# JEE-Main-29-01-2023 (Memory Based) [Morning Shift]

## **Physics**

**Question:** A projectile was projected at an angle  $\theta = 30^{\circ}$ , then find the ratio of kinetic energy at a starting point to the kinetic energy at highest point. **Options:** 



**Question:** Point 0 and two long wires are kept in same plane such that point 0 lies at middle of the line. Then magnetic field at point 0 due to the current i flowing in both the wires is equal to





**Question:** A block is sliding down an inclined plane of inclination 30° with an acceleration of g/4. Find the coefficient of friction between the block and incline



$$a = g \sin \theta - Mg \cos \theta$$
$$\frac{g}{4} = \frac{g}{2} - \mu g \frac{\sqrt{3}}{2}$$
$$\frac{\sqrt{3}}{2} \mu g = \frac{g}{4}$$
$$\mu = \frac{1}{2\sqrt{3}}$$

**Question:** A car is moving on a circular track of radius 50 cm with coefficient of friction being 0.34. On this horizontal track the maximum safe speed for turning is equal to  $(g = 10 \text{m/s}^2)$ 



**Question:** Two resistors **R** and **3R** are in parallel, find the ratio of power dissipated in them. **Options:** 

(a) 1 : 2 (b) 2 : 1 (c) 3 : 1 (d) 1 : 3 **Answer:** (c) **Solution:**   $P = \frac{V^2}{R}$   $P \propto \frac{1}{R}$  $\frac{P_1}{P_2} = \frac{R_2}{R_1} = \frac{3R}{R}$  **Question:** In a meter bridge experiment, null point is found at a particular point for resistance 2  $\Omega$  and 3  $\Omega$ . Then resistance of X  $\Omega$  is shunted on 3  $\Omega$  resistor, and null point shifts by 22.5 cm. Then find the value of X.



Question: Find the ratio of maximum wavelength of Lyman series of Hydrogen atom to minimum wavelength of Balmer series of Helium atom.

**Options:** (a) 4/3

(b) 1 (c) 3/2 (d) 3/4 Answer: (a) Solution:

$$\frac{1}{\lambda} = R \left[ \frac{1}{n_f^2 - \frac{1}{n_i^2}} \right] Z^2$$
$$\frac{1}{\lambda_1} = R \left[ \frac{1}{1^2} - \frac{1}{2^2} \right] \times 1^2$$
$$\frac{1}{\lambda_2} = R \left[ \frac{1}{2^2} - \frac{1}{\infty^2} \right] \times 2^2$$
$$\frac{\lambda_1}{\lambda_2} = \frac{4}{3}$$

**Question:** In YDSE 1st minima is formed directly opposite to upper slit, then find d ( $\lambda = 800$  nm, D = 5 cm)



**Question:** Two objects of equal mass m are moving in a circular path of radius "r" because of their mutual gravitational attraction force. Find the velocity of each particle. **Options:** 

(a) 
$$\sqrt{\frac{Gm}{r}}$$
  
(b)  $\sqrt{\frac{Gm}{2r}}$ 

(c) 
$$\sqrt{\frac{Gm}{4r}}$$
  
(d)  $\sqrt{\frac{2Gm}{r}}$ 

Answer: (c) Solution:

$$F = \frac{Gm^2}{4r^2} = \frac{Mv^2}{r}$$
$$V = \sqrt{\frac{Gm}{4r}}$$

Question: Find coordinates of null point.



**Question:** A soap bubble of Radius R & surface tension S is blown to double the radius. Find charge in surface energy.

#### **Options:** (a) $24\pi SR^2$ (b) $12\pi SR^2$ (c) $6\pi SR^2$ (d) $3\pi SR^2$ **Answer: (a) Solution:**

$$U_{i} = (S \times 4\pi R^{2}) \times 2$$
$$U_{i} = 8\pi SR^{2}$$
$$U_{final} = S \times 4\pi (2R)^{2} \times 2$$
$$U_{f} = 32S\pi R^{2}$$
$$\Delta U = 24\pi SR^{2}$$

**Question:** There is square loop of length 'L'. A very small ring of radius r is placed at centre of square. Find coefficient of mutual inductance.



