SRIGAYATRI EDUCATIONAL INSTITUTIONS
INDIA

Max. Marks: $\mathbf{3 0 0}$ M

## SECTION - I

## (SINGLE CORRECT ANSWER TYPE)

This section contains 20 multiple choice questions. Each question has 4 options (A), (B), (C) and (D) for its answer, out of which ONLY ONE option can be correct.

Marking scheme: $\mathbf{+ 4}$ for correct answer, $\mathbf{0}$ if not attempted and $\mathbf{- 1}$ if not correct.

## MATHEMATICS

1. If $i=\sqrt{-1}$ then $\sum_{k=0}^{200} i^{k}+\prod_{p=1}^{50} i^{p}=x+i y \Rightarrow \frac{x}{y}=\ldots \ldots$. ( $\Pi$ is product symbol)
A) 1
B) -1
C) 0
D) $\frac{1}{2}$
2. Let $\bar{a}, \bar{b}, \bar{c}$ are three non-coplanar vectors such that $[\bar{a} \bar{b} \bar{c}]=6$. Then the volume of the tetrahedron with three coterminous edges along $\bar{a} \times \bar{b}, \bar{b} \times \bar{c}, \bar{c} \times \bar{a}$ is equal to (in cubic units)
A) 6
B) 36
C) 72
D) 216
3. Let $\mathbf{R}$ be a relation over set $N \times N$ defined by $(\mathbf{a}, \mathbf{b}) \mathbf{R}(\mathbf{c}, \mathbf{d})$ such that $a+d=b+c$ then $\mathbf{R}$ is ........ (Here $\mathbf{N}$ is the set of all natural numbers)
A) Reflexive only
B) Symmetric only
C) Transitive only
D) Equivalence relation
4. If $f: R \rightarrow R$ satisfies $f(x+y)=f(x)+f(y)$ for all $x, y \in R$ and $f(1)=7$, then $\sum_{r=1}^{n} f(r)$ is
A) $\frac{7 n}{2}$
B) $\frac{7(n+1)}{2}$
C) $7 n(n+1)$
D) $\frac{7 n(n+1)}{2}$
5. If $p, q$ are two propositions, then $(p \vee \sim q) \wedge(\sim p \wedge q)$ is
A) a tautology
B) a contradiction
C) neither a tautology nor a contradiction
D) both a tautology and a contradiction
6. The number of integral values of $\alpha$ for which the quadratic expression $\alpha x^{2}+|2 \alpha-3| x-6$ is positive for exactly two integral values of $x$ is equal to
A) 3
B) 2
C) 1
D) 0
7. The value of $\tan ^{-1} \frac{4}{7}+\tan ^{-1} \frac{4}{19}+\tan ^{-1} \frac{4}{39}+\tan ^{-1} \frac{4}{67}+\ldots . . \infty$ equals
A) $\tan ^{-1} 1+\tan ^{-1} \frac{1}{2}+\tan ^{-1} \frac{1}{3}$
B) $\tan ^{-1} 1+\cot ^{-1} 3$
C) $\cot ^{-1} 1+\cot ^{-1} \frac{1}{2}+\cot ^{-1} \frac{1}{3}$
D) $\cot ^{-1} 1+\tan ^{-1} 3$
8. The number of positive integral solutions of the equation $\left|\begin{array}{ccc}x^{3}+1 & x^{2} y & x^{2} z \\ x y^{2} & y^{3}+1 & y^{2} z \\ x z^{2} & y z^{2} & z^{3}+1\end{array}\right|=11$ is
A) 0
B) 3
C) 6
D) 12
