## JEE-Mains-06-04-2023 [Memory Based] <br> [Evening Shift]

## Physics

Question: If initial velocity $30 \mathrm{~m} / \mathrm{s}$ and acc is $2 \mathrm{~m} / \mathrm{s}^{2}$ after how much time vel will be $60 \mathrm{~m} / \mathrm{s}$ Options:
(a) 5 sec
(b) 10 sec
(c) 15 sec
(d) 20 sec

Answer: (c)
Question: If $\mathrm{a}_{0}$ is bohr radius, find de broglie wavelength of electron in 3rd orbit of hydrogen atom

## Options:

(a) $6 \pi a_{0}$
(b) $3 \pi \mathrm{a}_{0}$
(c) $9 \pi \mathrm{a}_{0}$
(d) $\pi \mathrm{a}_{0}$

Answer: (a)

## Solution:

$r=a_{0} \frac{n^{2}}{2}=a_{0} n^{2}$
$\lambda=\frac{h}{m v}=\frac{h}{n h}=2 \pi a_{0} n^{2}$
$m v r=\frac{n h}{2 \pi}=2 \lambda a_{0} n=6 \pi a_{0}$

Question: If in 7 minutes body cools from $60^{\circ} \mathrm{C}$ to $40^{\circ} \mathrm{C}$. Surrounding temperature is $10^{\circ} \mathrm{C}$ then what will be the temperature of body in next 7 minutes?
Options:
(a) $16^{\circ}$
(b) $20^{\circ}$
(c) $28^{\circ}$
(d) $36^{\circ}$

Answer: (c)
Solution:
$-\frac{d \theta}{d t}=b\left(\theta_{a v}-\theta_{0}\right)$
$\Rightarrow \frac{20}{7}=b[50-10] \ldots .(i)$
$\Rightarrow \frac{40-T}{7}=b\left[\frac{40+T}{2}-10\right] \ldots(i i)$

Question: In circular motion $M=5 \mathrm{~kg}, \mathrm{~T}=\pi$ seconds, $\mathrm{R}=2 \mathrm{~m}$. Find centrifugal force.
Answer: 40.00
Question: Minimum amplitude is 3 V, modulation index is $60 \%$ Find maximum amplitude Options:
(a) 5 V
(b) 10 V
(c) 12 V
(d) 15 V

Answer: (c)

## Solution:

Question: Planet is moving in elliptical path which is incorrect?
Options:
(a) Areal velocity is constant
(b) Total energy is constant
(c) Velocity is constant
(d) none of these

Answer: (c)

## Solution:

Question: A body is dropped from a height of $h_{1}$ and after falling it rises to a height of $h_{2}$, such that the ratio of speed before and after collision is 4 . Find percentage loss in KE.
Options:
(a) $6.25 \%$
(b) $4.34 \%$
(c) $10.00 \%$
(d) $20.00 \%$

Answer: 93.75\%

## Solution:


$K E_{i}=\frac{1}{2} m u^{2}$
$K E_{f}=\frac{1}{2} m\left(\frac{u}{4}\right)^{2}$
$\% \Delta K E=\frac{K E_{f}-K E_{i}}{K E_{i}}=\frac{\frac{1}{2} m \frac{u^{2}}{16}-\frac{1}{2} m u^{2}}{\frac{1}{2} m u^{2}}=\left(\frac{1}{16}-1\right)=\frac{-15}{16}$
or $93.75 \%$
Question: A body weighs 100 N on surface of earth find weight of body at height $\mathrm{H}=\mathrm{R} / 4$ from surface of earth ( $\mathrm{R}=$ radius of earth)

## Options:

(a) 50 N
(b) 64 N
(c) 25 N
(d) 80 N

## Answer: (b)

## Solution:

Question: A solid sphere and a ring have equal masses and equal radii of gyration. If sphere is rotating about its diameter and ring about an axis passing through centre and perpendicular to its plane, then the ratio of radius is $\sqrt{\left(\frac{x}{2}\right)}$ then find the value of $x$.
Answer: 5.00

## Solution:


$m_{1}=m_{2}$
$K_{1}=K_{2}$
$\sqrt{\frac{2}{5} \frac{m_{1} R_{1}^{2}}{m_{1}}}=\sqrt{\frac{m_{2} R_{2}^{2}}{m_{2}}}$
$\sqrt{\frac{2}{5}} R_{1}=R_{2}$
$\frac{R_{1}}{R_{2}}=\sqrt{\frac{5}{2}}$
So $x=5$

Question: A proton is projected with speed $v$ in magnetic field $B$ of magnitude $1 T$. If angle between velocity and magnetic field is $60^{\circ}$ as shown below. Kinetic energy of proton is 2 eV (mass of proton $=1.67 \times 10^{-27} \mathrm{~kg}, \mathrm{e}=1.6 \times 10^{-19} \mathrm{C}$ ). The pitch of the path of proton is approximately.