**Question:** A Tennis ball is dropped from 9.8 m it hits the ground and rebounds to 5m calculate average acceleration at point of contact t = 0.2 sec



**Options:** (a) 50 m/s<sup>2</sup> (b) 120 m/s<sup>2</sup> (c) 100 m/s<sup>2</sup> (d) 200 m/s<sup>2</sup> **Answer: (b) Solution:** 

$$V_{1} = \sqrt{2gh} = \sqrt{2 \times 10 \times 9.8} = 14 \text{ ms}^{-1}$$

$$V_{2} = \sqrt{2 \times 10 \times 5} = 10 \text{ ms}^{-1}$$

$$\vec{a} = \frac{\Delta \vec{v}}{t} = \frac{(10\hat{i}) - (-14)\hat{j}}{0.2} = \frac{240}{2}\hat{j} = 120\hat{j}$$

$$= 120 \text{ ms}^{-1}$$

**Question:** Match quantities with correct dimensions.

Column A	Column B	
1. Latent Heat	(p) $MLA^{-1}T^{-3}$	
2. Electric field	(q) $ML^{-2}T^{-2}$	
3. Pressure Gradient	(r) $ML^2A^{-1}T^{-3}$	
4. Electric Potential	(s) $L^2T^{-2}$	
Options:		
(a) 1-s, 2-p, 3-q, 4-r		
(b) 1-p, 2-s, 3-q, 4-r		
(c) 1-s, 2-p, 3-r, 4-q		
(d) 1-q, 2-p, 3-s <mark>, 4-r</mark>		
Answer: (a)		
Solution:		
$L = \frac{Q}{m} = \frac{ML^2T^{-2}}{M}$		
$E = \frac{F}{q} = \frac{MLT^{-2}}{AT}$		
$\frac{dP}{dx} = \frac{ML^{-1}T^{-2}}{L} = ML^2T$	2	
$V = \frac{P.E.}{2} = \frac{ML^2T^{-2}}{AT}$		

**Question:** A nucleus  ${}^{236}X_{92}$  undergoes two  $\alpha$  decays and one  $\beta^-$  decay. Find atomic number and mass number.

#### **Options:**

(a) Z = 88, Z = 227(b) Z = 90, Z = 225(c) Z = 89, Z = 228(d) Z = 81, Z = 210 **Answer: (c) Solution:**   $2\alpha$  decays means A-8 and Z-4 236 - 8 = 228  $1\beta$  decay means Z + 1 $\therefore Z = 94 - 4 + 1 = 89$  **Question:** A wire has been made into a circle and placed in constant magnetic field. If the radius of wire starts decreasing at rate of 2 cm/s when wire starts shrinking, find the EMF induced in the wire when radius of wire is 10 cm. Magnetic field strength in the region is 0.8 T.

#### **Options:**

(a) 0.01 V (b) 0.1 V (c) 1 V (d) 10 V Answer: (a) Solution: Flux via circular wire is given by  $\phi = B.\pi r^2$ 

So emf induced  $|\varepsilon| = \left| \frac{d\phi}{dt} \right| = B.2\pi r \left| \frac{dr}{dt} \right|$ =  $0.8 \times 2\pi \times \frac{10}{100} \times \frac{2}{100}$ =  $\frac{3.2\pi}{1000} \approx \frac{10}{1000} = \frac{1}{100 \text{ V}}$ 

**Question:** Two antenna's has height 80 m, what is the range to which signal can be transmitted?

**Options:** (a) 28 km (b) 10 km (c) 64 km (d) 51 km **Answer:** (c) **Solution:**   $d = 2\sqrt{2Rh}$   $= 2\sqrt{2 \times 6400 \times 80 \times 10^{-3}}$   $= \sqrt{8 \times 80 \times 6400 \times 10^{-3}}$ = 64 km

**Question:** Cuboid of dimensions  $2L \times 2L \times L$ , A charge q is placed at centre for of S with area  $4L^2$  flux through opposite side is? **Solution:** 

 $q/6\varepsilon_0$ 

**Question:** A solid sphere is released from point 0 at the top of an incline as shown. Find the value of velocity of centre of mass of sphere at the bottom most point of the incline after it reaches there doing pure rolling ( $g = 10 \text{ m/s}^2$ )



Question: KE of a solid sphere is 4220 J on a horizontal plane. Find the velocity of COM of the sphere

- Options:
- (a)  $40\sqrt{7}$
- (b)  $40\sqrt{3}$
- (c)  $30\sqrt{7}$ (d)  $30\sqrt{3}$

Answer: (a) Solution:

$$mgh = \frac{1}{2}mv^{2} + \frac{1}{2}I\left(\frac{V^{2}}{R^{2}}\right)$$

$$V = \sqrt{\frac{2gh}{1 + \frac{I}{MR^{2}}}} = \sqrt{\frac{2 \times 10 \times 7}{1 + \frac{2}{5}}} = \sqrt{\frac{20 \times 7 \times 5}{7}} = 10 \text{ ms}^{-1}$$

$$K = \frac{1}{2}mV^{2} + \frac{1}{2}\left[\frac{2}{5}MR^{2}\right] \times \frac{V^{2}}{R^{2}}$$

$$K = \frac{7}{10}MV^{2} = 2240$$

$$V^2 = \frac{112 \times 100}{7} = 40 \text{ms}^{-1}$$

**Question:** A disk of radius R is given  $\omega_0$  angular speed and placed gently on a rough horizontal surface. Find the velocity of centre of disk when pure rolling starts.





