SRIGAYATRI EDUCATIONAL INSTITUTIONS

INDIA

Time: 3 Hours

JEE MAIN TOT GT-4

Max. Marks: 300 M

<u>SECTION – I</u>

(SINGLE CORRECT ANSWER TYPE)

This section contains 20 multiple choice questions. Each question has 4 options (A), (B), (C) and (D) for its answer, out of which **ONLY ONE** option can be correct. **Marking scheme:** +4 for correct answer, 0 if not attempted and -1 if not correct.

MATHEMATICS

SYLLABUS:

Let $a, b, c \in R$. If $f(x) = ax^2 + bx + c$ is such that 1. a+b+c=3 and $f(x+y)=f(x)+f(y)+xy \forall x, y \in \mathbb{R}$. Then $\sum_{n=1}^{10} f(n)$ is equal to B) 255 D) 165 A) 190 B) 255 C) 330 If $\overline{a}, \overline{b}, \overline{c}$ are unit vectors such that $\overline{a} + \overline{b} + \overline{c} = \overline{0}$ then $\overline{a}, \overline{b} + \overline{b}, \overline{c} + \overline{c}, \overline{a} =$ A) 190 2. B) $\frac{-3}{2}$ C) $\frac{1}{2}$ D) $-\frac{1}{2}$ A) $\frac{3}{2}$ If $81^{\sin^2 x} + 81^{\cos^2 x} = 30$ and $0 \le x \le \frac{\pi}{2}$ then x =3. B) $\frac{\pi}{4}, \frac{\pi}{2}$ C) $\frac{3\pi}{4}, \frac{2\pi}{4}$ D) $\frac{\pi}{2}, \frac{4\pi}{2}$ A) $\frac{\pi}{\epsilon}, \frac{\pi}{2}$ If the angles of elevation of the top of a tower from three collinear points A,B and C on a line 4. leading to the foot of the tower are $30^{\circ},45^{\circ}$ and 60° respectively, then the ratio AB : BC is B) $\sqrt{3}:\sqrt{2}$ C) $1:\sqrt{3}$ A) $\sqrt{3}:1$ D) 2:3 $\sqrt{-1 - \sqrt{-1 - \sqrt{-1 \dots \infty}}} =$ 5. A) 1 B) -1 C) ω D) $-\omega^2$ A ratio of the 5th term from the beginning to the 5th term from the end in the binomial 6. expansion of $\left(2^{\frac{1}{3}} + \frac{1}{2(3)^{\frac{1}{3}}}\right)^{10}$ is B) $2(36)^{\frac{1}{3}}:1$ C) $1:2(6)^{\frac{1}{3}}$ D) $4(36)^{\frac{1}{3}}$:1 A) 1:4(16) $\frac{1}{3}$ The H.M of the numbers $\frac{1}{5}, \frac{1}{10}, \frac{1}{15}, \frac{1}{20}, \frac{1}{25}, \frac{1}{30}, \frac{1}{35}$ is 7. A) $\frac{1}{20}$ B) $\frac{1}{16}$ D) $\frac{1}{13}$ C) $\frac{1}{15}$ A point is selected at random from the interior of a circle. The probability that the point is 8. closer to the center than to the boundary of the circle is C) $\frac{1}{4}$ D) $\frac{1}{3}$ A) $\frac{3}{4}$ B) $\frac{1}{2}$