## Options:

(a) $6.28 \times 10^{-2} \mathrm{~m}$
(b) $6.28 \times 10^{-4} \mathrm{~m}$
(c) $3.14 \times 10^{-2} \mathrm{~m}$
(d) $3.14 \times 10^{-4} \mathrm{~m}$

Answer: (b)
Solution:
Pitch $=\mathrm{V}^{\prime \prime} \mathrm{T}=\operatorname{vcos} 60^{\circ} \frac{2 \pi m}{9 \beta}$
$\Rightarrow \frac{1}{2} m v^{2}=k E \Rightarrow v=\sqrt{\frac{2 K E}{m}}$
$\Rightarrow P=\sqrt{\frac{2 K E}{m}} \cdot \frac{1}{2} \cdot \frac{2 \pi m}{9 \beta}$
$=(\sqrt{2 \mathrm{kEm}}) \frac{\pi}{9 B}=\frac{\pi \sqrt{2 \times 2 \times 1.6 \times 10^{-19} \times 1.67 \times 10^{-27}}}{1.6 \times 10^{-19} \times 1}=6.3 \times 10^{-4}$

Question: Find the ratio of RMS speed of oxygen molecules to that of Hydrogen molecules at same temperature

## Options:

(a) $1 / 4$
(b) $1 / 8$
(c) $1 / 16$
(d) $1 / 32$

## Answer: (a)

Solution:
$\frac{V_{r m s} O_{2}}{V_{r m s} H_{2}}=\frac{\sqrt{\frac{3 R T}{32 \times 10^{-3}}}}{\sqrt{\frac{3 R T}{2 \times 10^{-3}}}}=\sqrt{\frac{2}{32}}=\frac{1}{4}$
Question: Find current is $20 \Omega$ resistance


Answer: 0.4

## Solution:

Question: Assertion : Phase diff of two lightwave changes when 2 medium of same thickness and different RI are used.
Reason : Wavelength of wave depends upon RI of medium.

## Options:

(a) Assertion is true reason is true and reason is correct explanation of assertion
(b) Assertion is true reason is true but reason is not correct explanation of assertion
(c) Assertion is true reason is false
(d) Assertion is false reason is true

Answer: (a)

## Solution:

Question: Find peak value of current


Answer: 2.00

## Solution:

$c=150 \mu F$
$V_{0}=36$
$\omega=120 \pi$
$i_{0}=\frac{V_{0}}{\chi_{C}}=\frac{V_{0}}{1 / \omega C}=\left(\omega C V_{0}\right)$
$i_{0}=120 \pi \times 150 \times 10^{-6} \times 36=2.03$
$i_{0}=2 \mathrm{~A}$

Question: The temperature of an ideal gas is increased from 200 K to 800 K . If r.m.s. Speed of gas at 200 K is $\mathrm{V}_{0}$, r.m.s. Speed at 800 K is

## Options:

(a) $\mathrm{v}_{0} / 2$
(b) $\mathrm{V}_{0}$
(c) $4 v_{0}$
(d) $2 \mathrm{v}_{0}$

## Answer: (d)

## Solution:

Question: S1: Diffusion current is more in magnitude than drift in forward bias S 2 : Diffusion current is from n to p is forward bias


Options:
(a) S1-True, S2-False
(b) S1 - false, S2 - False
(c) S1-True, S2 - True
(d) S1 - false, S2 - True

Answer: (a)
Solution:

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## Chemistry

Question: Which of the following is not included in electrolysis of brine solutions?

## Options:

(a) NaOH
(b) $\mathrm{H}_{2}$
(c) HCl
(d) $\mathrm{Cl}_{2}$

Answer: (c)
Solution: When electricity is passed through brine solution it forms $\mathrm{NaOH} \mathrm{Cl}_{2}$ at anode $\mathrm{H}_{2}$ at cathode.