9. Let $\vec{u}, \vec{v}, \vec{w}$ be such that $|\vec{u}|=1,|\vec{v}|=2,|\vec{w}|=3$. If the projection of $\vec{v}$ along $\vec{u}$ is equal to that of $\vec{w}$ along $\vec{u}$ and vectors $\vec{v}, \vec{w}$ are perpendicular to each other then $|\vec{u}-\vec{v}+\vec{w}|$ equals
A) 2
B) $\sqrt{7}$
C) $\sqrt{14}$
D) 14
10. Two cubes have their faces painted either red or blue. The first cube has five red faces and one blue face. When the two cubes are rolled simultaneously, the probability that the two top faces show the same colour is $\frac{1}{2}$. Number of red faces on the second cube, is
A) 1
B) 2
C) 3
D) 4
11. $P(x, y)$ is called a good point if $\mathbf{x}, \mathbf{y} \in \mathbf{N}$. Total number of good points lying inside the quadrilateral formed by the lines $2 x+y=2, x=0, y=0$ and $x+y=5$ is equal to
A) 4
B) 2
C) 10
D) 6
12. The point on the line $\frac{x-2}{1}=\frac{y+3}{-2}=\frac{z+5}{-2}$ at a distance of 6 from the point $(2,-3,-5)$ is
A) $(2,-5,-3)$
B) $(4,-7,-9)$
C) $(0,2,-1)$
D) $(-3,5,3)$
13. The value of $\lim _{x \rightarrow 0}\left\{\tan \left(\frac{\pi}{4}+x\right)\right\}^{1 / x}$ is
A) $e^{-1 / 2}$
B) $e^{2}$
C) $e^{1 / 2}$
D) 1
14. If $f(x+y)=f(x) f(y)$ for all $x, y \in R, f(5)=2, f^{1}(0)=3$. Then, $f^{1}(5)$ equals
A) 6
B) 3
C) 5
D) 7
15. If $f(x)=\min \{|x|,|x-2|, 2-|x-1|\}$, then $f(x)$ is
A) Discontinuous at exactly three points
B) Maximum value of $f(x)$ is 2
C) $f(x)$ is non-differentiable at $x=\frac{1}{2}, 0,1 \frac{1}{2}, 2, \frac{5}{2}$
D) $f(x)$ is non-differentiable at $x=-\frac{1}{2}, 0,1,2, \frac{5}{2}$
16. If $\int \frac{d x}{x^{2}\left(x^{n}+1\right)^{(n-1) / n}}=-[f(x)]^{1 / n}+C$ then $f(x)$ is
A) $\left(1+x^{n}\right)$
B) $1+x^{-n}$
C) $x^{n}+x^{-n}$
D) $x^{n}-x^{-n}$
17. If $[\mathrm{x}]$ denotes the greatest integer less than or equal to $x$, then the value of the integral $\int_{0}^{2} x^{2}[x] d x$ equals
A) $\frac{5}{3}$
B) $\frac{7}{3}$
C) $\frac{8}{3}$
D) $\frac{4}{3}$
18. The area inside the parabola $5 x^{2}-y=0$ but outside the parabola $2 x^{2}-y+9=0$ is
A) $12 \sqrt{3}$
B) $6 \sqrt{3}$
C) $8 \sqrt{3}$
D) $4 \sqrt{3}$
19. The solution of differential equation $\frac{d y}{d x}+x \sin 2 y=x^{3} \cos ^{2} y$ is
A) $\cot y=\frac{1}{2}\left(x^{2}+1\right)+c e^{-x^{2}}$
B) $\tan y=\frac{1}{3}\left(x^{2}+2\right)+c e^{-2 x^{2}}$
C) $\tan y=\frac{1}{2}\left(x^{2}-1\right)+c e^{-x^{2}}$
D) $\tan y=\frac{1}{3}\left(x^{2}-2\right)+c e^{-2 x^{2}}$
20. The solution of differential equation $y d x+\left(x+x^{2} y\right) d y=0$ is
A) $-\frac{1}{x y}+\log x=c$
B) $-\frac{x}{y}+\log y=c$
C) $-\frac{y}{x}+\log x y=c$
D) $-\frac{1}{x y}+\log y=c$

