Sol.

$\mathrm{NH}_{3}$ has greater dipole moment than $\mathrm{NF}_{3}$
6.

(1)

(2)

(3)

(4)


## Answer (2)

Sol. This is an example of Clemmensen reduction reaction. In this reaction carbonyl group is reduced to methylene group.
7. Which of the following is the correct order of first ionization enthalpy?
(1) $\mathrm{Be}<\mathrm{B}<\mathrm{O}<\mathrm{F}<\mathrm{N}$
(2) $\mathrm{B}<\mathrm{Be}<\mathrm{O}<\mathrm{N}<\mathrm{F}$
(3) $\mathrm{B}<\mathrm{Be}<\mathrm{N}<\mathrm{F}<\mathrm{O}$
(4) $\mathrm{Be}<\mathrm{B}<\mathrm{N}<\mathrm{O}<\mathrm{F}$

Answer (2)
Sol. Be has more value of first ionization enthalpy than $B$ due to fully filled configuration and $N$ has more value of first ionization enthalpy than O due to half filled configuration
The correct order is $\mathrm{B}<\mathrm{Be}<\mathrm{O}<\mathrm{N}<\mathrm{F}$
8. Statement-1 : Aldol condensation is caused by acidity of $\alpha$ hydrogen
Statement-2 : Cross aldol is not possible between

(1) Both statement- 1 and statement- 2 are correct
(2) Both statement-1 and statement-2 are incorrect
(3) Statement-1 is correct but statement-2 is incorrect
(4) Statement-1 is incorrect but statement-2 is correct

## Answer (3)

Sol. Aldol reaction is given by those carbonyl compounds which have at least one $\alpha$ hydrogen atom because $\alpha$-hydrogen of carbonyl compounds is acidic. Benzaldehyde and acetaldehyde will form cross aldol because acetaldehyde has $\alpha$-hydrogen atom.
9. Select the correct structure of L-glucose.
(1)

(2)


## Answer (2)

Sol.


D-Glucose


L-Glucose

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As per student response sheet and NTA answerker

10. Decreasing order of the field strength of the following ligands will be:
co, $\stackrel{\ominus}{\mathrm{CN}}, \stackrel{\ominus}{\mathrm{Cl}, \mathrm{H}_{2} \mathrm{O}}$
(1) $\mathrm{CO}>\stackrel{\ominus}{\mathrm{CN}}>\mathrm{H}_{2} \mathrm{O}>\mathrm{Cl}^{-}$
(2) $\mathrm{CO}>\stackrel{\ominus}{\mathrm{CN}}>\mathrm{Cl}^{-}>\mathrm{H}_{2} \mathrm{O}$
(3) $\stackrel{\ominus}{\mathrm{CN}}>\mathrm{CO}>\mathrm{H}_{2} \mathrm{O}>\mathrm{Cl}^{-}$
(4) $\stackrel{\ominus}{\mathrm{CN}}>\mathrm{CO}>\mathrm{Cl}^{-}>\mathrm{H}_{2} \mathrm{O}$

Answer (1)
Sol. $\mathrm{CO}>\stackrel{\ominus}{\mathrm{C}} \mathrm{N}>\mathrm{H}_{2} \mathrm{O}>\stackrel{\ominus}{\mathrm{C}}$
11. Calculate the molarity of NaCl solution, if 5.85 gm of NaCl is dissolved in 500 ml of solution.
(1) 0.1 M
(2) 0.2 M
(3) 0.32 M
(4) 0.4 M

Answer (2)
Sol. Molarity $=\frac{\text { Number of moles of solute }}{\text { Volume of solution (in L) }}$

$$
=\frac{5.85 \times 1000}{58.5 \times 500}=0.1 \times 2=0.2 \mathrm{M}
$$

12. Which of the following does not give Lassaigne's test?
(1) Urea
(2) Azobenzene
(3) Hydrazine
(4) Phenylhydrazine

Answer (3)
Sol. Hydrazine $\left(\mathrm{NH}_{2}-\mathrm{NH}_{2}\right)$ does not contain carbon. On fusion with sodium metal, it cannot form NaCN . So hydrazine does not show Lassaigne's test.
13. Among the following, species that have one unpaired $e^{\ominus}$ ?
(1) $\mathrm{CN}^{\ominus}$
(2) $\mathrm{O}_{2}^{2-}$
(3) $\mathrm{O}_{2}^{+}$
(4) $\mathrm{NO}^{\ominus}$

Answer (3)
Sol. Unpaired $\mathrm{e}^{\ominus}$

$$
\begin{aligned}
& \mathrm{CN}^{\ominus} \rightarrow 14 \mathrm{e}^{\ominus} \rightarrow \text { zero } \\
& \mathrm{O}_{2}^{2-} \rightarrow 18 \mathrm{e}^{\ominus} \rightarrow \text { zero } \\
& \mathrm{O}_{2}^{+} \rightarrow 15 \mathrm{e}^{\ominus} \rightarrow \text { one } \\
& \mathrm{NO}^{\ominus} \rightarrow 16 \mathrm{e}^{\ominus} \rightarrow \text { two }
\end{aligned}
$$

14. For a given reaction


Relation between the molecules $P$ and $B$ are:
(1) Enantiomer
(2) Diastereomers
(3) Positional isomers
(4) Functional isomers

## Answer (3)

Sol. Positional isomers.