Question: Nessler reagent doesn't have
Options:
(a) K
(b) N
(c) Hg
(d) I

Answer: (b)
Solution: $\mathrm{K}_{2}\left[\mathrm{HgI}_{4}\right] \Rightarrow$ Nessler reagent

Question: Statement-1: Morphine and many of its homologues, when administered in medicinal doses, relieve pain and produce sleep
Statement-2: Morphine narcotics are sometimes referred to as opiates, since they are obtained from the opium poppy.

## Options:

(a) Statement-1 is true but Statement-2 is false.
(b) Statement-1 is false but Statement-2 is true.
(c) Both Statement-1 and Statement-2 are true.
(d) Both Statement-1 and Statement-2 are false.

Answer: (c)
Solution: Narcotic analgesics: Morphine and many of its homologues, when administered in medicinal doses, relieve pain and produce sleep. In poisonous doses, these produce stupor, coma, convulsions and ultimately death. Morphine narcotics are sometimes referred to as opiates, since they are obtained from the opium poppy.

These analgesics are chiefly used for the relief of postoperative pain, cardiac pain and pains of terminal cancer, and in childbirth.

Question: Which one doesn't gives $\mathrm{Pb}^{+2}$ test?
Options:
(a) Iodite
(b) Chromate
(c) Sulfate
(d) Nitrate

Answer: (d)
Solution:
$\mathrm{Pb}^{+2}+2 \mathrm{KI} \rightarrow \underset{\text { dark yellow }}{\mathrm{PbI}_{2}}$
$\mathrm{Pb}^{+2}+\mathrm{H}_{2} \mathrm{SO}_{4} \rightarrow \mathrm{PbSO}_{4}+\mathrm{H}^{+}$
$\mathrm{Pb}^{+2}+\mathrm{Na}_{2} \mathrm{SO}_{4} \rightarrow \mathrm{PbSO}_{4}+\mathrm{Na}^{+}$
$\mathrm{Pb}^{+2}+\mathrm{K}_{2} \mathrm{CrO}_{4} \rightarrow \mathrm{PbSO}_{4}+\mathrm{K}^{+}$

Question: Medium used in This reactions
$8 \mathrm{MnO}_{4}^{-}+3 \mathrm{~S}_{2} \mathrm{O}_{3}{ }^{2-}+\mathrm{H}_{2} \mathrm{O} \rightarrow 8 \mathrm{MnO}_{2}+6 \mathrm{SO}_{4}{ }^{2-}+2 \mathrm{OH}^{-}$
Options:
(a) Neutral or faintly alkaline solutions
(b) Acidic
(c) Basic
(d) None of these

Answer: (a)
Solution: In neutral or faintly alkaline solutions:
Thiosulphate is oxidised almost quantitatively to sulphate
$8 \mathrm{MnO}_{4}^{-}+3 \mathrm{~S}_{2} \mathrm{O}_{3}{ }^{2-}+\mathrm{H}_{2} \mathrm{O} \rightarrow 8 \mathrm{MnO}_{2}+6 \mathrm{SO}_{4}{ }^{2-}+2 \mathrm{OH}^{-}$

Question: If $\mathrm{a}_{0}$ is the radius of H atom, de-Broglie wavelength of electron in $3^{\text {rd }}$ orbit of $\mathrm{Li}^{2+}$ ion is $x \pi a_{0}$. Find out $x$.
Options:
(a) 1
(b) 2
(c) 3
(d) 4

Answer: (b)

## Solution:

$\lambda=\frac{\mathrm{h}}{\mathrm{MV}}$
$\mathrm{MVr}=\frac{\mathrm{nh}}{2 \pi} \Rightarrow \frac{\mathrm{~h}}{\mathrm{MV}}$
$\lambda=\frac{2 \pi \mathrm{r}}{\mathrm{n}}$
$r=a_{0} \times \frac{n^{2}}{Z}$
$\lambda=\frac{2 \pi}{\mathrm{n}} \mathrm{a}_{0} \times \frac{\mathrm{n}^{2}}{\mathrm{Z}}$
$=2 \pi \mathrm{a}_{0} \times \frac{\mathrm{n}}{\mathrm{Z}}$
$=2 \pi \mathrm{a}_{0} \frac{3}{3}$
$=2 \pi \mathrm{a}_{0}$
$x=2$

Question: Which of the following has highest hydration energy?
Options:
(a) $\mathrm{Be}^{+2}$
(b) $\mathrm{Mg}^{+2}$
(c) $\mathrm{Ca}^{+2}$
(d) $\mathrm{Ba}^{+2}$