# JEE-Mains-29-01-2023 (Memory Based) [Morning Shift]

### Chemistry

**Question:** Which of the following is paramagnetic in nature? **Options:** 

(a)  $Na_2O_2$ 

(b) KO<sub>2</sub>

(c) Both (a) and (b)

(d) None of the above

Answer: (b)

Solution: The molecular electronic configuration of

 $\mathbf{O}_{2}^{-} = (\sigma_{1s})^{2} (\sigma_{1s}^{*})^{2} (\sigma_{2s})^{2} (\sigma_{2s}^{*})^{2} (\sigma_{2p_{z}})^{2} (\pi_{2p_{x}})^{2} (\pi_{2p_{y}})^{2} (\pi_{2p_{x}}^{*})^{2} (\pi_{2p_{x}}^{*})^{1}$ 

 $\mathbf{O}_{2}^{2-} = (\sigma_{1s})^{2} (\sigma_{1s}^{*})^{2} (\sigma_{2s})^{2} (\sigma_{2s}^{*})^{2} (\sigma_{2p_{z}})^{2} (\pi_{2p_{x}})^{2} (\pi_{2p_{y}})^{2} (\pi_{2p_{x}}^{*})^{2} (\pi_{2p$ 

Since  $O_2^-$  has unpaired electron in its antibonding molecular orbital, thus it is paramagnetic and oppositely  $O_2^{2-}$  is diamagnetic.

Question: Arrange the following (increasing order of PKa)

Phenol, 2,4-dinitrophenol, 2,4,5-trimethyl phenol, 4-nitrophenol, 4-chlorophenol **Options:** 

(a) Phenol < 2,4-dinitrophenol < 2,4,5-trimethyl phenol < 4-nitrophenol < 4-chlorophenol (b) 2,4-dinitrophenol < Phenol < 2,4,5-trimethyl phenol < 4-nitrophenol < 4-chlorophenol (c) 2,4,5-trimethyl phenol < 4-chlorophenol < Phenol < 2,4-dinitrophenol < 4-nitrophenol (d) 2,4-dinitrophenol < 4-nitrophenol < Phenol < 4-chlorophenol < 2,4,5-trimethyl phenol Answer: (d)

Compound	Pka value			
Phenol	9.98			
2,4-dinitrophenol	4.114			
2,4,5-trimethyl phenol	10.57			
4-nitrophenol	7.14			
4-chlorophenol	10.07			

**Question:** At low pressure Van der Waals equation will be **Options:** 

(a) 
$$\left[1 + \frac{a}{RTV}\right]$$
  
(b)  $\left[1 - \frac{RTV}{a}\right]$ 



(c) 
$$\left[1 - \frac{a}{RTV}\right]$$
  
(d)  $\left[1 + \frac{RTV}{a}\right]$   
Answer: (c)  
Solution:  
 $pV + \frac{a}{v} = RT$   
 $\therefore \frac{pV}{RT} + \frac{a}{RTV} = 1 \quad \therefore Z = \frac{pV}{RT}$   
 $= \left[1 - \frac{a}{RTV}\right]$ 

 $\begin{array}{l} \textbf{Question: Order of Hydration: } Mg^{2+}, K^{+}, Rb^{+}, Cs^{+}, Ca^{2+} \\ \textbf{Options:} \\ (a) \ Mg^{2+} > Ca^{2+} > K^{+} > Rb^{+} > Cs^{+} \\ (b) \ Mg^{2+} > Ca^{2+} > Cs^{+} > Rb^{+} > K^{+} \\ (c) \ Ca^{2+} > Mg^{2+} > Cs^{+} > Rb^{+} > K^{+} \\ (d) \ Cs^{+} > Mg^{2+} > K^{+} > Rb^{+} > Ca^{2+} \\ \textbf{Answer: (a)} \\ \textbf{Solution: } Mg^{2+} > Ca^{2+} > K^{+} > Rb^{+} > Cs^{+} \end{array}$ 

**Question:** Which compound will give both lassaigne test of nitrogen and halogen? **Options:** 







#### Question: Number of Bridging CO in W(CO)<sub>6</sub> and Mn<sub>2</sub>(CO)<sub>10</sub> Options: (a) 0 (b) 1 (c) 2

(d) 3 Answer: (a)

Solution:



**Question:** Order of bond dissociation enthalpy of Cl<sub>2</sub>, **Br**<sub>2</sub>, **I**<sub>2</sub>, **F**<sub>2</sub> **Options:** 

(a)  $I_2 > Br_2 > Cl_2 > F_2$ (b)  $Cl_2 > Br_2 > F_2 > I_2$ (c)  $Br_2 > I_2 > F_2 > Cl_2$ (d)  $F_2 > Cl_2 > Br_2 > I_2$ 

#### Answer: (b)

**Solution:** Fluorine which is an exception for this order, due to the high electronegativity of fluorine atom it tends to break the bond present between the fluorine molecule hence requiring less energy for the breaking of the bond which results in requiring low bond dissociation enthalpy.

