## **Long Answer Questions (4-Marks)**

i) Define rubber. Write the chemical reactions for the preparation of following polymers.

a) teflon b) polyisoprene c) po	olyacrylonitrile d) SBR
ii) Explain the reactions involved in the preparat	ion of viscose rayon.
Ch	apter-16
Cite	apter 10
<b>Green Chemistry and Nanoch</b>	emistry Marks 3 with option 4
•	
Multiple choice	e questions (1 Mark)
i) Bottom ash of thermal power stations can be	used as raw material for cement and brick industry.
This example illustrates which of the following	ng principle of green chemistry
a) Atom economy.	b) Designing safer chemicals.
c) Design for energy efficiency.	d) Prevention of waste or by products.
ii) Less hazardous chemical synthesis point of vie	ew instead of harmful DDT Now a days is
used as insecticides	
a) Benzene	b) <b>ВНС</b>
c) Chlorobenzene	d) Ethanol
	and minimize hazardous effect on human health and
environment was coined by Paul T.Anastas	1) 1)
a) Green revolution	b) Blue revolution
c) Nano chemistry	d) Green Chemistry
iv) Nanorods are the example of	
	b) Two Dimensional new actum at was
a) One dimensional nanostructure	b) Two Dimensional nanostructure
c) Three dimensional nanostructure	d) Zero dimensional nanostructure
v) Which nanoparticles act as highly effective ba	cterial disinfectants, removing E.Coli from water?
a) Gold nanoparticles	b) Silver nanoparticles
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- c)  $TiO_2$  nanoparticles

  d) ZnO nanoparticles

  vi) Catalyst used for hydrogenation of oil is----
  a) $V_2O_5$ b) Fec)  $Raney\ Ni$ d)  $MnO_2$ vii) In green technology developed by Drath and Frost, adipic acid is enzymatically synthesized from

  a) Sucrose

  b) lactose

  c) maltose

  d) glucose
  - **Very short answer Questions (1 Mark)**
- i) Name the catalyst used to manufacture of H<sub>2</sub>SO<sub>4</sub> by contact process
- ii) Name the plant which is an example of self-cleaning
- iii) Write the name of a metal nanomaterial, which is used as a bacterial disinfectant.
- iv) Name the γ-isomer of Benzene hexachloride
- v) Which principle of green chemistry has its perspective towards to carrying out reactions at room temperature and pressure
- vi) Write any example of nanoparticles which are used in photo catalysis.
- vii) Write the name of the technique used to determine the geometry of nanomaterial.
- viii) Write the name of the nanostructured material that is used to increase the life of a car tyre.
- ix) Write the name of a nanomaterial synthesis process based on inorganic Polymerization.

## **Short Answer Questions (Type-I) (2-Marks)**

- 1) Define: Atom economy. Write the formula to calculate % atom economy
- 2) Define
- i) Green chemistry
- ii) Nanochemistry
- 3) Write one example of safer solvent and hazardous solvent
- 4) How does nanochemistry play an important role in water purification?

## **Short Answer Questions (Type-II) (3-Marks)**

- 1) Explain prevention of waste or by-products which is one of the principles of green chemistry.
- 2) Write any three advantages of nanoparticles and nanotechnology.

- 3) Write one example of nanomaterial used in following
  - i) water purification
  - ii) tyre of car
  - iii) ancient glass painting
- 4) Explain the role of green chemistry.
- 5) Explain any three characteristic features of nanoparticles.
- 6) State Disadvantages of nanoparticles and nanotechnology.
- 7) Define: a) Green chemistry b) Atom economy c) Sustainable development
- 8) Write three principles of green chemistry with examples.

## **Long Answer Questions (4-Marks)**

- 1) i) Explain the term sustainable development
  - ii) How is nanotechnology useful for the energy sector?
- 2) i) Write a short note on catalytic activity of nanoparticles.
  - ii) Complete and write the following table

S/N	Nanomaterial dimension	Nanomaterial type
a)	One dimension <100 nm	
b)	Two dimension <100 nm	

- 3) i) Explain use of safer solvent by giving suitable examples.
  - ii) Define a) Nanomaterial
    - b) Nanotechnology