## SECTION-II

## (Numerical Value Answer Type)

This section contains 5 questions. The answer to each question is a Numerical values comprising of positive or negative decimal numbers. Marking scheme: $\mathbf{+ 4}$ for correct answer, 0 in all other cases.
21. Given an acute triangle ABC such that $\sin C=\frac{4}{5}$, then $\tan A=\frac{24}{7}$ and $c=50$. The area of the triangle ABC
22. If $1, \alpha_{1}, \alpha_{2}, \ldots . . \alpha_{2008}$ are $(2009)^{\text {th }}$ roots of unity, then the value of $\sum_{r=1}^{2008} r\left(\alpha_{r}+\alpha_{2009-r}\right)$ equals
23. The area of the triangle formed by two rays whose combined equation is $y=|x|$ and the line $x+2 y=2$ is
24. The area of the triangle formed by the tangents from the point $(4,3)$ to the circle $x^{2}+y^{2}=9$ and the line joining their points of contact is
25. The ratio of the area of a triangle inscribed in the ellipse $\frac{x^{2}}{a^{2}}+\frac{y^{2}}{b^{2}}=1$ and that of a triangle formed by the corresponding points on the auxiliary circle is 0.5 , then the eccentricity of the ellipse is

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## PHYSICS

26. A lens has a convex surface of radius of radius 20 cm and a concave surface of radius 40 cm and is made of glass of refractive index 1.54. Compute the focal length of the lens
A) 74 cm
B) 23 cm
C) 15 cm
D) 29 cm
27. Red light falls normally on a diffraction grating ruled 4000 lines / cm and the second order image is diffracted $34.0^{\circ}$ from the normal. Compute the wavelength of the light $\left(\sin 34^{\circ}=0.559\right)$
A) 229 nm
B) 137 nm
C) 699 nm
D) 250 nm
28. Given below are symbols for some logic gates


The XOR gate and NOR gate respectively are
A) 1 and 2
B) 2 and 3
C) 3 and 4
D) 1 and 4
29. A body slides down a plane inclined at an angle $\theta$ to the horizontal. The coefficient of friction $\mu$ down the plane varies in direct proportion to the distance moved down the plane. The body will move down the plane with a
A) Constant acceleration $g \sin \theta$
B) Constant acceleration $(g \sin \theta-\mu g \cos \theta)$
C) Constant acceleration $(\mu g \cos \theta-g \sin \theta)$
D) Variable acceleration that first decreases, then becomes negative
30. Unit of LCR is
A) ohm $\times$ sec
B) $\mathrm{ohms}^{-1}$
C) $\mathrm{ohm} \times \mathrm{sec}^{2}$
D) No unit
31. Two factories are sounding their sirens at 800 Hz each. A man goes from one factory to the other at a speed of $2 \mathrm{~ms}^{-1}$. The speed of sound is $320 \mathrm{~ms}^{-1}$. Therefore, the number of beats heard by the person in one second will be
A) 2
B) 4
C) 8
D) 10
32. A container open to atmosphere contains air (assumed to be an ideal gas) at temperature $27^{\circ} \mathrm{C}$. The temperature is now raised to $227^{\circ} \mathrm{C}$. The ratio of number of atoms in the container now and at the beginning is
A) $\frac{3}{5}$
B) $\frac{5}{3}$
C) $\frac{3}{4}$
D) $\frac{4}{5}$
33. Four charges $Q$ each are located at four vertices of a regular tetrahedron of side $L$. The potential energy of the system is
A) $\frac{4 Q^{2}}{\pi \varepsilon_{0} L^{2}}$
B) $\frac{6 Q^{2}}{4 \pi \varepsilon_{0} L}$
C) $\frac{Q^{2}}{\pi \varepsilon_{0} L}$
D) $\frac{8 Q^{2}}{4 \pi \varepsilon_{0} L}$
34. The moment of inertia of a body about a given axis is $1.2 \mathrm{~kg}-\mathrm{m}^{2}$. Initially, the body is at rest. In order to produce a rotational kinetic energy of 1500 J , an angular acceleration of $25 \mathrm{rad}-\mathrm{s}^{-2}$ must be applied about that axis for a duration of
A) 4 s
B) 2 s
C) 8 s
D) 10 s
35. Light passes from a denser medium 1 to a rarer medium 2 . When the angle of incidence is $\theta$, the reflected and refracted rays are mutually perpendicular. The critical angle will be
A) $\sin ^{-1}(\cot \theta)$
B) $\sin ^{-1}(\tan \theta)$
C) $\sin ^{-1}(\cos \theta)$
D) $\sin ^{-1}(\sec \theta)$
36. A particle $A$ has charge $+\mathbf{q}$ and particle $B$ has charge $+4 q$ with each of them having same mass $m$. When allowed to fall from rest through the same electric potential difference, the ratio of their speeds $\frac{v_{A}}{v_{B}}$ will become
A) $2: 1$
B) $1: 2$
C) $1: 4$
D) $4: 1$
37. A radio transmitter operates at 880 kHz and its power is 10 kW . The number of photons emitted per second is
A) $1.72 \times 10^{31}$
B) $1.72 \times 10^{32}$
C) $3.44 \times 10^{31}$
D) None of these
38. The whistle of a railway engine is heard in winter at much longer distances. This is due to
A) Decrease in the velocity of sound in winter
B) Decrease in the density of air with respect to height from the surface of the earth
C) Cold air absorbs much small energy form sound waves
D) Increase in the density of air with respect to height from the surface of the earth
39. The position vector of an electron is $r=5 i+4 j-3 k$. To an observer moving along $x$ direction with speed 0.6 c , the magnitude of position vector is
A) $5 \sqrt{2}$
B) $\sqrt{34}$
C) $\sqrt{41}$
D) 5
40. The displacement of a particle is given by $y=2 \sin (\omega t)+2 \sin \left(\omega t+\frac{\pi}{3}\right)$ then, the incorrect statement is
A) The amplitude of motion is $\sqrt{12}$
B) The angular frequency is $\omega$
C) The velocity at $t=0$ is $3 \omega$
D) The initial phase of motion is $\frac{\pi}{3}$
41. Substances for which permeability $\mu$ is slightly greater than 1 , are called
A) Diamagnetic
B) Paramagnetic
C) Ferromagnetic
D) Non-magnetic
42. An isolated sphere $S$ of radius $R$ carries an electric charge. $S$ is momentarily connected to a distant uncharged sphere $T$ which has a radius $r$. The ratio of surface charge density of $S$ to that of $T$ is
A) $\frac{R}{r}$
B) $\frac{r}{R}$
C) $\frac{r}{2 R}$
D) $\frac{(r+R)}{R}$
43. In post office box, the graph of galvanometer deflection versus resistance $R$ (pulled out of resistance box) for the ratio $100: 1$ is given as shown (due to unsuitable values of $R$, galvanometer shows deflection). The two consecutive values of $\mathbf{R}$ are shown in the figure. The value of unknown resistance would be