**Question:** Which of the following ions does not liberate hydrogen gas on reaction with dilute acids?

#### **Options:**

(a)  $Cr^{2+}$ (b)  $Mn^{2+}$ (c)  $Ti^{2+}$ 

### (d) V<sup>2+</sup>

#### Answer: (b)

**Solution:**  $Mn^{2+}$  does not liberate hydrogen gas on reaction with dilute acids. +2 oxidation state of Mn is very stable due to exactly half filled electronic configuration.



Question: Which of the following is incorrect statement about the following graph?



#### **Options:**

(a)  $\lambda^{o}$ m value of weak electrolyte can be determined by extrapolating the graph

(b)  $\lambda^{o}$ m value of weak electrolyte cannot be determined by extrapolating the graph

(c)  $\lambda^{o}$ m value of strong electrolyte can be determined by extrapolating as linear graph is obtained

(d) All of these

#### Answer: (a)

**Solution:** For weak electrolyte molar conductivity at infinite dilution cannot be determined experimentally and by extrapolation because graph is not linear, maximum value of molar conductivity at infinite dilution cannot be obtained.

#### Question: Match the following.

<u> </u>	
Reaction (Column-I)	<b>Reagents (Column-II)</b>
(A) Hofmann Degradation	(i) Conc KOH
( <b>B</b> ) Clemmensen reaction	(ii) NaOH, Br <sub>2</sub>
(C) Cannizzaro reaction	(iii) Zinc-Hg, HCl
( <b>D</b> ) Reimer-Tiemann reaction	(iv) CHCl <sub>3</sub> , NaOH

#### **Options:**

(a) A - (ii); B - (i); C - (iii); D - (iv)

(b) A - (iii); B - (i); C - (ii); D - (iv)

(c) A - (iv); B - (iiii); C - (ii); D - (i)

(d) A - (ii); B - (iii); C - (i); D - (iv)

#### Answer: (d) Solution:

Solution				
Reaction (Column-I)	Reagents (Column-II)			
(A) Hofmann Degradation	(i) NaOH, Br <sub>2</sub>			
( <b>B</b> ) Clemmensen reaction	(ii) Zinc-Hg, HCl			
(C) Cannizzaro reaction	(iii) Conc KOH			
( <b>D</b> ) Reimer-Tiemann reaction	(iv) CHCl <sub>3</sub> , NaOH			

#### **Question:**





Answer: (d)



Ph – CH<sub>2</sub> – CH<sub>2</sub> – CH<sub>2</sub> – CH<sub>2</sub> – CH<sub>2</sub>CoNH<sub>2</sub>  $\downarrow$  (ii) LiAlH<sub>4</sub> / H<sub>3</sub>O<sup>+</sup> PhCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>

**Question:** Number of millimoles of  $Ca(OH)_2$  in 100 mL solution, given pH = 12 **Options:** 

(a) 1 (b) 0.5 (c) 0.05 (d) 2 **Answer: (b) Solution:**  pH = 12 pOH = 14 - Ph = 14 - 12 = 2  $\therefore [OH^{-}] = 10^{-2} M$   $\therefore [Ca(OH)_2] = 0.5 \times 10^{-2} M$ Mili moles of Ca(OH)<sub>2</sub> in solution = MV (in mL)  $= 0.5 \times 10^{-2} \times 100 = 0.5$ 

Question:  $K_f = 10^3$ ,  $K_b = 10^2$  find  $\Delta G^o$  at  $\Delta T = 27^oC$ Options: (a) -700 R (b) -500 R (c) -690 R (d) -400 R Answer: (c)



#### Solution:

 $\Delta G^{o} = -2.303 \text{ RT } \log \text{ K}_{\text{C}}$ = -2.303 R × 300 log  $\frac{\text{K}_{\text{F}}}{\text{K}_{\text{b}}}$ = -2.303 R × 300 log  $\frac{10^{3}}{10^{2}}$ = -2.303 R × 300 log\_{10} = -2.303 R × 300 × 1 = -690 R

**Question:** Which of the following metalist is purified by Mond Process **Options:** 

(a) Ti

(b) Ge (c) Ni (d) Zr Answer: (c) Solution: Ni + 4CO  $\xrightarrow{330-350K}$  Ni(CO)<sub>4</sub> Ni(CO)<sub>4</sub>  $\xrightarrow{450-470K}$  Ni + CO

**Question:** Which of the following is not odd electron?

 $NO_{2}, NO_{2}^{+}, ClO_{2}, BrCl_{3}$ 

#### **Options:**

(a) BrCl<sub>3</sub> (b) NO (c) NO<sub>2</sub> (d) NO<sub>2</sub><sup>+</sup> and ClO<sub>2</sub> Answer: (a) Solution:

:CI-Br-CI: CI:

**Question:** Which of the following compound used as Narrow spectrum antibiotics? **Options:** 

- (a) Penicillin
- (b) Gentamycin
- (c) Erythromycin
- (d) All of these

#### Answer: (d)

**Solution:** Narrow spectrum antibiotics are used against only limited pathogens. Penicillin, gentamycin and erythromycin are used to treat only certain pathogens such as Staphylococci, Clostridia, Streptococci and hence fall under the class of narrow spectrum antibiotics.