Answer: (a)
Solution:
H.E $\propto Z_{+} \times Z_{-}$
H.E $\propto \frac{1}{\mathrm{r}_{(+)}}+\frac{1}{\mathrm{r}_{(-)}}$

Increase in ionic size down the group.
$\mathrm{Be}^{2+}>\mathrm{Mg}^{2+}>\mathrm{Ca}^{2+}>\mathrm{Sr}^{2+}>\mathrm{Ba}^{2+}$

Question: IUPAC name of the compound $\mathrm{K}_{3}\left[\mathrm{Co}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{3}\right]$ is
Options:
(a) Potassium trioxalatocobaltate (III)
(b) Potassium trioxalatocobaltate (II)
(c) Potassium cobalttrioxalate (II)
(d) Potassium oxalatocobaltate (III)

Answer: (a)
Solution: IUPAC name of the compound $\mathrm{K}_{3}\left[\mathrm{Co}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{3}\right]$ is Potassium trioxalatocobaltate (III)

Question: Most basic Oxide
Options:
(a) $\mathrm{Tl}_{2} \mathrm{O}$
(b) $\mathrm{Al}_{2} \mathrm{O}_{3}$
(c) $\mathrm{B}_{2} \mathrm{O}_{3}$
(d) $\mathrm{Cr}_{2} \mathrm{O}_{3}$

Answer: (a)
Solution:
$\mathrm{Tl}_{2} \mathrm{O} \Rightarrow$ Basic
$\mathrm{Al}_{2} \mathrm{O}_{3} \Rightarrow$ Amphoteric
$\mathrm{B}_{2} \mathrm{O}_{3} \Rightarrow$ Acidic
$\mathrm{Cr}_{2} \mathrm{O}_{3} \Rightarrow$ Amphoteric

Question: Assertion: $\left[\mathrm{Ni}(\mathrm{CO})_{4}\right] \&\left[\mathrm{Fe}(\mathrm{CO})_{5}\right]$ have zero oxidation state.
Reason: Low OS is observed due to presence of $\pi$-donor ligand in addition to sigma bonding. Options:
(a) Both Assertion and Reason are true and Reason is the correct explanation of Assertion.
(b) Both A and R are true, but Reason is not the correct explanation of Assertion.
(c) Assertion is true, but Reason is false.
(d) Assertion is false, but Reason is true.

Answer: (c)
Solution: Assertion is true, and reason is false

Question: Statement-1: Number of transition state is 3
Statement-2: Number of intermediate is 2
Statement-3: Reaction is endothermic
Number of correct statements are:


## Options:

(a) 1
(b) 2
(c) 3
(d) 0

Answer: (c)
Solution: All the given statements are correct.

Question: Matching of Amino acids with letter code

| Column I (Amino acid) | Column-II (Letter code) |
| :--- | :--- |
| (A) Alanine | (P) N |
| (B) Asparagine | (Q) A |
| (C) Aspartic acid | (R) R |
| (D) Arginine | (S) D |

## Options:

(a) $\mathrm{A}-\mathrm{P} ; \mathrm{B}-\mathrm{R} ; \mathrm{C}-\mathrm{Q} ; \mathrm{D}-\mathrm{S}$
(b) A-Q;B-P;C-R;D-S
(c) $\mathrm{A}-\mathrm{S} ; \mathrm{B}-\mathrm{P} ; \mathrm{C}-\mathrm{Q} ; \mathrm{D}-\mathrm{R}$
(d) $\mathrm{A}-\mathrm{Q} ; \mathrm{B}-\mathrm{P} ; \mathrm{C}-\mathrm{S} ; \mathrm{D}-\mathrm{R}$

Answer: (d)
Solution: A - Q ; B - P; C - S ; D - R

Question: Most Acidic compound among the following is
(a)

(b)

(c)

(d)


## Answer: (b)

Solution:


Because of more -I

Question: How many of these are pesticides?
(i) D.D.T. (ii) Aldrin (iii) Sodium arsenite (iv) Sodium chlorate