15. From the given data, find enthalpy of hydrogenation of ethene in $\mathrm{kJ} / \mathrm{mol}$
(a) B.E. of $\mathrm{C}-\mathrm{C}=350 \mathrm{~kJ} / \mathrm{mol}$
(b) B.E. of $\mathrm{C}=\mathrm{C}=600 \mathrm{~kJ} / \mathrm{mol}$
(c) B.E. of $\mathrm{H}-\mathrm{H}=400 \mathrm{~kJ} / \mathrm{mol}$
(d) B.E. of $\mathrm{C}-\mathrm{H}=410 \mathrm{~kJ} / \mathrm{mol}$
(1) -170
(2) -580
(3) +170
(4) +580

Answer (1)

Sol.
 $+\mathrm{H}-\mathrm{H}$

$\Delta_{r} \mathrm{H}=\Delta \mathrm{H}(\mathrm{C}=\mathrm{C})+\Delta \mathrm{H}(\mathrm{H}-\mathrm{H})-\Delta \mathrm{H}(\mathrm{C}-\mathrm{C})$

$$
\begin{align*}
& =600+400-350-2(410)  \tag{C-H}\\
& =-170 \mathrm{~kJ} / \mathrm{mol}
\end{align*}
$$

16. Find out wavelength of a photon having frequency equal to $900 \mathrm{sec}^{-1}$.
(1) $3.33 \times 10^{5} \mathrm{~m}$
(2) $3.33 \times 10^{5} \mathrm{~cm}$
(3) $3.33 \times 10^{7} \mathrm{~m}$
(4) $3.33 \times 10^{4} \mathrm{~m}$

Answer (1)

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Sol. $v=\frac{C}{\lambda}$
$\lambda=\frac{C}{v}$
$\lambda=\frac{3 \times 10^{8} \mathrm{msec}^{-1}}{900 \mathrm{sec}^{-1}}$
$=\frac{3 \times 10^{8}}{900}$
$=\frac{3 \times 10^{6}}{9}$
$=\frac{1}{3} \times 10^{6}$
$=0.333 \times 10^{6}$
$=3.33 \times 10^{5} \mathrm{~m}$
17. Why $\mathrm{NH}_{4} \mathrm{Cl}$ is added before $\mathrm{NH}_{4} \mathrm{OH}$ for the ppt. of $\mathrm{Fe}^{3+}$ ions?
(1) To decrease $\mathrm{OH}^{-}$ion concentration
(2) To increase $\mathrm{Cl}^{-}$ion concentration
(3) To increase $\mathrm{NH}_{4}^{+}$ion concentration
(4) To decrease $\mathrm{H}^{+}$ion concentration

Answer (1)
Sol. $\mathrm{NH}_{4} \mathrm{OH} \rightleftharpoons \mathrm{NH}_{4}^{+}+\mathrm{OH}^{-}$
$\mathrm{NH}_{4} \mathrm{Cl} \rightleftharpoons \mathrm{NH}_{4}^{+}+\mathrm{Cl}^{-}$
Solid $\mathrm{NH}_{4} \mathrm{Cl}$ is added to $\mathrm{NH}_{4} \mathrm{OH}$ solution to decrease the $\mathrm{OH}^{-}$ion concentration due to common ion effect.
18. Consider the following sequence of reactions and identify the unknown reagents (A) and (B) respectively.

(1) (A) :Dil. aq NaOH at $20^{\circ} \mathrm{C}$
(B) : $\mathrm{HBr}, \mathrm{CH}_{3} \mathrm{COOH}$
(2) (A) : Dil. aq NaOH at $20^{\circ} \mathrm{C}$
(B) : $\mathrm{Br}_{2}, \mathrm{CHCl}_{3}$
(3) (A) : Alc. NaOH at $80^{\circ} \mathrm{C}$ (B) : $\mathrm{HBr}, \mathrm{CH}_{3} \mathrm{COOH}$
(4) (A) : Alc. NaOH at $80^{\circ} \mathrm{C}$
(B) : $\mathrm{Br}_{2}, \mathrm{CHCl}_{3}$

Answer (3)
Sol.

19. Match the following

| (i) | Nitrobenzene | (a) | +R |
| :---: | :---: | :---: | :---: |
| (ii) | Aniline | (b) | -R |
| (iii) |  | (c) | +E |
| (iv) |  | (d) | -E |

(1) (i) $\rightarrow$ (b), (ii) $\rightarrow$ (a), (iii) $\rightarrow$ (c), (iv) $\rightarrow$ (d)
(2) (i) $\rightarrow$ (a), (ii) $\rightarrow$ (b), (iii) $\rightarrow$ (c), (iv) $\rightarrow$ (d)
(3) (i) $\rightarrow$ (c), (ii) $\rightarrow$ (b), (iii) $\rightarrow$ (a), (iv) $\rightarrow$ (d)
(4) (i) $\rightarrow$ (d), (ii) $\rightarrow$ (c), (iii) $\rightarrow$ (a), (iv) $\rightarrow$ (b)

Answer (1)
Sol. (i) $\rightarrow$ (b), (ii) $\rightarrow$ (a), (iii) $\rightarrow$ (c), (iv) $\rightarrow$ (d)
20. Which of the following is not possible major product?
(1)

(2)
$\mathrm{CH}_{3}-\left(\mathrm{CH}_{2}\right)_{2}-\mathrm{NH}_{2} \xrightarrow[H \mathrm{XX}]{\mathrm{NaNO}_{2}} \mathrm{CH}_{3}\left(\mathrm{CH}_{2}\right)_{2}-\mathrm{NO}_{2}+\mathrm{N}_{2}$
(3)

(4)


## Answer (2)

Sol. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$


## SECTION - B

Numerical Value Type Questions: This section contains 10 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.
21. How many of the following compounds are $s p^{3}$ hybridised?

$$
\mathrm{ClO}_{3}^{-}, \mathrm{ClO}_{2}^{-}, \mathrm{NH}_{3}, \mathrm{NO}_{2}
$$

## Answer (3)

Sol.

22. Total number of structural isomers possible for a compound with molecular formula $\mathrm{C}_{7} \mathrm{H}_{16}$ are:

## Answer (5)

Sol. $\mathrm{C}_{7} \mathrm{H}_{16}$ has $\mathrm{DoU}=0$

(i) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(ii) $\mathrm{CH}_{3}-\mathrm{C}_{\mathrm{CH}}^{\mathrm{CH}}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(iii)

(iv)

(v)

(vi)

(vii)

(viii)

(ix)

23. The de-Broglie wavelength of an electron in $4^{\text {th }}$ orbit of hydrogen atom is $\qquad$ $\pi \mathrm{a}_{0}$ ( $\mathrm{a}_{0}=$ Bohr radius).