**Question:** Select the correct statement among the following **Options:** 



(a) Photochemical smog has high concentration of oxidising agent

(b) Classical smog has high concentration of oxidising agent

(c) Classical smog contains NO<sub>2</sub>

(d) None of these

#### Answer: (a)

#### Solution:

**Smog:** The word smog is derived from smoke and fog.

This is the most common example of air pollution that occurs in many cities throughout the world.

There are two types of smog:

(a) Classical smog occurs in cool humid climate. It is a mixture of smoke, fog and sulphur dioxide. Chemically it is a reducing mixture and so it is also called as reducing smog.
(b) Photochemical smog occurs in warm, dry and sunny climate. The main components of the photochemical smog result from the action of sunlight on unsaturated hydrocarbons and nitrogen oxides produced by automobiles and factories. Photochemical smog has high concentration of oxidising agents and is. therefore, called as oxidising smog.
Formation of photochemical smog, when fossil fuels are burnt, a variety of pollutants are emitted into the earth's

**Question:** In ostwald's process Ammonia is oxidized to give A and oxidised again to give B and forms an oxoacid of nitrogen. B gives positive brown ring test. What are A and B? **Options:** 

(a) A = NO,  $B = NO_2$ (b)  $A = NO_2$ , B = NO(c)  $A = HNO_2$ , B = NO(d)  $A = HNO_2$ ,  $B = NO_2$ 

#### Answer: (b)

**Solution:** Ostwald's process for the manufacture of nitric acid.  $4NH_3(g) + 5O_2(g) \rightarrow 4NO(g) + 6H_2O(g)$   $2NO(g) + O_2(g) \rightarrow 2NO_2(g)$  $4NO_2(g) + 2H_2O(l) + O_2(g) \rightarrow 4HNO_3(aq)$ 

# **Question:** Which of the following tank use to store hydrogen? **Options:**

- (a) High pressure tank
- (b) Low pressure tank
- (c) High temperature tank

(d) Low temperature tank

#### Answer: (a)

**Solution:** Hydrogen can be stored physically as either a gas or a liquid. Storage of hydrogen as a gas typically requires high-pressure tanks (350-700 bar [5,000-10,000 psi] tank pressure). Storage of hydrogen as a liquid requires cryogenic temperatures because the boiling point of hydrogen at one atmosphere pressure is -252.8°C.

**Question:** The compound formed in borax bead test, of CuSO<sub>4</sub> in oxidizing flame is \_\_\_\_\_ **Options:** 

(a) Cu(BO<sub>2</sub>)<sub>2</sub>
(b) Cu<sub>3</sub>B<sub>2</sub>
(c) Cu(BO)<sub>2</sub>
(d) Cu(BO<sub>3</sub>)<sub>2</sub>



Answer: (a) Solution:  $CuSO_4(aq) \xrightarrow{\Delta} CuO(s) + SO_3(g)$  $CuO(s) + B_2O_3(s) \rightarrow Cu(BO_2)_2$ Cupric metaborate





# JEE-Main-29-01-2023 (Memory Based) [Morning Shift]

## **Mathematics**

Question: If f(x+y) = f(x) + f(y),  $f(1) = \frac{1}{5}$  and  $\sum_{n=1}^{m} \frac{f(n)}{n(n+1)(n+2)} = \frac{1}{12}$ , then find m.

Answer: 10.00 Solution: f(x+y) = f(x) + f(y) f(x) = ax  $x = 1 \Rightarrow f(1) = a = \frac{1}{5}$   $f(x) = \frac{1}{5}x$   $\frac{1}{5}\sum \frac{n}{n(m+1)(n+2)} = \frac{1}{12}$   $\frac{1}{5}\left[\sum_{n=1}^{m} \left(\frac{1}{n+1} - \frac{1}{n+2}\right)\right] = \frac{1}{12}$   $\frac{1}{2} - \frac{1}{m+2} = \frac{5}{12}$   $\Rightarrow m = 10$ 

Question: Tangent at A(4, -11) and B(8, -5) to  $x^2 + y^2 - 3x + 10y - 15 = 0$  intersect at C. Find the radius of circle touching AB and having centre at C.