## Options:

(a) 1
(b) 2
(c) 3
(d) 4

Answer: (b)
Solution: DDT and Aldrin are the pesticides

Question: Number of Square Planar Species is/are :
$\mathrm{XeF}_{4}, \mathrm{SiF}_{4}, \mathrm{SF}_{4}, \mathrm{~B}(\mathrm{OH})_{4} \Theta,\left[\mathrm{Co}\left(\mathrm{NH}_{3}\right)_{6}\right]^{+2},\left[\mathrm{CoCl}_{4}\right]^{2-}$
Answer: 1

## Solution:

$\mathrm{XeF}_{4} \Rightarrow$ Hybridization is $\mathrm{sp}^{3} \mathrm{~d}^{2}$


Sq. Planar
$\mathrm{SiF}_{4} \Rightarrow$ Hybridization is $\mathrm{sp}^{3}$


Tetrahedral
$\mathrm{SF}_{4} \Rightarrow$ Hybridization is $\mathrm{sp}^{3} \mathrm{~d}$


Sea saw shape
$\mathrm{B}(\mathrm{OH})_{4} \Theta \Rightarrow$ Hybridization is $\mathrm{sp}^{3}$


Tetrahedral
$\left[\mathrm{Co}\left(\mathrm{NH}_{3}\right)_{6}\right]^{+2} \Rightarrow$ Hybridization is $\mathrm{sp}^{3} \mathrm{~d}^{2}$
Octahedral
$\left[\mathrm{CoCl}_{4}\right]^{2-} \Rightarrow$ Hybridization is $\mathrm{sp}^{3}$
Tetrahedral

Question: Number of aromatic amines having formula $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{~N}$ which can be synthesized by gabriel phthalimide synthesis
Answer: 3
Solution:




3 possibilities

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## Mathematics

Question: How many 4 letter word can be made from UNIVERSE, such that it has two vowels and two consonants.

## Answer: 504.00

## Solution:

UNIVERSE
Vowels: U, I, E, E
Consonants: N, V, R, S
${ }^{3} C_{2} \times{ }^{4} C_{2} \times 4!+{ }^{4} C_{2} \times \frac{4!}{2!}$
$=18 \times 24+6 \times 12$
$=72[6+1]$
$=504$

Question: If all the letters of the word PUBLIC are arranged in dictionary, then the rank of PUBLIC is
Answer: 582.00
Solution:
PUBLIC
B, C, I, L, P, U
$4 \times 5!+4 \times 4!+2 \times 2!+1+1$
$=480+96+4+2$
$=582$

Question: $\tan 9^{\circ}-\tan 27^{\circ}-\tan 63^{\circ}+\tan 81^{\circ}=$ $\qquad$
Answer: 4.00

## Solution:

$\tan 9^{\circ}-\tan 27^{\circ}-\tan 63^{\circ}+\tan 81^{\circ}$
$=\tan 9^{\circ}+\cot 9^{\circ}-\left[\tan 27^{\circ}+\cot 27^{\circ}\right]$
$=\frac{2}{\sin 18^{\circ}}-\frac{2}{\sin 54^{\circ}}$
$=2\left[\frac{1}{\frac{\sqrt{5}-1}{4}}-\frac{1}{\frac{\sqrt{5}+1}{4}}\right]$
$=8\left[\frac{2}{4}\right]$
$=4$

Question: $1^{2}-2^{2}+3^{2}-4^{2}+\ldots .(2022)^{2}+(2023)^{2}=$ $\qquad$
Answer: $2023 \times 2012$

## Solution:

$(1-2)(1+2)+(3-4)(3+4)+\ldots+(2021-2022)(2021+2022)+(2023)^{2}$
$-(1+2+3+4+\ldots .+2021+2022)+(2023)^{2}$
$-\frac{2022 \times 2023}{2}+(2023)^{2}$
(2023)(1012)

Question: If $f(x)+f(\pi-x)=\pi^{2}$, then $\int_{0}^{\pi} f(x) \times \sin x d x=$ $\qquad$
Answer: $\pi^{2}$
Solution:
$\int_{0}^{\pi} f(x) \times \sin x d x$
$2 I=\int_{0}^{\pi}(f(x)+f(\pi-x)) \sin x d x$
$2 I=\pi^{2} \times 2$
$I=\pi^{2}$
Question: $x^{2}+y^{2}-2 x+y=5$. Two tangents are drawn from $\left(\frac{9}{4}, 2\right)$. AB is chord of contact. Find area of $\triangle P A B$.
Answer: $\frac{5}{8}$