9. The value of $\sum_{r=16}^{30} (r+2)(r-3)$ is equal to A) 7785 B) 7780 C) 7775 D) 7770

The negation of $\neg s \lor (\neg r \land s)$ is equivalent to : 10. B) $s \wedge (r \wedge \sim s)$ C) $s \vee (r \vee \sim s)$ A) $s \wedge \sim r$ D) $s \wedge r$ let $\int \frac{x^2 - 1}{x^3 \sqrt{2x^4 + 2x^2} - 1} dx = f(x) + C$ where f(1) = -1 and C is constant of integration then 11. $\lim_{x\to\infty} f(x)$ is equal to A) $\frac{2}{\sqrt{3}}$ B) $-\frac{2}{\sqrt{3}}$ C) $\frac{\sqrt{3}}{2}$ D) $-\frac{\sqrt{3}}{2}$ $\int_{0}^{7\frac{1}{2}} [x-1] dx =$ where [x] denotes the greatest integer less than or equal to x 12. C) $\frac{5}{4}$ A) $\frac{5}{2}$ **B**) 1 D) 17 13. If $e^y + xy = e$, then ordered pair $\left(\frac{dy}{dx}, \frac{d^2y}{dx^2}\right)$ at x=0 is equal to B) $\left(-\frac{1}{e}, \frac{1}{e^2}\right)$ C) $\left(\frac{1}{e}, \frac{1}{e^2}\right)$ D) $\left(-\frac{1}{e}, -\frac{1}{e^2}\right)$ A) $\frac{1}{-1}, -\frac{1}{-2}$ The solution of the differential equation $x \frac{dy}{dx} + 2y = x^2 (x \neq 0)$ with y(1) = 1 is 14. B) $y = \frac{x^3}{5} + \frac{1}{5x^2}$ C) $y = \frac{x^2}{4} + \frac{3}{4x^2}$ D) $y = \frac{3}{4}x^2 + \frac{1}{4}x^2$ A) $y = \frac{4}{5}x^3 + \frac{1}{5x^2}$ An apache helicopter of enemy is flying along the curve given by $y = x^2 + 7$. A soldier, placed 15. at (3, 7) wants to shoot down the helicopter when it is nearest to him then the nearest distance is A) $\sqrt{2}$ B) $\sqrt{5}$ C) $\sqrt{7}$ D) $\sqrt{11}$ The equation of the line passing through (-4, 3, 1) parallel to the plane x+2y-z=5 and 16. intersecting the line $\frac{x+1}{-3} = \frac{y-3}{2} = \frac{z-2}{-1}$ B) $\frac{x+4}{1} = \frac{y-3}{1} = \frac{z-1}{2}$ A) $\frac{x-4}{2} = \frac{y+3}{1} = \frac{z+3}{4}$ C) $\frac{x+4}{3} = \frac{y-3}{-1} = \frac{z-1}{1}$ D) $\frac{x+4}{-1} = \frac{y-3}{1} = \frac{z-1}{1}$ 17. Two vertices of a triangle are (5,-1) and (-2, 3) if the orthocenter is at the origin, then its third vertex is B) (4,-7) C) (-4, 7) A) (4, 7) The point diametrically opposite to the P (1, 0) on the circle $x^2 + y^2 + 2x + 4y - 3 = 0$ is 18. B) (-3, 4) C) (-3,-4) D) (3.4) A) (3.-4) If $f(x) = \begin{cases} ax^2 + b & \text{if } 0 \le x < 1 \\ 4 & \text{if } x = 1 \\ x+3 & \text{if } x > 1 \end{cases}$ then the value of (a, b) for which f(x) is cannot be continuous at 19. x=1 is A) (2, 2) B) (3.1) C) (4, 0) D) (5.2) If the angle between the line $x = \frac{y-1}{2} = \frac{z-3}{\lambda}$ and the plane x + 2y + 3z = 4, is $Cos^{-1}\left(\sqrt{\frac{5}{14}}\right)$ 20. A) $\frac{2}{5}$ B) $\frac{5}{2}$ C) $\frac{2}{2}$ D) $\frac{3}{2}$

(Numerical Value Answer Type)

This section contains 5 questions. The answer to each question is a Numerical values comprising of positive or negative decimal numbers. Marking scheme: +4 for correct answer, 0 in all other cases.

If a, b, c are non zero complex number satisfying $a^2 + b^2 + c^2 = 0$ and 21.

 $b^2 + c^2$ ab $\begin{vmatrix} b + c & ab & ac \\ ab & c^2 + a^2 & bc \\ ac & bc & a^2 + b^2 \end{vmatrix} = Ka^2b^2c^2, \text{ then K is equal to}$ If $\left| Z + \frac{6}{Z} \right| = 5$ then the greatest value of |Z| =

- 22.
- If $n_{c_{r+1}} = 330$, $n_{c_r} = 462$, $n_{C_{r+1}} = 462$ then $\mathbf{r} = _$ 23.
- If the area enclosed by the curves $y^2 = 4\lambda x$ and $y = \lambda x$ is $\frac{1}{9}$ square units then value of λ is 24. equal to
- If the curves $y = x^2 + px + q$ and $y = rx x^2$ teach each other at (1,0) then the value of 25. $p^2 + q^2 + r^2$ equals

<u>SECTIO</u>N – I (SINGLE CORRECT ANSWER TYPE)

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PHYSICS

SYLLABUS:

- The resistance of a metallic wire is R = V/I. In a circuit the potential difference across the 26. resistance is given by $V = (8 \pm 0.5) V$ and current in the circuit is given by $I = (4 \pm 0.2) A$, then the value of resistance with its percentage error. A) $(4 \pm 2\%) \Omega$ B) $(2 \pm 11.25\%) \Omega$ C) $(9 \pm 2\%) \Omega$ D) $(12 \pm 15\%) \Omega$
- 27. A ball is dropped from the top of a building. The ball takes 0.5s to fall past the 3m length of window some distance from the top of the building. If the speed of the ball at the top and at the bottom of the window are V_1 and V_2 then $(g = 9.8 \text{m/s}^{2})$