Answer:  $\frac{2\sqrt{13}}{3}$ Solution:





Radius  $r = \sqrt{\frac{9}{4} + 25 + 15} = \frac{13}{2}$ 

$$AM = \frac{1}{2}AB$$
$$= \frac{1}{2}\sqrt{16+36}$$
$$= \sqrt{13}$$

$$\frac{CM}{AM} = \cot \theta$$

$$CM = \sqrt{13} \cot \theta$$

$$\Delta OAM \Longrightarrow OM = \sqrt{\left(\frac{13}{2}\right)^2 - 13} = \sqrt{\frac{169 - 52}{4}} = \frac{\sqrt{117}}{2}$$

$$CM = \sqrt{13} \cot \theta$$

$$CM = \sqrt{13} \times \frac{\sqrt{13}}{\sqrt{117}} \times 2 = \sqrt{13} \times \frac{\sqrt{13}}{\sqrt{13} \times 3} \times 2 = \frac{2\sqrt{13}}{3}$$

**Question:** Five digit numbers are formed using 1, 2, 3, 5, 7 (repetition is allowed), and these numbers are arranged in descending order. Find the rank of 35337.

# Answer: 1436.00 Solution:

Five digit number using 35337

$$\frac{7}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad = 5^{4}$$

$$\frac{3}{3} \quad \frac{7}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad = 5^{3}$$

$$\frac{3}{3} \quad \frac{5}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad \frac{5}{5} \quad = 5^{3}$$



 $\frac{3}{3} \quad \frac{5}{5} \quad \frac{7}{5} \quad \frac{5}{5} \quad \frac{5}{5} = 5^{2}$   $\frac{3}{3} \quad \frac{5}{5} \quad \frac{5}{3} \quad \frac{5}{7} \quad \frac{5}{5} = 5^{2}$   $\frac{3}{3} \quad \frac{5}{5} \quad \frac{3}{3} \quad \frac{7}{5} \quad \frac{5}{5} = 5$   $\frac{3}{3} \quad \frac{5}{5} \quad \frac{3}{3} \quad \frac{3}{5} \quad \frac{7}{5} = 1$ 

So total =  $5^4 + 5^4 + 5^3 + 5^3 + 5^2 + 5^2 + 5 + 5 + 1 = 1436$ 

Question: A function f(x) is such that f(x+y) = f(x) + f(y) - 1,  $\forall x, y \in R$ . If f'(0) = 2Answer: 3.00 Solution: f(x+y) = f(x) + f(y) - 1  $x = y = 0 \Rightarrow f(0) = f(0) + f(0) - 1 \Rightarrow f(0) = 1$   $f'(x) = \lim_{h \to 0} \frac{f(x) + f(h) - 1 - f(x)}{h}$   $= \lim_{h \to 0} \frac{f(h) - 1}{h}$  = f'(0) = 2 f'(x) = 2 f(x) = 2x + c  $1 = 0 + c \Rightarrow c = 1$   $\therefore f(x) = 2x + 1$ |f(-2)| = |-3| = 3

**Question:** If the 3 consecutive coefficients in the expansion of  $(1+2x)^n$  are in the ratio 2:5:8 then the middle term is **Answer:**  ${}^8C_4(2x)^4$  **Solution:**  ${}^nC_{r-1}(2)^{r-1}:{}^nC_r2^r:{}^nC_{r+1}2^{r+1}::2:5:8$ 



$$\frac{{}^{n}C_{r}2^{r}}{{}^{n}C_{r-1}2^{r-1}} = \frac{5}{2}$$
$$2\left(\frac{n-r+1}{r}\right) = \frac{5}{2}$$

Similarly,  $2\left(\frac{n-r}{r+1}\right) = \frac{8}{5}$ 

Comparing n = 8

Mid term  $= {}^{8}C_{4}(2x)^{4}$ 

Question: If  $\frac{dy}{y} = \left(\frac{x+1}{x^2}\right) dx$ ; y(1) = e, then  $\lim_{x \to 0^+} f(x) = ?$ Answer: 0.00 Solution:  $\int \frac{dy}{y} = \int \left(\frac{1}{x} + \frac{1}{x^2}\right) dx$  $\ln|y| = \ln|x| - \frac{1}{x} + c$ 

Given (1, e)

1 = 0 - 1 + c

$$c = 2$$

: We have

$$\ln|y| = \ln|x| - \frac{1}{x} + 2$$
$$y = e^{\ln|x| - \frac{1}{x} + 2}$$

As we take  $\lim_{x\to 0^+}$ 

$$y = e^{-\infty} = 0$$

**Question:**  $\alpha, \beta$  are positive numbers. *A* is a 3×3 matrix such that  $A^2 = 3A + \alpha I$  and  $A^4 = 21A + \beta I$ . Find  $\alpha, \beta$ . **Answer: -1, -8 Solution:** Given,  $A^2 = 3A + \alpha I$  and  $A^4 = 21A + \beta I$  $A^4 = A^2 \cdot A^2$ 