## Solution:

$$
x^{2}+y^{2}-2 x+y-5=0
$$



$$
\begin{aligned}
& \text { Area of } \triangle P A B=\frac{R L^{3}}{R^{2}+L^{2}} \\
& \begin{aligned}
R & =\sqrt{g^{2}+f^{2}-c} \\
& =\sqrt{1+\frac{1}{4}-(-5)} \\
& =\sqrt{\frac{25}{4}} \\
& =\frac{5}{2} \\
L & =\sqrt{S_{1}} \\
= & \sqrt{\frac{81}{16}+4-\frac{9}{2}+2-5} \\
= & \sqrt{\frac{81}{16}-\frac{9}{2}+1} \\
= & \sqrt{\frac{81-72+16}{16}} \\
= & \sqrt{\frac{25}{16}} \\
= & \frac{5}{4}
\end{aligned}
\end{aligned}
$$

Area of $\triangle P A B=\frac{\frac{5}{2} \times\left(\frac{5}{4}\right)^{3}}{\left(\frac{5}{2}\right)^{2}+\left(\frac{5}{4}\right)^{2}}=\frac{\frac{5^{4}}{2 \times 2^{6}}}{5^{2}}\left[\frac{1}{4}+\frac{1}{16}\right]=\frac{5}{8}$

Question: 3 dice are thrown. Find probability that none of them shows the same number.
Answer: $\frac{5}{9}$

## Solution:

$$
\frac{{ }^{6} C_{3} \times 3!}{216}=\frac{20}{36}=\frac{5}{9}
$$

Question: Find area bounded by the curve $y=|x-1|+|x-2|$ and the line $y=3$
Answer: 4.00

## Solution:


$A=\frac{1}{2}(1+3) \times 2=4$

Question: $20^{19}+2 \times 20^{18} \times 21+3 \times 20^{17} \times 21^{2}+\ldots+20 \times 21^{19}=k \times 20^{19}$. Find $k$.
Answer: 400.00

## Solution:

$$
\begin{aligned}
& 20^{19}+2 \times 20^{18} \times 21+3 \times 20^{17} \times 21^{2}+\ldots+20 \times 21^{19} \\
& k=1+2 \times\left(\frac{21}{20}\right)^{1}+3 \times\left(\frac{21}{20}\right)^{2}+\ldots .20\left(\frac{21}{20}\right)^{19} \\
& \frac{21}{20} k=1 \times\left(\frac{21}{20}\right)^{2}+2 \times\left(\frac{21}{20}\right)^{2}+\ldots .+\quad+20\left(\frac{21}{20}\right)^{20} \\
& =20\left(\frac{21}{20}\right)^{20}-\left[1+\frac{21}{20}+\ldots+\left(\frac{21}{20}\right)^{19}\right] \\
& =20 \times\left(\frac{21}{20}\right)^{20}-\left[\frac{\left(\frac{21}{20}\right)^{20}-1}{\frac{21}{20}-1}\right] \\
& k=400
\end{aligned}
$$

Question: Eccentricity of $\frac{x^{2}}{a^{2}}+\frac{y^{2}}{b^{2}}=1$ is $e_{1} \&$ eccentricity of $2 x^{2}-2 y^{2}=1$ is $e_{2}$ such that $e_{1}=\frac{1}{e_{2}}$. Ellipse intersect Hyperbola orthogonally. Find (LR) ${ }^{2}$ of ellipse.

## Answer: 2.00

## Solution:

They are confocal foci at $(1,0) \&(-1,0)$
$e_{1}=\frac{1}{\sqrt{2}}$
$e_{2}=\sqrt{2}$
$a e_{1}=1 \Rightarrow a=\sqrt{2}$
$(L R)^{2}=\left(\frac{2 b^{2}}{a}\right)^{2}=\left[2 a\left(1-e^{2}\right)\right]^{2}$
$=8 \times \frac{1}{4}=2$

Question: Solve: $(1+\ln x) \frac{d x}{d y}-x \ln x=e^{y}$
Answer: ()

## Solution:

$(1+\ln x) \frac{d x}{d y}-x \ln x=e^{y}$
$e^{-y}(1+\ln x) d x-e^{-y} x \ln x d y=d y$
$d\left[e^{-y}(x \ln x)\right]=d y$
$e^{-y} x \ln x=y+C$

Question: Find the square of the distance of $(12,12,18)$ from the plane containing the line of intersection of $\vec{r} \cdot(\hat{i}+\hat{j}+\hat{k})=6 \& \vec{r} \cdot(2 \hat{i}+3 \hat{j}+4 \hat{k})=-5$ and passing through $(0,2,-2)$.