Answer (8)
Sol. $\because \quad \lambda_{\text {de-Broglie }}=\frac{2 \pi r}{n}=\frac{2 \pi}{n} \times 0.529 \frac{n^{2}}{z} \AA$

$$
\text { or, } \begin{aligned}
\lambda_{\text {de-Broglie }} & =2 \pi \times \mathrm{n} \times \mathrm{a}_{0} \AA \\
& =2 \pi \times 4 \times \mathrm{a}_{0} \AA \\
& =8 \pi \mathrm{a}_{0} \AA
\end{aligned}
$$

24. 50 mL of $\mathrm{KMnO}_{4}$ solution is used for titration with 20 mL of 2 M oxalic acid solution in Acidic medium. The molarity of $\mathrm{KMnO}_{4}$ solution is $x \times 10^{-2} \mathrm{M}$. The value of $x$ is

Answer (32)

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Sol. $\underset{\substack{\text { n.f. }=5}}{\mathrm{MnO}_{4}^{\ominus}}(\mathrm{aq})+\underset{\substack{\mathrm{C}_{2} \mathrm{O}_{4}^{2-} \\ \text { n.f. }=2}}{\mathrm{H}^{+}}(\mathrm{aq}) \xrightarrow{\mathrm{Mn}^{2+}}+\mathrm{CO}_{2} \uparrow$
$5 \times \mathrm{M}_{\mathrm{KMNO}_{4}} \times 50=2 \times 20 \times 2$
$\mathrm{M}_{\mathrm{KMnO}}^{4}-1=\frac{8}{25}=32 \times 10^{-2} \mathrm{M}$
$x=32$
25. A solution having non-volatile solute in water shows elevation in boiling point of $2^{\circ} \mathrm{C}$. Find out vapour pressure of solution (in mm Hg ) (Nearest integer)
Vapour pressure of pure water $=760 \mathrm{~mm} \mathrm{Hg}$ $\mathrm{K}_{\mathrm{b}}$ of water $=0.52 \mathrm{~K} . \mathrm{kg} \mathrm{mole}^{-1}$
Answer (711)
Sol. $\Delta \mathrm{T}_{\mathrm{b}}=\left(\mathrm{K}_{\mathrm{b}}\right)(\mathrm{m})$
$2=(0.52)(m)$
$\mathrm{m}=3.846$
$X_{\text {Solute }}=\frac{m}{m+55.5}=0.0648$
$\frac{760-X}{760}=0.0648$
$\Rightarrow P_{\text {solution }}=710.74 \mathrm{~mm} \mathrm{Hg}$
$\approx 711 \mathrm{~mm} \mathrm{Hg}$
26. $\mathrm{MnO}_{2}+\mathrm{KOH}+\mathrm{O}_{2} \longrightarrow \mathrm{~A}$
' $A$ ' disproportionate into ' $B$ ' and ' $C$ '. Find the sum of magnetic moment (spin only) (in B.M.) of B and C (Nearest integer)

## Answer (4)

Sol. $2 \mathrm{MnO}_{2}+4 \mathrm{KOH}+\mathrm{O}_{2} \longrightarrow 2 \mathrm{~K}_{2} \mathrm{MnO}_{4}+2 \mathrm{H}_{2} \mathrm{O}$ (A)
$3 \mathrm{MnO}_{4}^{2-}+4 \mathrm{H}^{+} \xrightarrow{\text { Disproportionation }} 2 \mathrm{MnO}_{4}^{-}+\mathrm{MnO}_{2}+2 \mathrm{H}_{2} \mathrm{O}$
(A)

B and C are $\mathrm{MnO}_{4}^{-}$and $\mathrm{MnO}_{2}$
Mn in $\mathrm{MnO}_{2}$ has +4 oxidation state hence it has $(n-1) d^{3} n s^{0}$ electronic configuration unpaired e $=3$
Mag. moment : 3.87 B.M. by $\sqrt{n(n+2)}$
$\mathrm{KMnO}_{4} / \mathrm{MnO}_{4}^{-}$is diamagnetic hence magnetic moment $=0$ because it has no unpaired electron.
Hence, sum of mag. moment $=3.87$ B.M.
Nearest integer $=4$
27. How many of the following coordination compounds have even number of unpaired electrons?
$\left[\mathrm{V}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}, \quad\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}, \quad\left[\mathrm{Cu}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}$, $\left[\mathrm{Ni}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+},\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}$
Answer (3)
Sol. $\left[\mathrm{V}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+} \Rightarrow d^{2} s p^{3} \Rightarrow \mathrm{n}=3$
$\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+} \Rightarrow s p^{3} a^{2} \Rightarrow \mathrm{n}=4$
$\left[\mathrm{Cu}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+} \Rightarrow s p^{3} d^{2} \Rightarrow \mathrm{n}=1$
$\left[\mathrm{Ni}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+} \Rightarrow s p^{3} d^{R} \Rightarrow \mathrm{n}=2$
$\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+} \Rightarrow s p^{3} d^{2} \Rightarrow \mathrm{n}=4$
28. Consider the following reaction sequence :
$A \underset{k_{2}}{\stackrel{k_{1}}{\rightleftharpoons}} B \xrightarrow{\mathrm{k}_{3}} C$
Overall $k=\frac{k_{1} k_{2}}{k_{3}}$
if $\mathrm{E}_{\mathrm{a}_{1}}=300 \mathrm{~kJ} /$ mole

$$
\mathrm{E}_{\mathrm{a}_{2}}=200 \mathrm{~kJ} / \mathrm{mole}
$$

Overall, $\left(E_{a}\right)_{\text {eff }}=400 \mathrm{~kJ} / \mathrm{mole}$
Find out $\mathrm{E}_{\mathrm{a}_{3}}$ (in $\mathrm{kJ} /$ mole)