 $A^{4} = (3A + \alpha I)(3A + \alpha I) = 21A + \beta I$   $9A^{2} + 3A\alpha I + 3\alpha AI + \alpha^{2}I = 21A + \beta I$   $9A^{2} + 6\alpha A + \alpha^{2}I = 21A + \beta I$ Again using  $A^{2} = 3A + \alpha I$  in LHS  $\Rightarrow 9(3A + \alpha I) + 6\alpha A + \alpha^{2}I = 21A + \beta I$   $\Rightarrow (27 + 6\alpha)A + (9\alpha + \alpha^{2})I = 21A + \beta I$   $\therefore 27 + 6\alpha = 21 & 89\alpha + \alpha^{2} = \beta$   $6\alpha = -6 & 8 & 1 - 9 = \beta$  $\alpha = -1 & 8 & \beta = -8$ 

Question: Consider a function  $f(x) = \frac{x^2 + 2x + 1}{x^2 + 1}$ , then which of the following is correct?

#### **Options:**

(a) f(x) is one-one for  $x \in (0, \infty)$ (b) f(x) is one-one for  $x \in (1, \infty)$ (c) f(x) is one-one for  $x \in (2, \infty)$  and many-one for  $x \in (-\infty, 0]$ (d) Answer: (c) Solution:  $f(x) = \frac{(x^2+1)+2x}{x^2+1} = 1 + \frac{2x}{x^2+1}$   $f'(x) = \frac{(x^2+1)^2 - 2x(2x)}{(x^2+1)^2}$  $f'(x) = \frac{2-2x^2}{(x^2+1)^2}$ 

**Question:** If real part of the product of  $z_1 \& z_2$  is zero i.e.,  $\operatorname{Re}(z_1z_2) = 0 \& \operatorname{Re}(z_1 + z_2) = 0$ then  $\operatorname{Im}(z_1) \& \operatorname{Im}(z_2)$  is



Answer:  $Im(z_1) \& Im(z_2)$  are of opposite signs Solution:

Given product of  $z_1 \& z_2$  is zero

i.e., 
$$\operatorname{Re}(z_1 \cdot z_2) = 0$$
  
 $z_1 \cdot z_2 = (x_1 + iy_1) \cdot (x_2 + iy_2)$   
 $= (x_1x_2 - y_1y_2)$   
 $\operatorname{Re}(z_1z_2) = x_1x_2 - y_1y_2 = 0$   
 $\Rightarrow x_1x_2 = y_1y_2 \quad \dots(i)$   
 $\operatorname{Re}(z_1 + z_2) = (x_1 + iy_1) + (x_2 + iy_2)$   
 $\operatorname{Re}(z_1 + z_2) = (x_1 + x_2) = 0$   
 $x_2 = -x_1$   
Substitute  $x_2$  in (i)  
 $-x_1 \cdot x_1 = y_1y_2$   
 $\Rightarrow y_1y_2 = -x_1^2 = -ve$ 

 $\operatorname{Im}(z_1)$  &  $\operatorname{Im}(z_2)$  is one positive and one negative.

**Question:** If  $a_1, a_2, a_3, \dots, a_n$  is an increasing GP such that  $a_4 \times a_6 = 9$  and  $a_5 + a_7 = 12$ . Find  $a_7 + a_9 = ?$  **Answer: 36.00 Solution:** Given  $a_5 + a_7 = 12$  and  $a_4 \times a_6 = 9$   $a_5 + a_7 = 12$  $ar_4 + ar^6 = 12$ 

$$ar^{4}(1+r^{2}) = 12 \qquad \dots(1)$$
$$a_{4} \cdot a_{6} = 9$$
$$ar^{3} \cdot ar^{5} = 9$$
$$a^{2} \cdot r^{8} = 9$$

 $a \cdot r^4 = 3$ 

Substitute in (1)



 $3(1+r^{2}) = 12$   $r^{2} = 3$   $r = \pm\sqrt{3}$   $a = \frac{1}{3}$ Now  $a_{7} + a_{9}$   $\Rightarrow ar^{6} + ar^{8} = ar^{4}(r^{2} + r^{4})$   $= 3 \times r^{2}(1+r^{2})$   $= 3 \times 3(1+3)$   $= 9 \times 4$  = 36

**Question:**  $\Delta$  is the area between  $x^2 + y^2 \le 21$ ,  $y^2 \le 4x$  and  $x \ge 1$ . Find





$$2 \times \int_{1}^{3} 2\sqrt{x} dx = 4 \left[ \frac{2}{3} x^{\frac{3}{2}} \right]_{1}^{3} = \frac{8}{3} \left( 3\sqrt{3} - 1 \right)$$

$$2 \int_{3}^{\sqrt{21}} \sqrt{21 - x^{2}} = 2 \left[ \frac{1}{2} x\sqrt{21 - x^{2}} + \frac{1}{2} \times 21 \sin^{-1} \left( \frac{x}{\sqrt{21}} \right) \right]_{3}^{21}$$

