

NKT/KS/17/5082

Bachelor of Science (B.Sc.) Semester-II (CBS) Examination

CH-201 : CHEMISTRY

Compulsory Paper—1

(Organic Chemistry)

Time : Three Hours]

[Maximum Marks : 50

N.B. :— (1) All **FIVE** questions are compulsory and carry equal marks.

(2) Write chemical equations and draw diagrams wherever necessary.

1. (A) What is hybridization ? Explain the structure of ethane on the basis of hybridization. 5
 (B) What are Reactive intermediates ? Explain the formation and stability of carbocations. 5

OR

- (C) Write a note on Inductive effect. 2½
 (D) Define free radicals. Discuss the stability of free radicals. 2½
 (E) Define the terms :
 (i) Bond length and
 (ii) Bond angle. 2½
 (F) Discuss with example Homolytic bond fission. 2½
2. (A) What is Resolution ? Describe various methods used for resolving racemic mixture. 5
 (B) What is conformation ? Discuss the conformations of ethane. 5

OR

- (C) Discuss the optical isomerism in Lactic acid. 2½
 (D) Write a note on geometrical isomerism in 2-Butene. 2½
 (E) State the sequence rules for assignment of RS-configuration. 2½
 (F) Write a note on 'Asymmetric synthesis.' 2½

3. (A) Discuss Baeyer's strain theory. What are the limitations of this theory ? 5
 (B) What is Markownikoff's rule ? Explain the ionic mechanism of addition of HBr to propene. 5

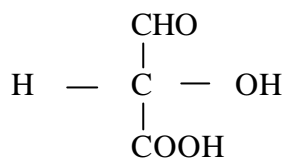
OR

- (C) Discuss the mechanism of chlorination of methane. 2½
 (D) How does ethene react with :
 (i) Ozone and
 (ii) Cold aqueous KMnO_4 ? 2½
 (E) Write a note on L.P.G. 2½
 (F) Give the preparation of propene from :
 (i) 1-Propanol and
 (ii) 2-Bromopropane. 2½
4. (A) What are Alkadienes ? How are they classified ? What is the action of :
 (i) Bromine and
 (ii) HBr on 1, 3-butadiene ? 5
 (B) What is aromatic electrophilic substitution reaction ? Discuss mechanism of nitration of benzene with energy profile diagram. 5

OR

- (C) Describe Diel's Alder reaction. 2½
 (D) Discuss molecular orbital structure of Benzene. 2½
 (E) Write a note on acidic nature of acetylene. 2½
 (F) Discuss the structure of benzene on the basis of resonance. 2½
5. Attempt any **TEN** of the following :
 (i) What is Hydrogen bonding ?
 (ii) Define the term nucleophile.
 (iii) Draw molecular orbital diagram of ethylene molecule.
 (iv) Define the term enantiomers.

(v) Write R-or-S configuration for the following :



(vi) Draw Newman's projection formulae of n-Butane.

(vii) State Saytzeff's rule

(viii) What is peroxide effect ?

(ix) Define Octane number.

(x) Give polymerisation reaction of Acetylene.

(xi) What is the action of Bromine on 1, 3-Butadiene ?

(xii) State Huckel's rule of aromaticity.

1×10=10