## Answer (100)

Sol. $\left(E_{a}\right)_{\text {eff }}=E_{a_{1}}+E_{a_{2}}-E_{a_{3}}$
$400=300+200-E_{a_{3}}$
$\mathrm{E}_{\mathrm{a}_{3}}=100 \mathrm{~kJ} / \mathrm{mole}$
29. xg of ethylamine on reaction with $\mathrm{NaNO}_{2}$ and HCl , produces 2.24 L of $\mathrm{N}_{2}(\mathrm{~g})$ at NTP. The value of 2 x will be
Answer (9)
Sol. $\mathrm{NaNO}_{2}+\mathrm{HCl} \longrightarrow \mathrm{NaCl}+\mathrm{HNO}_{2}$
$\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{NH}_{2}+\mathrm{HNO}_{2} \longrightarrow \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}+\mathrm{N}_{2} \uparrow+\mathrm{H}_{2} \mathrm{O}$
Mole of $\mathrm{N}_{2}(\mathrm{~g})$ produced $=\frac{2.24}{22.4}=0.1 \mathrm{~mol}$
So, mole of $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{NH}_{2}$ used $=0.1 \mathrm{~mol}$
Mass of $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{NH}_{2}=45 \times 0.1=4.5 \mathrm{~g}$
So, $2 \mathrm{x}=2 \times 4.5$

$$
=9
$$

30. 

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## MATHEMATICS

## SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

## Choose the correct answer:

1. If $f(x)=\left\{\begin{array}{ll}x-2, & 0<x \leq 2 \\ -2, & -2 \leq x \leq 0\end{array}\right.$ and
$h(x)=f(|x|)+|f(x)|$ then
find $\int_{0}^{k} h(x) d x$ is equal to $(k>0)$
(1) 0
(2) $\frac{k}{2}$
(3) $2 k$
(4) $k$

## Answer (1)

Sol. Graph of $f(x)$


$$
f(|x|)
$$



$|f(x)|= \begin{cases}2, & x<0 \\ 2-x, & x>0\end{cases}$
$\Rightarrow h(x)=f(|x|)+|f(x)|= \begin{cases}-x, & x<0 \\ 0, & x>0\end{cases}$
$\Rightarrow \int_{0}^{k} h(x) d x=\int_{0}^{k} 0 d x=0$
2. Let three urn $\mathrm{A}, \mathrm{B}, \mathrm{C}: \mathrm{A}=7$ red, 5 black

$$
\begin{aligned}
& B=5 \text { red, } 7 \text { black } \\
& C=6 \text { red, } 6 \text { black }
\end{aligned}
$$

Urn is selected and black ball is taken. Then the probability that the selected urn is $A$ is equal to
(1) $\frac{7}{18}$
(2) $\frac{5}{17}$
(3) $\frac{7}{19}$
(4) $\frac{5}{18}$

## Answer (4)

Sol. Urn A has 7 red, 5 black balls Urn B has 5 red, 7 black balls. Urn C has 6 red, 6 black balls
If ball drawn is black then probability that it is chosen from urn A .

$$
=\frac{\frac{1}{3} \times \frac{5}{12}}{\frac{1}{3} \times \frac{5}{12}+\frac{1}{3} \times \frac{7}{12}+\frac{1}{3} \times \frac{6}{12}}
$$

$$
=\frac{\frac{5}{36}}{\frac{5}{36}+\frac{7}{36}+\frac{6}{36}}
$$

$$
5
$$

$$
=\frac{\frac{36}{18}}{\frac{18}{36}}=\frac{5}{18}
$$

3. $\int_{-\pi / 2}^{\pi / 2} \frac{\sin ^{2} x}{1+2^{x}} d x=$
(1) $\left(\frac{\pi}{4}\right)$
(2) $\frac{\pi}{8}$
(3) $4 \pi$
(4) $\frac{\pi}{2}$

Answer (1)

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Sol. $I=\int_{0}^{\pi / 2}\left(\frac{\sin ^{2} x}{1+2^{x}}+\frac{\sin ^{2}(x)}{1+2^{-x}}\right) d x$
$I=\int_{0}^{\pi / 2} \sin ^{2} x d x$
$I=\int_{0}^{\pi / 2} \cos ^{2} x d x$
$2 I=\int_{0}^{\pi / 2} 1 d x$
$I=\frac{\pi}{4}$
4. If $f(x)=\frac{2 x^{2}-3 x+8}{2 x^{2}+3 x+8}$ then sum of maximum and minimum values of $f(x)$ is
(1) $\frac{136}{55}$
(2) $\frac{146}{55}$
(3) $\frac{146}{11}$
(4) $\frac{136}{11}$

## Answer (2)

Sol. $y=\frac{2 x^{2}-3 x+8}{2 x^{2}+3 x+8}, \quad 2 x^{2}+3 x+8>0 \forall x \in R$
$\Rightarrow x^{2}(2 y-2)+x(3 y+3)+8 y-8=0$
Since $x \in R$, the equation has real roots
$\Rightarrow$ Discriminant is greater than or equal to 0
$\Rightarrow(3 y+3)^{2}-4(2 y-2)(8 y-8) \geq 0$
$\Rightarrow 9(y+1)^{2}-64 y(y-1)^{2} \geq 0$
$\Rightarrow(3 y+3)^{2}-(8 y-8)^{2} \geq 0$
$\Rightarrow(11 y-5)(-5 y+11) \geq 0$
$\Rightarrow\left(y-\frac{5}{11}\right)\left(y-\frac{11}{5}\right) \leq 0$
$\Rightarrow y \in\left[\frac{5}{11}, \frac{11}{5}\right]$
$\Rightarrow$ Sum of $y_{\text {max }}$ and $y_{\text {min }}=\frac{5}{11}+\frac{11}{5}$

$$
\begin{aligned}
& =\frac{121+25}{55} \\
& =\left(\frac{146}{55}\right)
\end{aligned}
$$