## Answer: 620.00

## Solution:

The planes are
$x+y+z-6=0$
$2 x+3 y+4 z+5=0$
Required plane $(x+y+z-6)+\lambda(2 x+3 y+4 z+5)=0$
$(0,2,-2) \Rightarrow-6+\lambda(3)=0$
$\lambda=2$
$\therefore$ Plane is
$5 x+7 y+9 z+4=0$

Square of distance $=\left|\frac{5(12)+7(12)+9(18)+4}{\sqrt{5^{2}+7^{2}+9^{2}}}\right|^{2}=620$

Question: Sum of all values of $\alpha$ for which $\hat{i}-2 \hat{j}+3 \hat{k}, 2 \hat{i}-3 \hat{j}+4 \hat{k},(\alpha+1) \hat{i}+2 \hat{k}$ and $9 \hat{i}+(\alpha-8) \hat{j}+6 \hat{k}$ are coplanar

## Options:

(a) 6
(b) 2
(c) -2
(d) 4

Answer: (b)
Solution:
$\vec{a}=\hat{i}-2 \hat{j}+3 \hat{k}$
$\vec{b}=2 \hat{i}-3 \hat{j}+4 \hat{k}$
$\vec{c}=(\alpha+1) \hat{i}+2 \hat{k}$
$\vec{d}=9 \hat{i}+(\alpha-8) \hat{j}+6 \hat{k}$
$\overrightarrow{A B}=\hat{i}-\hat{j}+\hat{k}$
$\overrightarrow{A C}=\alpha \hat{i}+2 \hat{j}-\hat{k}$
$\overrightarrow{A B}=8 \hat{i}+(\alpha-6) \hat{j}+3 \hat{k}$
$\left|\begin{array}{ccc}1 & -1 & 1 \\ \alpha & 2 & -1 \\ 8 & \alpha-6 & 3\end{array}\right|=(6+\alpha-6)+(3 \alpha+8)+\left(\alpha^{2}-6 \alpha-16\right)=0$
$\Rightarrow \alpha^{2}-2 \alpha-8=0$

Question: $(p \Rightarrow q) \vee(\sim p \wedge q)$ : tautology
$(q \Rightarrow p) \Rightarrow(\sim p \wedge q)$ : contradiction

## Options:

(a) Both are true
(b) Neither true
(c) Only first one is true
(d) Only second is true

Answer: (b)

## Solution:

$(p \Rightarrow q) \vee(\sim p \wedge q)$
$(\sim p \vee q) \vee(\sim p \wedge q)$


Not universal set
Hence not tautology
$(q \Rightarrow p) \Rightarrow(\sim p \wedge q)$
$(\sim q \vee p) \Rightarrow(\sim p \wedge q)$
$a \Rightarrow \sim a$
$\sim a \vee \sim a$
Not contradiction

Question: Parallelopiped with coinitial edges $\vec{a}, \vec{b}$ and $\vec{c}$ is V . Then volume of parallelopiped with coinitial edges $\vec{a}, \vec{b}+\vec{c}, \vec{a}+2 \vec{b}+3 \vec{c}$.
Answer: V

## Solution:

$V=\left[\begin{array}{lll}\vec{a} & \vec{b} & \vec{c}\end{array}\right]$
$V^{\prime}=\left[\begin{array}{lll}\vec{a} & \vec{b}+\vec{c} & \vec{a}+2 \vec{b}+3 \vec{c}\end{array}\right]$
$=\left[\begin{array}{lll}\vec{a} & \vec{b} & \vec{a}+2 \vec{b}+3 \vec{c}\end{array}\right]+\left[\begin{array}{lll}\vec{a} & \vec{c} & \vec{a}+2 \vec{b}+3 \vec{c}\end{array}\right]$
$=\left[\begin{array}{lll}\vec{a} & \vec{b} & 3 \vec{c}\end{array}\right]+\left[\begin{array}{lll}\vec{a} & \vec{c} & 2 \vec{b}\end{array}\right]$
$=3\left[\begin{array}{lll}\vec{a} & \vec{b} & \vec{c}\end{array}\right]-2\left[\begin{array}{lll}\vec{a} & \vec{b} & \vec{c}\end{array}\right]$
$=\left[\begin{array}{lll}\vec{a} & \vec{b} & \vec{c}\end{array}\right]=V$

Question: Coefficient of $x^{-7}$ in $\left(a x-\frac{1}{3 b x^{2}}\right)^{11}$ is equal to coefficient of $x^{7}$ in $\left(a x^{2}+\frac{1}{2 b x}\right)^{11}$.
Find relation in $a \& b$.