A)
$$V_1 + V_2 = 12ms^{-1}$$
 B) $V_1 + V_2 = 4.9ms^{-1}$ C) $V_1 + V_2 = 1ms^{-1}$ D) $\frac{V_2}{V_1} = 2ms^{-1}$

A uniform rod of length 30cm and mass 3kg is arranged as shown in the figure . The strings 28. are pulled with forces 20N and 32N. Find the force exerted by 20cm part of the rod on 10cm part



- A) 4N D) 32N 29. A force acts on a 3g body. The position of the body as a function of time 't' is given by $x = t^3 - 4t^2 + 3t$. The work done in first four seconds is A) 578mJ B) 528mJ C) 498mJ D) 458mJ
- 'n' moles of an ideal gas is taken through the process 'ABC'. The total work done 30.



A) 4nRT₀ln2
B) 2nRT₀ln2
C) -2nRT₀ln2
D) Zero
31. The length of the wire is 'l' when the tension in it is 'F' and 'x l' when the tension in it is 'yF'. The natural length of the wire is

900

A)
$$\frac{(x-y)l}{x-1}$$
 B) $\frac{(y-x)l}{y-1}$ C) $\frac{(x-y)l}{x+1}$ D) $\frac{(y-x)l}{y+1}$

Jh

32. Three rods one arranged as shown in the figure the temperature at the junction is

Oc

B) $60^{\circ}C$ A) $45^{\circ}C$ C) 30° C D) 20° C 33. A large tank filled with water to a height 'h' is to be emptied through a small hole at the bottom. The ratio of times taken for the level of water to fall from h to $\frac{h}{2}$ and $\frac{h}{2}$ to zero. B) $\frac{1}{\sqrt{2}}$ D) $\frac{1}{\sqrt{2}-1}$ C) $\sqrt{2} - 1$ A) $\sqrt{2}$ A pendulum of length l = 1 m is released from $\theta = 60^{\circ}$. The rate of change of speed of the bob at 34. $\theta = 30^{\circ} (g = 10 \text{ms}^{-B)}$ A) $5\sqrt{3}ms^{-2}$ C) 10ms⁻² B) $5ms^{-2}$ D) 2.5ms^{-2} The x and y- co-ordinates of a particle are $x = A \sin \omega t$ and $y = A \sin \left(\omega t + \frac{\pi}{2} \right)$. The motion of 35. the particle is A) Circular anti clockwise B) Circular clockwise C) Elliptical clockwise D) Rectilinear from left to right The lowest frequency standing wave that can be setup in a string has a frequency of 512Hz. 36. The length of the string is 80cm. The speed of the wave in the string is B) 410ms⁻¹ D) 330ms⁻¹ A) 820ms⁻¹ C) 1000ms^{-1} A plane mirror is moving with velocity $4\hat{i}+5\hat{j}+8\hat{k}$. A point object in front of the mirror 37. moves with a velocity $3\hat{i} + 4\hat{j} + 5\hat{k}$. Here, \hat{k} is the normal to the plane of the mirror facing towards the object. The velocity of the image is B) $3\hat{i} + 4\hat{j} + 11\hat{k}$ C) $-3\hat{i} - 4\hat{j} + 11\hat{k}$ D) $7\hat{i} + 9\hat{j} + 11\hat{k}$ A) $-3\hat{i}-4\hat{i}+5\hat{k}$ **YDSE** experiment is carried out by green, red and blue light. The fringe width are β_G , β_R , β_R 38. then B) $\beta_B > \beta_R > \beta_G$ C) $\beta_R > \beta_B > \beta_G$ D) $\beta_R > \beta_G > \beta_B$ A) $\beta_{C} > \beta_{R} > \beta_{R}$

39. Find the current through the branch BD



D) 4A

D) 6V

40. Find potential difference across 4.5µF



A) 8V

A) 5A

- 41. A rectangular ferro magnetic material of dimensions 10cm × 0.5cm × 2cm is placed in a magnetic field of intensity 2×10⁵ A/m. The induced magnetic moment is 6A-m². The value of magnetic induction is