A) $3.2 \Omega$
B) $3.24 \Omega$
C) $3.206 \Omega$
D) $3.237 \Omega$
44. The forward bias characteristics of two diodes $D_{1}$ and $D_{2}$ are shown, the knee voltages for $D_{1}$ and $D_{2}$ are respectively (approx.)

A) 0.4 V and 0.7 V
B) 0.6 V and 0.9 V
C) 0.6 V and 0.8 V
D) 0.4 V and 0.9 V
45. The output current of a $60 \%$ modulated AM generator is 1.5 A . To what value will the current rise, if the generator is additionally modulated by another audio wave of modulation index 0.7?
A) 0.64 A
B) 1.64 A
C) 2.34 A
D) 5.32 A

## SECTION- II

## (Numerical Value Answer Type)

This section contains 5 questions. The answer to each question is a Numerical values comprising of positive or negative decimal numbers. Marking scheme: +4 for correct answer, 0 in all other cases.
46. A quarter cylinder of radius $R$ and refractive index 1.5 is placed on a table. A point object $P$ is kept at a distance $m R$ from it. Find the value of $m$ for which a ray from $P$ will emerge parallel to the table as shown in figure.

47. A flat thin circular disc has a radius 4 cm and a circular hole of radius $\frac{1}{2} \mathrm{~cm}$ is made in it with its centre at a distance of 1 cm from the centre of disc. The mass of the disc is 10 kg . If the moment of inertia of the system about an axis passing through the centre of the hole is $N \times 10^{-3} \mathrm{~kg} \mathrm{~m}^{2}$, Find the value of $N$.
48. A small sphere of mass 2.0 g and having charge 0.5 mC is suspended by a string between the plates of a parallel plate capacitor as shown in the figure. What potential difference (in volt) between the plates (separation 20 cm ) should be applied so that the string makes an angle of $45^{0}$ with the vertical? (Take $g=10 \mathrm{~ms}^{-2}$ )

49. A ball is allowed to fall freely from a height of $\mathbf{3}$ metres on to a fixed plate. The successive rebound heights are $h_{1}, h_{2}, h_{3}, \ldots$. If the distance covered by the ball before coming to rest is $x$ metres, find the value of $x$. (Given that the coefficient of restitution is 0.5 )
50. Find the resistance (in $\Omega$ ) between the terminals $A$ and $B$ of the network shown below