5. The coefficient of $x^{7}$ in
$\left(1-x-x^{2}+x^{3}\right)^{6}$ equals to
(1) 132
(2) 144
(3) -132
(4) -144

## Answer (4)

Sol. Coefficient of $x^{7}$ in $(1-x)^{6}\left(1-x^{2}\right)^{6}$
${ }^{6} C_{1}{ }^{6} C_{3}-{ }^{6} C_{3}{ }^{6} C_{2}+{ }^{6} C_{5}{ }^{6} C_{1}$
$120-15 \times 20+6 \times 6$
$120-300+36$
$=-144$
6. If $(\bar{z})^{2}+|z|=0$ and if $\alpha$ is sum of roots and $\beta$ is product of non-zero roots, then $4\left(\alpha^{2}+\beta^{2}\right)$ is
(1) $\frac{1}{4}$
(2) 1
(3) 4
(4) 2

## Answer (3)

Sol. $(\bar{z})^{2}+|z|=0$
Let $z=x+i y$
$\Rightarrow(x-i y)^{2}+\sqrt{x^{2}+y^{2}}=0$
$\Rightarrow\left(x^{2}-y^{2}\right)+\sqrt{x^{2}+y^{2}}-2 x y i=0$
$\Rightarrow x^{2}-y^{2}+\sqrt{x^{2}+y^{2}}=0$ and $2 x y=0$
$\Rightarrow x=0$ and $y \neq 0$

## Case I

$\Rightarrow-y^{2}+|y|=0 \Rightarrow|y|=y^{2} \Rightarrow y= \pm 1$

## Cas II

$x \neq 0$ and $y=0$

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$\Rightarrow x^{2}+|x|=0 \Rightarrow x=0$ only not possible
$\Rightarrow x=0, y=0$ satisfies
$\Rightarrow z=i,-i, 0$ are solution
$\alpha=i-i=0$
$\beta=$ (i) $(-i)=-1 \Rightarrow 4\left(\alpha^{2}+\beta^{2}\right)=4$
7. If $\alpha \& \beta$ are roots of $a x^{2}+b x+c=0$ then equation whose roots are $\frac{1}{\alpha}, \frac{1}{\beta}$ is
(1) $c x^{2}+b x+a=0$
(2) $b x^{2}+a x+c=0$
(3) $a x^{2}+b x+c=0$
(4) $c x^{2}+a x+b=0$

Answer (1)
Sol. $a x^{2}+b x+c=0<\beta$
$\alpha+\beta=\frac{-b}{a}$
$\alpha \beta=\frac{c}{a}$
Now $\frac{1}{\alpha}+\frac{1}{\beta}=\frac{\alpha+\beta}{\alpha \beta}=-\frac{b}{c}$
$\frac{1}{\alpha \beta}=\frac{a}{c}$
$x^{2}-\left(\frac{1}{\alpha}+\frac{1}{\beta}\right) x+\frac{1}{\alpha \beta}=0$
$x^{2}+\frac{b}{c} x+\frac{a}{c}=0$
$c x^{2}+b x+a=0$
8. Let $f(x)= \begin{cases}\frac{1-\cos \alpha x}{x^{2}} ; & x<0 \\ 2^{2} ; & x=0 \\ \frac{\beta \sqrt{1-\cos x}}{x} ; & x>0\end{cases}$
is continuous at $x=0$. Then $\alpha^{2}+\beta^{2}$ equals to
(1) 10
(2) 12
(3) 13
(4) 9

Answer (2)

Sol. Given $f(x)$ is continuous at $x=0$
$\therefore \lim _{x \rightarrow 0^{-}} f(x)=f(0)=\lim _{x \rightarrow 0^{+}} f(x)$
When $x<0, x=0-h$
$\therefore \lim _{h \rightarrow 0} \frac{1-\cos (\alpha(0-h))}{(0-h)^{2}}$
$=\lim _{h \rightarrow 0} \frac{1-\cos (h \alpha)}{h^{2}}$
$=\lim _{h \rightarrow 0}\left(\frac{1-\cos (\alpha h)}{\alpha^{2} . h^{2}}\right) \alpha^{2}$
$=\alpha^{2} \lim _{h \rightarrow 0} \frac{1-\cos (\alpha h)}{(\alpha h)^{2}}$
$=\frac{\alpha^{2}}{2}$
When $x>0$
$x=0+h$

$$
\begin{align*}
\lim _{h \rightarrow 0} \frac{\beta \sqrt{1-\cos h}}{h} & =\lim _{h \rightarrow 0} \frac{\beta \sqrt{\frac{1-\cosh }{h^{2}} \cdot h^{2}}}{h} \\
& =\frac{\beta}{\sqrt{2}} \quad \ldots(2) \tag{2}
\end{align*}
$$

as $f(0)=2$
$\therefore$ From (1), (2) and (3)
$\frac{\alpha^{2}}{2}=2, \quad \frac{\beta}{\sqrt{2}}=2$
$\alpha=2, \quad \beta=2 \sqrt{2}$
$\alpha^{2}+\beta^{2}=4+8=12$
9. If the length of focal chord of $y^{2}=12 x$ is 15 and if the distance of the focal chord from origin is $P$ then $10 P^{2}$ is equal to
(1) 36
(2) 25
(3) 72
(4) 144

Answer (3)


Sol.

