

Time : Three Hours]

[Maximum Marks : 50

N.B. :— (1) All questions are compulsory.

(2) Draw neat diagrams wherever necessary.

1. EITHER

- (a) What is de-Broglie concept of matter waves ? Describe Davisson and Germer's experiment. 5
- (b) (i) State Planck's quantum hypothesis. Obtain Planck's radiation formula. 3
- (ii) Photons of energy 0.1 MeV undergo Compton scattering. Find the energy of photon a scattered at 60° .
($h = 6.63 \times 10^{-34}$ Js, $m_e = 9.1 \times 10^{-31}$ kg ; $C = 3 \times 10^8$ m/s). 2

OR

- (c) State Heisenberg's uncertainly principle. Explain its Physical significance. $2\frac{1}{2}$
- (d) Obtain the relation between group velocity and phase velocity. $2\frac{1}{2}$
- (e) The velocity of electron is 3×10^4 m/s measured with accuracy of 0.01%. Find the uncertainty in measurement of its position. ($m_e = 9.1 \times 10^{-31}$ kg, $h = 6.6 \times 10^{-34}$ J-s). $2\frac{1}{2}$
- (f) What is a wave packet ? Give the analytical treatment for the formation of a wave packet. $2\frac{1}{2}$

2. EITHER

- (a) Derive Schrodinger's time dependent equation. Give the physical significance of wavefunction. 5
- (b) (i) What is eigen function and eigen value of a state ? Explain with an example. 3
- (ii) Show that the function $f(x) = \cos 3x$ is an eigen function corresponding to the operator $\frac{d^2}{dx^2}$. 2

OR

- (c) Obtain an expression for momentum operator in quantum mechanics. $2\frac{1}{2}$
- (d) What are the conditions for well-behaved, wave function ? $2\frac{1}{2}$
- (e) Calculate first two energy levels for :
- (i) A $1 \mu\text{gm}$ dust particle moving in 1-D box of 0.1 mm,
- (ii) An electron confined to 1\AA . $2\frac{1}{2}$
- (f) What is degeneracy ? Calculate the degree of degeneracy for (231) state of particle in three dimensional box. $2\frac{1}{2}$

3. (a) What are nanomaterials and nano technology ? Explain Bottom up approach for manufacturing process of nanomaterials. 5
- (b) (i) What are quantum well quantum wire and quantum dots ? Give one example of each. 3
- (ii) Calculate the surface to volume ratio of a nanosphere of radius 2nm. 2

OR

- (c) Differentiated nanomaterials from the bulk materials. 2½
- (d) Describe top-down method for the synthesis of nano materials. Mention any two methods. 2½
- (e) What are the properties and applications of carbon nanotubes (CNTs) ? 2½
- (f) A block has a surface area of 6m^2 and volume of 1m^3 . Calculate the surface to volume ratio of the block system. 2½

4. EITHER

- (a) What are the different methods of synthesis of nanomaterials ? Explain wet chemical method for synthesis of nanomaterials. 5
- (b) (i) Explain the formation of image in SEM. 3
- (ii) Particles are diffracted by X-rays of wavelength 1.54\AA at diffracting angle 27° with FWHM of 0.5° . Determine the crystallite size of the particle. 2

OR

- (c) What are the advantages of TEM over SEM ? 2½
- (d) Explain the applications of nano technology in drug delivery and in medicine. 2½
- (e) Explain how particle size can be determined by BET method. 2½
- (f) Calculate the interplanar spacing of nanocrystalline material of crystallite size 70nm and FWHM 2° . Given : Wavelength of Cu K_α X-ray is 1.54\AA . (Take only first order) 2½

Attempt any TEN of the following :

- (i) Mention two failures of Classical theory.
- (ii) What is Compton effect ?
- (iii) Calculate de-Broglie's wavelength for an electron moving with momentum $6.6 \times 10^{-24} \text{ kg m/sec}$.
- (iv) State Schrodinger's time independent equation for free particle.
- (v) What is an expectation value of a dynamical variable ?
- (vi) A wave function is given by $\psi(x) = e^x$. Find the eigen value for the operator $\frac{d}{dx}$.
- (vii) State any two applications of nanowires.
- (viii) Why does the melting point of nanoparticles decrease with the decrease in size of nano particles ?
- (ix) Why is the surface/volume ratio very large for nanoparticles compared to bulk materials ?
- (x) What are the applications of nano electronics ?
- (xi) Why are nanoparticles highly reactive ?
- (xii) Express 0.8° in radians. 1×10