

Reaction Name	Reactants	Key Reagents	Main Product	Unique Feature/Tip
Aldol Condensation	Aldehydes/Ketones with α -H	Dilute NaOH	β -hydroxy aldehyde	Formation of C-C bond.
Cannizzaro Reaction	Aldehydes with no α -H	Conc. NaOH	Alcohol + Carboxylic Acid	Disproportionation reaction.
Reimer-Tiemann	Phenol + CHCl_3	Aqueous NaOH	Salicylaldehyde	Introduces -CHO group to Phenol.
Kolbe's Reaction	Phenol	NaOH, CO_2 , H^+	Salicylic Acid	Synthesis of Aspirin precursor.
Rosenmund Reduction	Acid Chloride	Pd/ BaSO_4 , H_2	Aldehyde	"Poisoned" catalyst prevents over-reduction.
Stephen Reaction	Nitriles	SnCl_2/HCl	Aldehyde	Selective reduction of Cyanides.
Clemmensen Reduction	Carbonyl Group	Zn-Hg/ HCl	Alkane	Best for acid-stable molecules.
Wolff-Kishner	Carbonyl Group	$\text{NH}_2\text{NH}_2/\text{KOH}$	Alkane	Best for base-stable molecules.
Williamson Synthesis	Alkyl Halide + Alkoxide	NaOR	Ether	$\text{S}_\text{N}2$ mechanism; use 1° halides.
Hoffmann Bromamide	Amide	Br_2/NaOH	1° Amine	Degrades chain by 1 Carbon atom.
Gabriel Phthalimide	Phthalimide	KOH, Alkyl Halide	Pure 1° Amine	No 2° or 3° amine contamination.
Sandmeyer Reaction	Diazonium Salt	CuCl/CuBr	Haloarene	Replaces $-\text{N}_2^+$ with Halogen.
Wurtz Reaction	Alkyl Halide	Na / Dry Ether	Higher Alkane	Doubles the carbon chain length.
Fittig Reaction	Aryl Halide	Na / Dry Ether	Biphenyl	Coupling of two aromatic rings.
Friedel-Crafts Alk.	Benzene + R-Cl	Anhydrous AlCl_3	Alkylbenzene	Electrophilic substitution.
Etard Reaction	Toluene	CrO_2Cl_2	Benzaldehyde	Selective oxidation of methyl group.
Hell-Volhard-Zelinsky	Carboxylic Acid	Red P/ X_2	α -halo acid	Halogenation at the α -carbon.
Gatterman-Koch	Benzene + CO/ HCl	Anhyd. AlCl_3	Benzaldehyde	Industrial synthesis of Benzaldehyde.
Hunsdiecker	Silver Salt of Acid	Br_2/CCl_4	Alkyl Bromide	Decarboxylative bromination.
Swarts Reaction	Alkyl Chloride/Bromide	AgF or Hg_2F_2	Alkyl Fluoride	Best way to synthesize Fluorides.