$$= 0 + 21 \frac{\pi}{2} 3 \times 2\sqrt{3} - 21 \sin^{-1} \left( \frac{\sqrt{3}}{\sqrt{7}} \right)$$

$$= 21 \left( \cos^{-1} \sqrt{\frac{3}{7}} \right) - 6\sqrt{3}$$

$$= 21 \sin^{-1} \frac{2}{\sqrt{7}} - 6\sqrt{3}$$

$$\Delta = \frac{8}{3} \left( 3\sqrt{3} - 1 \right) + 21 \sin^{-1} \frac{2}{\sqrt{7}} - 6\sqrt{3}$$

$$\Delta = \sqrt{3} - \frac{4}{3}$$





$$(1 \times 1) + 2(\sqrt{2} - 1) + \int_{\sqrt{2}}^{2} x^{2} dx$$
$$= \frac{5 + 4\sqrt{2}}{3}$$

**Question:** 4 apples are picked one by one without replacing from a bag containing 3 rotten and 7 normal apples. Let x be no. of rotten apples. Find  $\overline{x} + v_x$ .

Answer:  $\frac{6}{5}, \frac{14}{25}$ Solution:

Solution.				
Х	0	1	2	3
P(X)	$\frac{{}^{7}C_{4}}{{}^{10}C_{4}}$	$\frac{{}^{3}C_{1} \times {}^{7}C_{4}}{{}^{10}C_{4}}$	$\frac{{}^{3}C_{2} \times {}^{7}C_{2}}{{}^{10}C_{4}}$	$\frac{{}^7C_1}{{}^{10}C_4}$

$$\overline{x} = \sum x_i P_i$$
$$\overline{x} = \frac{6}{5}$$

Variance =  $V = \sum x_i^2 P_i - (\overline{x})^2$  $V = \frac{14}{25}$ 

Question: Domain of 
$$f(x) = \frac{\log_x(x-1)}{\log_{x-1}(x-4)}$$
 is:

**Answer:**  $x \in (4, \infty) \setminus \{5\}$ 

#### Solution:

For domain

$$x > 0, x - 1 > 0, x \neq 1$$
  
&  $x - 1 > 0, x - 1 \neq 1, x - 4 > 0$   
$$\log_{x-1} (x - 4) \neq 0$$
  
$$\Rightarrow x - 4 \neq 1 \Rightarrow x \neq 5$$
  
$$\therefore x \in (4, \infty) - \{5\}$$

Question: If the coefficient of  $x^5$  in the expansion of  $\left(ax^3 + \frac{1}{\beta x}\right)^{11}$  and  $\left(\alpha x + \frac{1}{\beta x^3}\right)^{11}$  are

equal, then the value of  $(\alpha + \beta)^2$  is Answer: 1.00 Solution:

General term of 
$$\left(\alpha x^3 + \frac{1}{\beta x}\right)^{11}$$
 is  
 $T_{k+1} = {}^{11}C_k \left(\alpha x^3\right)^{11-k} \left(\frac{1}{\beta x}\right)^k$   
 $= {}^{11}C_k \alpha^{11-k} \beta^{-k} x^{33-4k}$ 

Now for coefficient of  $x^9$ , we have

$$33-4k = 9$$
$$\Rightarrow 4k = 24$$
$$\Rightarrow k = 6$$

Similarly, general term of  $\left(\alpha x + \frac{1}{\beta x^3}\right)^{11}$  is

$$T_{k+1} = {}^{11}C_k \left(\alpha x\right)^{11-k} \left(\frac{1}{\beta x^3}\right)^k$$
$$= {}^{11}C_k \alpha^{11-k} \beta^{-k} x^{11-4k}$$

For coefficient of  $x^{-9}$ , we have

$$11-4k = 9$$
  

$$\Rightarrow 4k = 20$$
  

$$\Rightarrow k = 5$$
  

$${}^{11}C_6 \frac{\alpha^5}{\beta^6} = {}^{11}C_5 \cdot \frac{\alpha^6}{\beta^5}$$
  

$$\Rightarrow \alpha\beta = \frac{{}^{11}C_6}{{}^{11}C_5} = 1$$
  

$$\Rightarrow (\alpha\beta)^2 = 1$$

**Question:** Consider 3 coplanar vector  $\vec{a} = 3\hat{i} - 4\hat{j} + \lambda\hat{k}$ ,  $\vec{b} = 4\hat{i} + 3\hat{j} - \hat{k}$  and  $\vec{c} = \hat{i} + 3\hat{j} - 4\hat{k}$ . Then  $9\lambda$  is \_\_\_\_\_

Answer: 87.00

#### Solution:



 $\begin{vmatrix} 3 & -4 & \lambda \\ 4 & 3 & -1 \\ 1 & 3 & -4 \end{vmatrix} = 0$  $-27 - 60 + 9\lambda = 0$  $9\lambda = 87$ 

