NRT/KS/19/2049

Bachelor of Science (B.Sc.) Semester—II Examination

CHEMISTRY (Organic Chemistry) (CH201)

Compulsory Paper—1

Tin	ne : Tl	nree Hours] [Maximum Marks	: 50
	Not	e:—(1) All FIVE questions are compulsory and carry equal marks.	
		(2) Write chemical equation and draw diagram wherever necessary.	
1.	(A)	What is hybridisation ? Explain the structure of Ethylene molecule on the basi hybridisation.	s of 5
	(B)	Discuss Homolytic and Heterolytic fission of covalent bond with suitable exmple.	5
		OR	
	(C)	Write a note on Inductive effect.	21/2
	(D)	Discuss hydrogen bonding with reference to alcohol.	21/2
	(E)	Define the terms with suitable example:	
		(i) Substitution reaction and	
		(ii) Elimination reaction.	21/2
	(F)	Classify the following as electrophiles and nucleophiles:	
		(i) $FeCl_3$ (ii) H_2O (iii) CH_3NH_2 (iv) BF_3 and (v) $AlCl_3$.	21/2
2.	(A)	What is Optical activity? Discuss optical isomerism in Tartaric acid and Lactic acid.	5
	(B)	Define conformation. Discuss conformational isomerism of n-butane.	5
		OR	
	(C)	Explain geometrical isomerism in Maleic and Fumaric acid.	21/2
	(D)	Write a note on Newman's projection and Sawhorse formulae of Ethane.	21/2
	(E)	Discuss D and L nomenclature with suitable example.	21/2
	(F)	What is Resolution? Explain biological method of resolution	21/2
3.	(A)	What is Peroxide effect? Explain free radical mechanism of addition of HBr to propene.	5
	(B)	Discuss Baeyer's strain theory. Give its limitations.	5
		OR	
	(C)	Discuss the mechanism of chlorination of methane molecule.	2½
	(D)	Write a note on L.P.G.	2½
	(E)	How will you prepare ethylene from:	
		(i) Ethyl alcohol and	
		(ii) Ethyl bromide ?	2½
	(F)	Discuss Ionic mechanism of addition Br ₂ to ethylene.	2½

4. (A) What are dienes? How they are classified?

What is the action of:

- (i) Bromine and
- (ii) HBr on 1, 3-butadiene?

5

(B) What is aromatic electrophilic substitution reaction? Discuss mechanism of nitration of benzene with energy profile diagram.

OR

(C) Explain acidic nature of ethyne.

 $2\frac{1}{2}$

(D) Discuss the Diels—Alder reaction.

2½

(E) Discuss molecular orbital structure of Benzene.

 $2\frac{1}{2}$

(F) What is Huckel rule? Explain aromaticity of cyclopentadienyl anion.

 $2\frac{1}{2}$

- 5. Solve any **TEN** of the following:—
 - (i) Define Carbonium ion.
 - (ii) Draw M.O. Structure of Ethane molecule.
 - (iii) Give one example of addition reaction.
 - (iv) What is Asymmetric synthesis?
 - (v) What is meant by stereogenic centre?
 - (vi) What is racemisation?
 - (vii) Draw structure and give IUPAC name of neopentane.
 - (viii) What is the action of O₃ on Ethylene?
 - (ix) Define Octane number.
 - (x) What is oxyacetylene flame?
 - (xi) Give the polymerisation reaction of acetylene.
 - (xii) State Huckel's rule of aromaticity.

 $1 \times 10 = 10$