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## CHEMISTRY

51. Which of the following has the maximum number of atoms?
A) 24 g C
B) 56 g of Fe
C) 27 g of Al
D) 108 g of Ag
52. Which of the following will not be oxidized by $O_{3}$ ?
A) $K I$
B) $\mathrm{FeSO}_{4}$
C) $\mathrm{KMnO}_{4}$
D) $\mathrm{K}_{2} \mathrm{MnO}_{4}$
53. The decreasing order of energy for the electrons represented by the following sets of quantum numbers is
1) $n=4, l=0, m=0, s= \pm 1 / 2$
2) $n=3, l=1, m=1, s=-1 / 2$
3) $n=3, l=2, m=0, s=+1 / 2$
4) $n=3, l=0, m=0, s=-1 / 2$
A) $1>2>3>4$
B) $2>1>3>4$
C) $3>1>2>4$
D) $4>3>2>1$
54. The $\Delta_{f} H^{\ominus}$ for $\mathrm{CO}_{2}(\mathrm{~g}), \mathrm{CO}(\mathrm{g})$, and $\mathrm{H}_{2} \mathrm{O}(\mathrm{g})$ are $-393.5,-110.5$, and $-214.8 \mathrm{KJmol}^{-1}$, respectively. The standard enthalpy change(in $\mathrm{KJmol}^{-1}$ ) for the reaction
$\mathrm{CO}_{2}(g)+\mathrm{H}_{2}(g) \rightarrow \mathrm{CO}(g)+\mathrm{H}_{2} \mathrm{O}(g)$ is
A) 524.1
B) +41.2
C) -262.5
D) -41.2
55. $C s C l$ crystallizes in body centred cubic lattice. If ' $\mathbf{a}$ ' is its edge length then which of the following expressions is correct?
A) $r_{C s^{\oplus}}+r_{C l^{\ominus}}=\frac{\sqrt{3}}{2} a$
B) $r_{C s^{\oplus}}+r_{C l^{\ominus}}=\sqrt{3} a$
C) $r_{C s^{\oplus}}+r_{C l^{\ominus}}=3 a$
D) $r_{C s^{\oplus}}+r_{C l^{\ominus}}=\frac{3}{2} a$
56. Silver is removed electrolytically from 200 mL of a 0.1 N solution of $\mathrm{AgNO}_{3}$ by a current of 0.1A. How long will it take to remove half of the silver from the solution?
A) 0.1 s
B) 100 s
C) 965 s
D) 9650 s
57. What is the nature of $A l_{2} O_{3}$ and $B_{2} O_{3}$ ?
A) Acidic, acidic
B) Acidic, amphoteric
C) Amphoteric, amphoteric
D) Amphoteric, acidic
58. In terms of polar character, the correct order is
A) $\mathrm{H}_{2} \mathrm{~S}>\mathrm{HF}>\mathrm{H}_{2} \mathrm{O}>\mathrm{NH}_{3}$
B) $\mathrm{HF}>\mathrm{H}_{2} \mathrm{O}>\mathrm{NH}_{3}>\mathrm{H}_{2} \mathrm{~S}$
C) $\mathrm{HF}>\mathrm{H}_{2} \mathrm{~S}>\mathrm{NH}_{3}>\mathrm{H}_{2} \mathrm{O}$
D) $\mathrm{H}_{2} \mathrm{~S}>\mathrm{NH}_{3}>\mathrm{H}_{2} \mathrm{O}>\mathrm{HF}$
59. The correct order of stability for the following superoxides is
A) $\mathrm{KO}_{2}>\mathrm{RbO}_{2}>\mathrm{CsO}_{2}$
B) $\mathrm{RbO}_{2}>\mathrm{CsO}_{2}>\mathrm{KO}_{2}$
C) $\mathrm{CsO}_{2}>\mathrm{RbO}_{2}>\mathrm{KO}_{2}$
D) $\mathrm{KO}_{2}>\mathrm{CsO}_{2}>\mathrm{RbO}_{2}$
60. $\mathrm{H}_{3} \mathrm{BO}_{3}$ is
A) a monobasic acid and weak Lewis acid
B) a monobasic acid and weak Bronsted acid
C) a monobasic acid and strong Lewis acid
D) a tribasic and weak Bronsted acid
61. Acid strength is in the order
A) $\mathrm{HClO}_{4}>\mathrm{HIO}_{4}>\mathrm{HBrO}_{4}$
B) $\mathrm{HClO}_{4}>\mathrm{HBrO}_{4}>\mathrm{HIO}_{4}$
C) $\mathrm{HClO}_{4}<\mathrm{HBrO}_{4}>\mathrm{HIO}_{4}$
D) None
62. Both $\left[\mathrm{Ni}(\mathrm{CO})_{4}\right]$ and $\left[\mathrm{Ni}(\mathrm{CN})_{4}\right]^{2-}$ are diamagnetic. The hybridization of nickel in these complexes, respectively, are
A) $\mathrm{sp}^{3}, \mathrm{sp}^{3}$
B) $\mathrm{sp}^{3}, \mathrm{dsp}^{2}$
C) $\mathrm{dsp}^{2}, \mathrm{sp}^{3}$
D) $\mathrm{dsp}^{2}, \mathrm{dsp}^{2}$
63. Ellingham diagram represents
A) Change of $\Delta G$ with temperature
B) Change of $\Delta H$ with temperature
C) Change of $\Delta G$ with pressure
D) Change of $(\Delta G-T \Delta S)$ with temperature
64. The IUPAC name of the following compound is