$\Rightarrow A B=15$
$\left(3 t^{2}-\frac{3}{t^{2}}\right)^{2}+\left(6 t+\frac{6}{t}\right)^{2}=225$
$\Rightarrow 9\left(t^{2}-\frac{1}{t^{2}}\right)^{2}+36\left(t+\frac{1}{t}\right)^{2}=225$
$\Rightarrow 9\left(t+\frac{1}{t}\right)^{2}\left[\left(t-\frac{1}{t}\right)^{2}+4\right]=225$
$\Rightarrow 9\left(t+\frac{1}{t}\right)^{2}\left(t+\frac{1}{t}\right)^{2}=225$
$\Rightarrow t+\frac{1}{t}=\left(\frac{225}{9}\right)^{1 / 4}=(25)^{1 / 4}=\sqrt{5}$
Equation of $A B \equiv(y-0)=\frac{2}{\left(t-\frac{1}{t}\right)}(x-3) \Rightarrow\left|t-\frac{1}{t}\right|=1$
$\Rightarrow y=2 x-6 \Rightarrow y-2 x+6=0$
Distance from origin $\Rightarrow P=\frac{6}{\sqrt{5}} \Rightarrow 10 P^{2}=\frac{10 \times 36}{5}$ $=72$
10. Numbers $-3,4,7,-6, \alpha, \beta$

Mean $=2$, Variance $=23$, then
Mean deviation about mean equals to
(1) $\frac{13}{8}$
(2) $\frac{13}{3}$
(3) $\frac{13}{7}$
(4) $\frac{13}{9}$

Answer (2)

Sol. Mean $=\frac{-3+4+7+(-6)+\alpha+\beta}{6}=2$
$=2+\alpha+\beta=2 \times 6$
$\Rightarrow \alpha+\beta=10$
Variance $=\frac{\Sigma x i^{2}}{n}-\left(\frac{\bar{x}}{n}\right)^{2}=23$
$=\frac{\Sigma x i^{2}}{n}=23+4$
$=\Sigma x i^{2}=27 \times 6$
$=9+16+49+36+\alpha^{2}+\beta^{2}=162$
$\Rightarrow \alpha^{2}+\beta^{2}=52$
$\Rightarrow$ We get $\alpha$ and $\beta$ as 4 and 6
So, mean deviation about mean
$=\frac{|-3-2|+|4-2|+|7-2|+|-6-2|+|4-2|+|6-2|}{6}$
$=\frac{5+2+5+8+2+4}{6}$
$=\frac{26}{6}=\frac{13}{3}$
11. If $\frac{d y}{d x}=\frac{2 x^{2}+2 x+3}{x^{4}+2 x^{3}+3 x^{2}+2 x+2}$
and $y(-1)=-\frac{\pi}{4}$
then $y(0)$ is
(1) $\frac{\pi}{3}$
(2) $\frac{\pi}{4}$
(3) $\frac{\pi}{2}$
(4) $\frac{\pi}{6}$

Answer (2)
Sol. $\int d y=\int \frac{2 x^{2}+2 x+3}{x^{4}+2 x^{3}+3 x^{2}+2 x+2} d x$

$=\int \frac{2 x^{2}+2 x+3}{\left(x^{2}+1\right)\left(x^{2}+2 x+2\right)} d x$
$=\int \frac{1}{x^{2}+2 x+2} d x+\int \frac{1}{x^{2}+1} d x$
$=\int \frac{1}{1+(x+1)^{2}} d x+\tan ^{-1} x+C$
$y=\tan ^{-1}(x+1)+\tan ^{-1} x+C$
$y(-1)=-\frac{\pi}{4}$
$-\frac{\pi}{4}=0-\frac{\pi}{4}+C$
$\Rightarrow C=0$
$\therefore \quad y=\tan ^{-1}(x+1)+\tan ^{-1}(x)$
Now $y(0)=\tan ^{-1}(1)+\tan ^{-1}(0)=\frac{\pi}{4}$
12. If $\vec{c}$ is a variable unit vector and $\vec{c}$ makes angle of $45^{\circ}$ with $\vec{b}$ and $60^{\circ}$ with $\vec{a}$ with $\vec{b}=\hat{i}-\hat{k}$ and $\vec{a}=2 \hat{i}+2 \hat{j}-\hat{k}$ then $|\vec{c}+2 \vec{a}-3 \vec{b}|$ is
(1) 19
(2) 20
(3) $\sqrt{19}$
(4) $\sqrt{20}$

## Answer (3)

Sol. $\bar{c}$ is unit vector

$$
\begin{aligned}
& \vec{b}=\hat{i}-\hat{k} \\
& \vec{a}=2 \hat{i}+2 \hat{j}-\hat{k} \\
& |\vec{a}|=3,|\vec{b}|=\sqrt{2},|\vec{c}|=1 \\
& |\vec{c}+2 \vec{a}-3 \vec{b}|^{2}=|\vec{c}|^{2}+4|\vec{a}|^{2}+9|\vec{b}|^{2}+4 \vec{a} \cdot \vec{c} \\
& -12 \vec{a} \cdot \vec{b}-6 \vec{b} \cdot \vec{c}
\end{aligned}
$$

$$
=1+36+18+4|\vec{a}||\vec{c}| \cos 60^{\circ}-12[3]
$$

$$
-6|\vec{b}||\vec{c}| \cos 45^{\circ}
$$

$$
=55+12 \times \frac{1}{2}-36-6 \sqrt{2} \times \frac{1}{\sqrt{2}}
$$

$=55+6-36-6$
$=19$
$|\vec{c}+2 \vec{a}-3 \vec{b}|=\sqrt{19}$
13. If the system of equations
$A+\sqrt{2} \sin x B+\sqrt{2} \cos x C=0$
$A+\sin x B-\cos x C=0$
$A+\cos x B+\sin x C=0$ has non-trivial solution then the value of $x, x \in\left(0, \frac{\pi}{2}\right)$ is
(1) $\frac{5 \pi}{12}$
(2) $\frac{\pi}{12}$
(3) $\frac{5 \pi}{24}$
(4) $\frac{\pi}{8}$