 A) 1T
 B) 2T
 C) 3T
 D) 4T
- 42. The electric field in a region $\vec{E} = \hat{5i} + 2\hat{j}$ N/C. The electric flux due to this field through an area of 2m² lying is YZ plane is _____(SI Umt) A) 10 B) 20 C) $10\sqrt{2}$ D) $2\sqrt{29}$
- 43. A conducting rod of length l = 1m is moving with a velocity V = 4m/s making an angle 30^{0} with its length in a magnetic field B = 2T exists perpendicular to plane of motion, then potential difference across the ends of the rod

A) V_A-V_B=8V
B) V_A-V_B=4V
C) V_B-V_A=8V
D) V_B-V_A=4V
44. V-I characteristic curve of a diode is given. Find the ratio of forward to reverse bias resistance

 $\begin{array}{c} 1 & \text{C} \\ 1 &$

D) 10⁻⁶

45. In a photoelectric experiment a graph is drawn between stopping potential and frequency of incident radiation. Find the threshold wavelength of photo sensitive metal



A) 100

SECTION-II

(Numerical Value Answer Type)

This section contains 5 questions. The answer to each question is a Numerical values comprising of positive or negative decimal numbers. Marking scheme: +4 for correct answer, 0 in all other cases.

- 46. The power factor of a circuit is $\frac{1}{\sqrt{2}}$. The capacitance of the circuit is equal to_____
- 47. On interchanging the resistance in left and right gaps in a meter bridge, balancing point shifts 60cm towards left. If the sum of the two resistors in 25Ω what is the left gap resistance _____
- 48. Nitrogen of mass 10g is enclosed in a vessel at 300K. What heat must be supplied to it to double the rms velocity of its molecules. _____
- 49. A particle of mass 3kg is moving under the action of a central force whose potential energy is given by $U = 10 r^3 J$. For what angular momentum will the orbit be a circle of radius 10 m
- 50. A particle is moving in X -Y plane. At a certain instant the components of its velocity and acceleration are $V_x = 3ms^{-1}$, $V_y = 4ms^{-1}$, $a_x = 2ms^{-2}$, $a_y = 1ms^{-2}$. The rate of change of speed at this moment is _____

<u>SECTION – I</u>

(SINGLE CORRECT ANSWER TYPE)

This section contains 20 multiple choice questions. Each question has 4 options (A), (B), (C) and (D) for its answer, out of which **ONLY ONE** option can be correct. **Marking scheme:** +4 for correct answer, 0 if not attempted and -1 if not correct.

CHEMISTRY

SYLLABUS:

51. The kinetic energy of an electron in the second Bohr orbit of a hydrogen atom is $[a_0$ is Bohr radius]:

A)
$$\frac{h^2}{4\pi^2 m a_0^2}$$
 B) $\frac{h^2}{16\pi^2 m a_0^2}$ C) $\frac{h^2}{32\pi^2 m a_0^2}$ D) $\frac{h^2}{64\pi^2 m a_0^2}$
52. Among Al₂O₃, SiO₂, P₂O₃ and SO₂ the correct order of acid strength is
A) $Al_2O_3 < SiO_2 < SO_2 < P_2O_3$ B) $SiO_2 < SO_2 < Al_2O_3 < P_2O_3$
C) $SO_2 < P_2O_3 < SiO_2 < Al_2O_3$ D) $Al_2O_3 < SiO_2 < P_2O_3 < SO_2$
53. The molecule which has zero dipole moment is :
A) CH₂Cl₂ B) BF₃ C) NF₃ D) ClO₂
54. Which of the following volume (V) – temperature (T) plots represents the behavior of one mole of an ideal gas at one atmospheric pressure ?
 $V_{(224L)}^{V(L)} = (28.6L, 373K) = (22.4L, 273K) = (22.4L, 273K) = (22.4L, 273K) = (22.4L, 273K) = (14.2L, 373K) = (14.2L, 32.2L) = (14.2L, 32.2L) = (14.2L, 32.2L) = (14.2L, 32.2L) = (14.2L) = (14.2L, 32.2L) =$

