <sup>2</sup> .mole <sup>-1</sup> . The degree of	f				
	<b>dissociation is?</b> $\left(\lambda_{(H^+)}^0 = 349.6 \text{ s.cm}^2.mole^{-1}\right)$	$= 4.1^{\circ}$ 54.6 s cm <sup>2</sup> m:1c <sup>-1</sup>					
	dissociation is: $\chi_{(H^+)} = 349.6$ s.cm .mote $\alpha$	$(Hacoo^{-}) = 34.6 \text{ s.cm} \text{ .mule}$					
	1) 0.114 2) 21.3	3) 3.66 4) 0.35					
<b>85.</b>	A first order reaction is 50% complete in 2	3min. The time required to complete 90% of the	e				
	reaction is						
	1) 23min 2) 56min	3) 76.5min 4) 92min					
86.	<b>Among</b> $[Fe(CN)_6]^{-4}$ , $PO_4^{-3}$ , $SO_4^{-2}$ and $Cl^-$ , <b>w</b>	hich coagulates positive sol readily					
	1) $[Fe(CN)_6]^{-4}$ 2) $PO_4^{-3}$	3) $SO_4^{-2}$ 4) $Cl^-$					
<b>87.</b>	Identify the correct matching						
	1) Thermal Stability - $NH_3 < PH_3 < AsH_3 < SbH_3$						
	2) Reducing Power - $NH_3 > PH_3 > AsH_3 > SbH_3$						
	3) Basic Character - $NH_3 < PH_3 > A$	$asH_3 < SbH_3$					
	4) Volatile nature - $PH_2 > AsH_2 > 1$						

88.	<b>Identify acidic oxide</b>				
	1) $Cl_2O_7$	2) <i>CO</i> <sub>2</sub>	3) $N_2O_5$	4) All the above	
89.	/	which one has the high	. 2 3	,	
0,7	1) HOCl	2) $HClO_2$	3) <i>HClO</i> <sub>3</sub>	4) $HClO_4$	
90.	,	d number of lone pairs	, 3	, <del>,</del>	
<b>90.</b>	-	_		•	
	1) $sp^3d,3$	2) $sp^3d^2, 2$	3) $sp^3d^3,1$	4) $sp^3,1$	
		<u>BIOI</u>	<u>LOGY</u>		
91.	Which of the following	g taxonomic categories	includes all the other	categories?	
	1) order	2) kingdom	3) species	4) family	
92.	Muscles which regula	ite the diameter of pup			
	1) Ectodermal striated		2) Mesodermal striate		
	3) Ectodermal unstriate		4) Mesodermal unstri	ated	
93.		g is not related to Rocl			
	1) It is a Rhodophycea		2) It contains chl.a an		
0.4	3) Diplontic life cycle	•	4) Two unequal latera		
94.		sociated with nerve fib	<del>-</del>		
	A) Axolemma	B) Neurilemma	C) Endoneurium	D) Myelin sheath	
	E) Axoplasm	- C 41 1	] . 4 4 4 . 4		
	_	of the above from inside			
95.	1) E,A,B,D,C	2) E,A,D,C,B are correct related to T	3) E,A,C,B,D	4) E,A,D,B,C	
93.		ion takes place through	_		
	ii) It is free floating hy	-	Oliset		
	iii) Pulvinus petiole is	* •			
	iv) It drains $CO_2$ from		a) =		
0.4	1) All are correct	,	3) Two are correct	4) one is correct	
96.	Identify the correct st				
	•	crease rate of heart bea	_		
		ough the sympathetic n	erves can increase rat	e of heart beat and cardia	
	output		4 61 41		
		norepinephrine can inc		_	
		neural signals can incre		-	
07	1) A,B	2) B,C	3) C,D	4) A,D	
97.	Mismatch is	11 -	2) Cultananaida I	1:	
	1) Lycopsida – Selagir		2) Sphenopsida – <i>Lyc</i>	-	
00	3) Pteropsida – Adiant		4) Psilopsida– (psilott	um)	
98.		combinations from the	Iollowing		
	A) R.C.Dagar – Polyl				
	B) Kyoto protocol – I	Depletion of $O_3$			
	· ·	nservation of wildlife in			
	D) Burning of plastic	s – Polychlorinated bip	henyls		
	1) all the above	2) A,B	3) C,D	4) A,B,C	
99.	_	liospores produced in t	he following manner		
1) Endogenously, Endogenously 2) Exogenously, E			2) Exogenously, Endo	Endogenously	
	3) Endogenously, Exogenously 4) Exogenously, Exogenously				
100.	_	otors or inability to secu			
	1) Diabetes mellitus	2) Diabetes insipidus	3) Uremia	4) Renal Calculi	
101.	Pneumatophores are	present in			
	1) Rhizopus	2) Rhizobium	3) Vanda	4) Rhizophora	