## Answer (3)

Sol. For non-trivial solution

$$
\begin{aligned}
& \left|\begin{array}{ccc}
1 & \sqrt{2} \sin x & \sqrt{2} \cos x \\
1 & \sin x & -\cos x \\
1 & \cos x & \sin x
\end{array}\right| \text { is zero } \\
& \Rightarrow \quad 1-1\left(\sqrt{2} \sin ^{2} x-\sqrt{2} \cos ^{2} x\right)+1(-2 \sqrt{2} \sin x \cos x)=0 \\
& \Rightarrow 1+\sqrt{2}(\cos 2 x)-\sqrt{2} \sin 2 x=0 \\
& \Rightarrow \quad \sqrt{2}(\cos 2 x-\sin 2 x)=-1 \\
& \Rightarrow \cos \left(2 x+\frac{\pi}{4}\right)=\frac{-1}{2} \\
& x \in\left(0, \frac{\pi}{2}\right) \\
& 2 x \in(0, \pi) \\
& 2 x+\frac{\pi}{4} \in\left(\frac{\pi}{4}, \frac{5 \pi}{4}\right) \\
& \Rightarrow \cos \left(2 x+\frac{\pi}{4}\right)=\frac{-1}{2} \Rightarrow 2 x+\frac{\pi}{4}=\frac{2 \pi}{3} \\
& \Rightarrow \quad x=\frac{5 \pi}{24}
\end{aligned}
$$

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14. A line $L_{1}$ having equation $y=x+3$. A square is inscribed in a circle $x^{2}+y^{2}-10 x-6 y+30=0$ such that one side of square is parallel to $L_{1}$. Find $\sum_{i=1}^{4}\left(x_{i}^{2}+y_{i}^{2}\right)$ where $\left(x_{i}, y_{i}\right) i \in\{1,2,3,4\}$ are the vertices of square.
(1) 152
(2) 162
(3) 172
(4) 182

Answer (1)

Sol.


Distance of $(5,3)$ to the line $y=x+c$ is $\sqrt{2}$

$$
\begin{aligned}
& \Rightarrow \frac{|3-5-c|}{\sqrt{2}}=\sqrt{2} \\
& |c+2|=2 \\
& \Rightarrow c=0 \\
& \quad c=-4
\end{aligned}
$$

So, the lines are $y=x$ and $y=x-4$
Now, solving these lines with the circle
$y=x$ and $x^{2}+y^{2}-10 x-6 y+30=0$
$2 x^{2}-16 x+30=0$
$x^{2}-8 x+15=0$
$x=3, y=3$
$x=5, y=5$
$y=x-4$ and $x^{2}+y^{2}-10 x-6 y+30=0$
$2 x^{2}-24 x+70=0$
$x^{2}-12 x+35=0$
$x=5, y=1$
$x=7, y=3$
$\sum_{i=1}^{4} x_{i}^{2}+y_{i}^{2}=9+9+25+25+25+1+49+9=152$
15.
16.
17.
18.
19.
20.

## SECTION - B

Numerical Value Type Questions: This section contains 10 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.
21. The number of rational numbers in the expansion of $\left(2^{1 / 5}+5^{1 / 3}\right)^{15}$ is
Answer (02)
Sol. $\mathrm{T}_{r+1}={ }^{15} C_{r}\left(5^{1 / 3}\right)^{r}\left(2^{1 / 5}\right)^{15-r}, r \in\{0,1, \ldots .15\}$
$={ }^{15} C_{r} 5^{\left(\frac{r}{3}\right)} \cdot 2^{\left(3-\frac{r}{5}\right)}, \quad r \in\{0,1, \ldots 15\}$
For rational terms,
$\frac{r}{3} \in$ integer and $\frac{r}{5} \in$ integer
$\Rightarrow 3$ and 5 divides $r \Rightarrow 15$ divides $r$
$\Rightarrow r=0$ and 15
$\Rightarrow$ only 2 rational terms.
22. In $\triangle A B C$ there are 18 points, on side $A B$ there are $P_{1}, P_{2}, P_{3}, P_{4}, P_{5}$ points, on $B C$ there are $P_{6}, P_{7}$ $\ldots P_{11}$ points and on $C A P_{12} \ldots P_{18}$ points. By joining any three points from $P_{1}, P_{2} \ldots P_{18}$ form a triangle. Then number of triangles possible are

## Answer (751)

Sol. Total ways to select three points out of 18 points $=$ ${ }^{18} \mathrm{C}_{3}$

Total ways to select 3 points from $P_{1} \ldots P_{5}={ }^{5} C_{3}$
Total ways to select 3 points from $P_{6} \ldots P_{11}={ }^{6} C_{3}$
Total ways to select 3 points from $P_{12} \ldots P_{18}={ }^{7} C_{3}$
Total number of triangles possible

$$
\begin{aligned}
& ={ }^{18} C_{3}-{ }^{5} C_{3}-{ }^{6} C_{3}-{ }^{7} C_{3} \\
& =751
\end{aligned}
$$

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## Our Stars


23. If $\operatorname{limit}_{x \rightarrow 1} \frac{(5 x+1)^{1 / 3}-(x+5)^{1 / 3}}{(2 x+3)^{1 / 2}-(x+4)^{1 / 2}}=\frac{m(5)^{1 / 2}}{n(2 n)^{2 / 3}}$

Then $8 m+12 n$ is

## Answer (100)

Sol. $\operatorname{limit}_{x \rightarrow 1} \frac{(5 x+1)^{1 / 3}-(x+5)^{1 / 3}}{(2 x+3)^{1 / 2}-(x+4)^{1 / 2}}$
$\operatorname{limit}_{x \rightarrow 1} \frac{\frac{1}{3}(5 x+1)^{-2 / 3} \cdot 5-\frac{1}{3}(x+5)^{-2 / 3}}{2 \times \frac{1}{2}(2 x+3)^{-1 / 2}-\frac{1}{2}(x+4)^{-1 / 2}}$
$=\frac{\frac{1}{3} \times \frac{5}{(6)^{2 / 3}}-\frac{1}{3} \times \frac{1}{(6)^{2 / 3}}}{\frac{1}{2} \times \frac{2}{(5)^{1 / 2}}-\frac{1}{2} \times \frac{1}{(5)^{1 / 2}}}$
$=\frac{\frac{4}{3 \times(6)^{2 / 3}}}{\frac{1}{2 .(5)^{1 / 2}}}=\frac{8(5)^{1 / 2}}{3(6)^{2 / 3}}=\frac{m(5)^{1 / 2}}{n(2 n)^{2 / 3}}$
$\Rightarrow m=8, n=3$
$8 m+12 n=64+36=100$
24. In a G.P. $T_{1}=2, T_{2}=P, T_{3}=Q$, these are also terms of A.P $\left(7^{\text {th }}, 8^{\text {th }}\right.$ and $13^{\text {th }}$ term $)$.
If $5^{\text {th }}$ term of $\mathrm{G} . \mathrm{P}=n^{\text {th }}$ term of A.P3. Then $n$ is
Answer (27)
Sol. $T_{1}=2$