A) +60e.u B) +100e.u C) -60e.u D) -100e.u

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57.	Which one is more acidic	in aqueous solution.		
	A) NiCl ₂	B) FeCl ₃	C) AlCl ₃	D) BeCl ₂
58.	Which one of the following alkaline earth metal sulphates has its hydration enthalpy greater			
	than its lattice enthalpy ?			
	A) BaSO ₄	B) SrSO ₄	C) CaSO ₄	D) BeSO ₄
59.	Which of the following ha	as the highest nucleo	philicity?	
	A) F	B) OH ⁻	C) CH_{-}^{-}	D) NH_{-}^{-}
60	Cycloboyono on ozonolyg	is followed by reseti	on with ging dust and we	\mathbf{E}
00.	Compound E on further treatment with aqueous KOH yields compound F. Compound F is			
				CO ₂ H
				CO_2H
(1	A) \sim	B) ∼ 34h - Jan Jan - 4h 4 2m		D) ∨
01.	CsBr has bee structure with edge length 4.5pm. The shortest linter ionic distance in between			
	Cs^2 and Br^2 is :	\mathbf{D} 1.00	\sim 7.44	\mathbf{D} (2)
()	A) 3.72	B) 1.80	C) /.44	D) 4.3
62.	The Henry's law constant for the solubility of N_2 gas in water at 298K is 1.0×10° atm. The			
	mole of N_2 in air is 0.8. The number of moles of N_2 from air dissolved in 10 moles of water at			
	298 K and 5atm pressure	e IS	C) 5 0 10-4	D) 40 10-0
(\mathbf{a})	A) 4.0×10	B) 4.0×10^{-1}	C) 5.0×10^{-1}	D) 4.0×10^{-1}
63.	The equivalent conductance of NaCl at concentration C and at infinite dilution are			
	λ_c and λ_{∞} respectively. The correct relationship between λ_c and λ_{∞} is given as : (where the			
	constant B is positive)			
	A) $\lambda_C = \lambda_{\infty} + (B)C$	B) $\lambda_C = \lambda_\infty - (B)C$	C) $\lambda_C = \lambda_{\infty} - (B)\sqrt{C}$	D) $\lambda_C = \lambda_{\infty} + (B)\sqrt{C}$
64	The half-life period of a r	adioactive elements	is 140 days After 560 da	avs one gram of the
UT.	element will reduced to :			
		1	1	1
	A) $\frac{1}{2}g$	B) $\frac{1}{4}g$	C) $\frac{1}{2}g$	D) $\frac{1}{16}g$
65	L In the context of the Hell	4 Honoult process for	ð • the systemation of Al wh	10 Nich of the following
05.	statements is false ? $A = A^{3+}$ is reduced at the exthade to form A1			
	A) Al is reduced at the cathode to form Al D) No. AlE, some as the electrolyte only			
	B) Na_3AIF_6 serves as the electrolyte only C) CO and CO, are produced in this process			
	C) CO and CO ₂ are produced in this process D) Al O is mixed with CoE, which lowers the matrix a point of the minture and brings conductivity.			
"	D) $A_{12}O_{3}$ is mixed with Car_{2} which lowers the method point of the mixture and offigs conductivity			
00.	Extra pure N_2 can be obt	D) NU NO	$(\mathbf{N}\mathbf{H})$ $(\mathbf{N}\mathbf{H})$ $C_{\mathbf{r}}$	\mathbf{D}) $\mathbf{P}_{\mathbf{a}}(\mathbf{N})$
(7	A) NH_3 with CuO	D) $IN\Pi_4INU_3$	C) $(N\Pi_4)_2 CI_2O_7$	D) $Da(N_{3})_{2}$
07.	A C	$\frac{101110111011001}{D} 2.84$	$C \rightarrow 4.00$	$\Gamma(CO)_6$ is
(0	A) U	D) 2.84	C 4.90	D) 3.92
00.	The structure of the major product formed in the following reaction			
	$\xrightarrow{\text{CH}_2\text{Cl}} \xrightarrow{\text{NaCN}} \xrightarrow{\text{DMF}}$			
		Ì		
	is			
	CH ₂ Cl	CH ₂ CN	CH ₂ CN	CH ₂ CN
			(\mathbf{C})	CN'
	A) CN	В) 1	\mathbf{C}) \mathbf{C}	D) I

69.







<u>SECTION-II</u> (Numerical Value Answer Type)

This section contains 5 questions. The answer to each question is a Numerical values comprising of positive or negative decimal numbers Marking scheme: +4 for correct answer, 0 in all other cases.

- 71. The dissociation constant of a substituted benzoic acid at 25° C is 1.0×10^{-4} . The pH of a 0.01 M solution of its sodium salt is
- 72. The total number of cyclic isomers possible for a hydrocarbon with the molecular formula C_4H_6 is
- 73. The total number of basic group in the following form of lysine is

$$H_{3} \overset{+}{N} - CH_{2} - CH_{2} - CH_{2} - CH_{2} \\ CH - C \\ H_{2}N \\ O^{-}$$

- 74. In neutral or faintly alkaline solution, 8 moles of permanganate anion quantitatively oxidize thiosulphate anions to produce X moles of a sulphur containing product. The magnitude of X is.
- 75. Among the following , the number of compounds those can react with PCl₅ to give POCl₃ is $O_2, CO_2, SO_2, H_2O, H_2SO_4, P_4O_{10}$