## Options:

(a) $243 a b=64$
(b) $32 a b=729$
(c) $64 a b=243$
(d) $729 a b=32$

Answer: (d)

## Solution:

$x^{7} \rightarrow\left(a x^{2}+\frac{1}{2 b x}\right)^{11}$
$T_{r+1}={ }^{11} C_{r}\left(a x^{2}\right)^{11-r}\left(\frac{1}{2 b x}\right)^{r}$
$={ }^{11} C_{r} \frac{a^{11-r}}{(2 b)^{r}} x^{22-3 r}$
$22-3 r=7$
$\Rightarrow r=5$
$\frac{{ }^{11} C_{5} a^{6}}{32 b^{5}}$
$x^{-7} \rightarrow\left(a x-\frac{1}{3 b x^{2}}\right)^{11}$
$r=\frac{n p-m}{p+q}=\frac{11(1)-(-7)}{1+2}=6$
$T_{7}={ }^{11} C_{6}(a)^{11-6}\left(\frac{-1}{3 b}\right)^{6}$
$=\frac{{ }^{11} C_{6} a^{5}}{729 b^{6}}$
Equalising (1) and (2) we will get
$729 a b=32$

Question: $\lim _{n \rightarrow \infty}\left(2^{\frac{1}{2}}-2^{\frac{1}{3}}\right)\left(2^{\frac{1}{2}}-2^{\frac{1}{5}}\right) \times \ldots . \times\left(2^{\frac{1}{2}}-2^{\frac{1}{2 n+1}}\right)=$ ?
Answer: 0.00

## Solution:

Zero

Question: The system of the equations: $x+y+z=6 ; x+2 y+\alpha z=5$ and $x+2 y+6 z=\beta$ has
Options:
(a) infinitely many solutions for $\alpha=6, \beta=3$
(b) infinitely many solutions for $\alpha=6, \beta=5$
(c) unique solution for $\alpha=6, \beta=5$
(d) No solution for $\alpha=6, \beta=5$

Answer: (b)
Solution:
$x+y+z=6$
$x+2 y+\alpha z=5$
$x+2 y+6 z=\beta$
$\Delta=\left|\begin{array}{ccc}1 & 1 & 1 \\ 1 & 2 & \alpha \\ 1 & 2 & 6\end{array}\right|=0$
$\Rightarrow 12-2 \alpha-1(6-\alpha)+0=0$
$6-\alpha=0$
$\alpha=6$
$\Delta_{x}=\left|\begin{array}{lll}6 & 1 & 1 \\ 5 & 2 & 6 \\ \beta & 2 & 6\end{array}\right|$
$6(0)-1(30-6 \beta)+(10-2 \beta)=0$
$4 \beta-20=0$
$\beta=5$
Substituting in given system of equations we get,
$x+y+z=6$
$x+2 y+6 z=5$
Which are non parallel planes
Hence infinitely many solutions.

Question: $S(1)=(2002)^{2023}-(1919)^{2002}$ is divisible by 8 .
$S(2): 13 \times 13^{n}-12 n-13$ is divisible by 144 (where $n \in N$ ), then

## Options:

(a) $S(1)$ and $S(2)$ both are true
(b) Only $S(1)$ is true
(c) Only $S(2)$ is true
(d) Neither $S(1)$ and $S(2)$ are true.

Answer: (c)

## Solution:

## Statement 1:

$\because(2002)^{2023}=8 m$
$\because(2002)^{2023}$ is divisible by 8 .
Also, $(1919)^{2002}$ is not divisible by 8 .
$\therefore(2002)^{2023}-(1919)^{2002}$ is not divisible by 8 .

## Statement 2:

$13 \times 13^{n}-12 n-13$

$$
\begin{aligned}
& =13(1+12)^{n}-12 n-13 \\
& =13\left(1+12 n+{ }^{n} C_{2} 12^{2}+\ldots\right)-12 n-13 \\
& =144 n+144^{n} C_{2}+\ldots \\
& =144\left(n+{ }^{n} C_{2}+\ldots\right) \\
& =144 k
\end{aligned}
$$

