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NJR/KS/18/3138

Bachelor of Science (B.Sc.) Semester—V (C.B.S.) Examination **QUANTUM MECHANICS, NANOMATERIALS AND NANOTECHNOLOGY**

Paper—2 (Physics)

Time: Three Hours]

[Maximum Marks: 50

N.B.:— (1) **ALL** questions are compulsory.

(2) Draw neat diagram wherever necessary.

Given :—

 $h = 6.6 \times 10^{-34} \text{ J.sec}$

 $c = 3 \times 10^8 \text{ m/s}$

 $m_{e} = 9.1 \times 10^{-31} \text{ kg}$

 $e = 1.60 \times 10^{-19} c$

EITHER

- (A) State de-Broglie's hypothesis for matter waves. Explain Davission and Germer experiment. 5
 - (B) (i) What is compton effect? On what factors do compton shift depend? 3
 - (ii) An X-ray photon of initial energy 90 keV under-goes compton scattering at angle of 60°, find energy of recoiled electron. 2

OR

- (C) Calculate the de-Broglie wavelength of an electron accelerated through a potential difference of 1.25 kV. $2\frac{1}{2}$
- (D) Show that $v_g = v_p \lambda \frac{dv_p}{d\lambda}$

(Symbols used have their usual meaning).

 $2\frac{1}{2}$

- (E) How the classical mechanics failed to explain black body radiation spectrum?
- $2\frac{1}{2}$ $2\frac{1}{2}$

(F) Give analytical treatment for the formation of wave packet.

EITHER

- (A) What is free particle? Obtain an expression for eigen energy and eigen wave function of 2. free particle in one dimensional box. 5
 - (B) (i) What is operator? Obtain an expression for momentum operator.

(ii) Show that the function $f(x) = \sin ax$ is eigen function corresponding to the operator

$$\frac{d^2}{dx^2}$$
.

OR

(C) Give physical significance of wave function ψ .

 $2\frac{1}{2}$

3

(D) Obtain Schrodinger's time dependent equation.

 $2\frac{1}{2}$

(E) What are the postulates of quantum mechanics?

- $2\frac{1}{2}$
- (F) Find the lowest energy of an electron confined to move in one-dimensional potential box of length 5Å. $2\frac{1}{2}$

	EITHER	
3.	(A) Explain top down approach and bottom up approach for the synthesis of nanomater	rials.
	(B) (i) What are physical and chemical properties of nanomaterials ?	3
	(ii) What would be the surface to volume ratio of a nanosphere of radius 2 nm?	2
	OR	
	(C) Differentiate nanomaterials from bulk materials.	21/2
	(D) What are quantum dots? Enlist their properties.	21/2
	(E) The surface to volume ratio of nanocubes is 0.5 per nanometer. What would be the length of nanocubes ?	side 2½
	(F) What are fullerences? State their properties.	21/2
	EITHER	
4.	 (A) What is transmission electron microscopy? Explain the working of TEM with the hence neat diagram. State its dis-advantages. (B) (i) Explain synthesis of nanoparticles by ball milling method. (ii) Calculate the FWHM XRD pattern of Ni with crystallite size of 20 nm diffrage. 	elp of 5
	(B) (i) Explain synthesis of nanoparticles by ball milling method.	3
		ecting
	angle of 38.2° with X-rays of wavelength 1.54 Å.	2
	OR AND THE PROPERTY OF THE PRO	
	(C) Explain sol-gel method for synthesis of nanomaterials.	21/2
	(D) What are the different methods for determination of size of nanoparticles? State Sche	
	formula with symbols.	2½
	(E) How nanotechnology is useful in home appliances ?	2½
	(F) X-rays of wavelength 1.54 Å are diffracted from nanoparticles at diffracting angle 27° full width at half maxima of 1°. Calculate the size of nanocrystals.	with 2½
5.	Attempt any ten:	
	(i) State Planck's quantum hypothesis.	
	(ii) Define wave packet. (iii) An electron has a speed of 300 m/sec accurate to 0.01% with what fundamental acc	
	(iii) An electron has a speed of 300 m/sec accurate to 0.01% with what fundamental acc can we locate the position of electron.	uracy
	(iv) Define expectation value.	
	(v) If wave function $\psi(x) = e^{3x}$, find eigen value for the operator $\frac{d}{dx}$.	
	(vi) What is probability density?	
	(vii) What is quantum well ?	
	(viii) State two examples of two dimensional nanomaterials.	
	(ix) If radius of quantum dot is 1.5 pm. Calculate surface to volume ratio.	
	(x) What are the limitations of SEM ?	
	(xi) What is aero gel?	
	(xii) Convert F.W.H.M. of 0.7 radian into degree.	1×10

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