$$
a=2
$$

$$
\begin{array}{ll}
T_{2}=P & 2 r=P \Rightarrow r=\frac{P}{2} \\
T_{3}=Q & 2 r^{2}=Q \Rightarrow r^{2}=\frac{Q}{2} \\
a^{\prime}+6 d=2 & \ldots(1) \\
a^{\prime}+7 d=P & \ldots(2) \\
a^{\prime}+12 d=Q & \ldots(3)
\end{array}
$$

$d=2(r-1)$
$2 r(r-1)=5 d$
$\frac{5 d}{d}=\frac{-2 r(r-1)}{2(r-1)}$
$r=5 \Rightarrow d=8$
$a+48=2$
$a=-46$
$2.3^{4}=-46+(n-1) \times 8$
$\Rightarrow n=27$
25. Domain of $\sin ^{-1}\left(\frac{3 x-22}{2 x-19}\right)+\log _{e}\left(\frac{3 x^{2}-8 x+5}{x^{2}-3 x-10}\right)$
is $(\alpha, \beta]$. Then $3 \alpha+10 \beta$ equals to

## Answer (97)

Sol. $-1 \leq \frac{3 x-22}{2 x-19} \leq 1$
$\frac{3 x-22-2 x+19}{2 x-19} \leq 0$
$\frac{x-3}{2 x-19} \leq 0$

$\left[3, \frac{19}{2}\right)$
$\frac{3 x-22+2 x-19}{2 x-19} \geq 0$
$\frac{5 x-41}{2 x-19} \geq 0$

$\left(-\infty, \frac{41}{5}\right] \cup\left(\frac{19}{2}, \infty\right)$
Taking intersection

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Taking intersection of individual domains

$\left(5, \frac{41}{5}\right]$
$3 \alpha+10 \beta=3 \times 5+10 \times \frac{41}{5}$
$=15+82=97$
26. If $a=\frac{1}{2!}+\frac{{ }^{2} C_{2}}{3!}+\frac{{ }^{3} C_{2}}{4!}+\frac{{ }^{4} C_{2}}{5!}+\ldots$.

$$
b=1+\frac{{ }^{1} C_{0}+{ }^{1} C_{1}}{1!}+\frac{{ }^{2} C_{0}+{ }^{+2} C_{1}+{ }^{2} C_{2}}{2!}+\ldots .
$$

Then $\frac{2 b}{a^{2}}$ equals to

## Answer (8)

Sol. $a=\frac{1}{2}+\sum_{n=2}^{\infty} \frac{{ }^{n} C_{2}}{(n+1)!}$

$$
=\frac{1}{2}+\sum_{n=2}^{\infty} \frac{\frac{n(n+1)}{2}}{(n+1)!}
$$

$=\frac{1}{2}+\sum_{n=2}^{\infty} \frac{1}{2} \times \frac{1}{(n-1)!}$
$=\frac{1}{2}+\frac{1}{2}(e-1)$
$=\frac{e}{2}$
$b=1+\frac{2^{1}}{1!}+\frac{2^{2}}{2!}+\frac{2^{3}}{3!}+\ldots .$.
$b=e^{2}$
$\frac{2 b}{a^{2}}=\frac{2 \times e^{2}}{\frac{e^{2}}{4}}=8$
27. If $A=\left[\begin{array}{lll}1 & 2 & \alpha \\ 1 & 0 & 1 \\ 0 & 1 & 2\end{array}\right]$ and $\operatorname{Det}\left(\operatorname{Adj}\left(A-2 A^{\top}\right) \operatorname{Adj}(2 A-\right.$ $\left.A^{\top}\right)$ ) $=^{8}$ then $\operatorname{det}(A)^{2}$ is

## Answer (16.00)

Sol. $\left|\operatorname{Adj}\left(A-2 A^{T}\right) \operatorname{Adj}\left(2 A-A^{T}\right)\right|=2^{8}$
$P=A-2 A^{T}$
$Q=2 A^{T}-A \Rightarrow Q^{T}=2 A^{T}-A=-P$
$\mid \operatorname{adj}(P)$ adj $Q\left|=2^{8}, \Rightarrow\right| Q^{\top}|=|-P| \Rightarrow| Q|=-|P|$
$|P|^{2}|Q|^{2}=2^{8} \Rightarrow|P Q|=-2^{4}$
$\Rightarrow|P|(-|P|)=-2^{4} \Rightarrow|P|=4$ and $|Q|=-4$
$\left|A-2 A^{T}\right|=4$
$A-2 A^{T}=\left[\begin{array}{lll}1 & 2 & \alpha \\ 1 & 0 & 1 \\ 0 & 1 & 2\end{array}\right]-2\left[\begin{array}{lll}1 & 1 & 0 \\ 2 & 0 & 1 \\ \alpha & 1 & 2\end{array}\right]=\left[\begin{array}{ccc}-1 & 0 & \alpha \\ -3 & 0 & -1 \\ -2 \alpha & -1 & -2\end{array}\right]$
$\Rightarrow\left|A-2 A^{\top}\right|=1+3 \alpha=4 \Rightarrow \alpha=1 \Rightarrow|A|=-4 \Rightarrow$ $|A|^{2}=16$
28.
29.
30.

Aakashians Gonquer JEE (Main) 2024
SESSION-1