A) 4-Bromo-3-cynophenol
B) 2-Bromo-5-hydroxy benzene carbonitrile
C) 2-Cyno-4-hydroxy bromo benzene
D) 6-Bromo-3-hydroxy benzonitrile
65. The E-isomer among the following is
А)

B)

C)

D)

66. Among the following, the least stable resonance structure is
A)

B)

c)

D)

67. 


A)

B)

C)

D)

68. ( $X) \xrightarrow[\mathrm{H}_{2} \mathrm{C}_{2} \mathrm{SO}_{4}]{\mathrm{K}_{7}}(Y) \xrightarrow{\mathrm{LiAlH}_{4}}(X)$; $\mathbf{X}, \mathbf{Y}$ are
A) $\mathrm{CH}_{3} \mathrm{COOH}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$
B) $\mathrm{CH}_{3} \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}, \mathrm{CH}_{3} \mathrm{COOH}$
C) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}, \mathrm{CH}_{3} \mathrm{COOH}$
D) $\mathrm{CH}_{3} \mathrm{CHO}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$
69. Acetone $\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}\right) \xrightarrow{\text { Aldol condensation }}(\mathrm{X}) \mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O} \xrightarrow[\mathrm{NaOH}+\mathrm{Cl}_{2}]{\mathrm{NaOCl}}(Y) \mathrm{C}_{5} \mathrm{H}_{7} \mathrm{ONa}$; Y is:

## $\mathrm{CH}_{3}$.

A) $\mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{COONa}$
B)

c)

D)

70. Which of the following is the most basic
A)

B)

C)

D)


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71. $\mathbf{1 0} \mathbf{~ m L}$ of a solution of $\mathrm{H}_{2} \mathrm{O}_{2}$ of $\mathbf{1 0}$ volume strength decolourises 100 mL of $\mathrm{KMnO}_{4}$ solution acidified with dil. $\mathrm{H}_{2} \mathrm{SO}_{4}$. The amount of $\mathrm{KMnO}_{4}$ in the given solution is ( $\mathbf{K}=\mathbf{3 9}, \mathbf{M n = 5 5}$ )
72. A certain buffer solution contains equal concentration of $X^{-}$and $\mathbf{H X}$. The $K_{b}$ for $X^{-}$is $\mathbf{1 0}^{\mathbf{- 1 0}}$. The pH of the buffer is
73. Among $\mathrm{PbS}, \mathrm{CuS}, \mathrm{HgS}, \mathrm{MnS}, \mathrm{Ag}_{2} \mathrm{~S}, \mathrm{NiS}, \mathrm{CoS}, \mathrm{Bi}_{2} \mathrm{~S}_{3}$ and $\mathrm{SnS}_{2}$, the total number of BLACK coloured sulphides is:
74. An organic compound contains $66 \% \mathrm{C}$ and $13.3 \% \mathrm{H}$. Its vapour density is 37 . The possible number of isomers of all types for the compound is:
75. How many chirality centers are there in an aldohexose?

